







Note: the *r*-component of the Navier-Stokes equation in spherical coordinates may be simplified by adding $0 = \frac{2}{2}\nabla \cdot \underline{v}$ to the component shown above. This term is zero due to the continuity equation (mass conservation). See Bird et. al. Equation of Motion for incompressible, Newtonian fluid (Navier-Stokes equation) 3 components in Cartesian Equation of Motion for incompressible, Newtonian fluid (Navier-Stokes equation), 3 components in spherical Equation of Motion for incompressible, Newtonian fluid (Navier-Stokes equation), 3 components in cylin-2. R. B. Bird, R. C. Armstrong, and O. Hassager, Dynamics of Polymeric Fluids: Volume 1 Fluid Mechanics, $\rho\left(\frac{\partial v_{\theta}}{\partial t} + v_{r}\frac{\partial v_{\theta}}{\partial r} + \frac{v_{\theta}}{r}\frac{\partial v_{\theta}}{\partial \theta} + \frac{v_{r}v_{\theta}}{r} + v_{z}\frac{\partial v_{\theta}}{\partial z}\right) = -\frac{1}{r}\frac{\partial P}{\partial \theta} + \mu\left(\frac{\partial}{\partial r}\left(\frac{1}{r}\frac{\partial(rv_{\theta})}{\partial r}\right) + \frac{1}{r^{2}}\frac{\partial^{2}v_{\theta}}{\partial \theta^{2}} + \frac{\partial^{2}v_{\theta}}{\partial z^{2}}\right) + \rho_{\theta}$ $\rho\left(\frac{\partial v_r}{\partial t} + v_r\frac{\partial v_r}{\partial r} + \frac{v_\theta}{r}\frac{\partial v_r}{\partial \theta} - \frac{v_\theta^2}{r} + v_z\frac{\partial v_r}{\partial z}\right) = -\frac{\partial P}{\partial r} + \mu\left(\frac{\partial}{\partial r}\left(\frac{1}{r}\frac{\partial(rv_r)}{\partial r}\right) + \frac{1}{r^2}\frac{\partial^2 v_r}{\partial \theta^2} - \frac{2}{r^2}\frac{\partial v_\theta}{\partial \theta} + \frac{\partial^2 v_r}{\partial z^2}\right) + \rho g_r$ $= -\frac{\partial P}{\partial r} + \mu \left(\frac{\partial}{\partial r} \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 v_r \right) \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial v_r}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 v_r}{\partial \phi^2} \\ - \frac{2}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(v_\theta \sin \theta \right) - \frac{2}{r^2 \sin \theta} \frac{\partial v_\theta}{\partial \phi} \right) + \rho g_r$ $+\frac{2}{r^2\sin\theta}\frac{\partial v_r}{\partial\phi}+\frac{2\cot\theta}{r^2\sin\theta}\frac{\partial v_\theta}{\partial\phi}\Big)+\rho g_\phi$ $+\frac{2}{r^2}\frac{\partial v_r}{\partial \theta}-\frac{2\cot\theta}{r^2\sin\theta}\frac{\partial v_\phi}{\partial \phi}\Big)+\rho g_\theta$ 1. R. B. Bird, W. E. Stewart, and E. N. Lightfoot, Transport Phenomena, 2nd edition, Wiley: NY, 2002. $\rho\left(\frac{\partial v_z}{\partial t} + v_r\frac{\partial v_z}{\partial r} + \frac{v_\theta}{r}\frac{\partial v_z}{\partial \theta} + v_r\frac{\partial v_z}{\partial z}\right) = -\frac{\partial P}{\partial z} + \mu\left(\frac{1}{r}\frac{\partial}{\partial r}\left(r\frac{\partial v_z}{\partial r}\right) + \frac{1}{r^2}\frac{\partial^2 v_z}{\partial \theta^2} + \frac{\partial^2 v_z}{\partial z^2}\right) + \rho g_z$ $= -\frac{1}{r\sin\theta}\frac{\partial P}{\partial\phi} + \mu \left(\frac{1}{r^2}\frac{\partial}{\partial r} \left(r^2\frac{\partial v_{\phi}}{\partial r}\right) + \frac{1}{r^2}\frac{\partial}{\partial\theta} \left(\frac{1}{\sin\theta}\frac{\partial}{\partial\theta} \left(v_{\phi}\sin\theta\right)\right) + \frac{1}{r^2\sin^2\theta}\frac{\partial^2 v_{\phi}}{\partial\phi^2}$
$$\begin{split} \rho\left(\frac{\partial v_x}{\partial t} + v_x \frac{\partial v_x}{\partial x} + v_y \frac{\partial v_x}{\partial y} + v_z \frac{\partial v_x}{\partial z}\right) &= -\frac{\partial P}{\partial x} + \mu\left(\frac{\partial^2 v_x}{\partial x^2} + \frac{\partial^2 v_x}{\partial y^2} + \frac{\partial^2 v_x}{\partial z^2}\right) + \rho g_x\\ \rho\left(\frac{\partial v_y}{\partial t} + v_x \frac{\partial v_y}{\partial x} + v_y \frac{\partial v_y}{\partial y} + v_z \frac{\partial v_y}{\partial z}\right) &= -\frac{\partial P}{\partial y} + \mu\left(\frac{\partial^2 v_y}{\partial x^2} + \frac{\partial^2 v_y}{\partial y^2} + \frac{\partial^2 v_y}{\partial z^2}\right) + \rho g_y\\ \rho\left(\frac{\partial v_z}{\partial t} + v_x \frac{\partial v_z}{\partial x} + v_y \frac{\partial v_z}{\partial y} + v_z \frac{\partial v_z}{\partial z}\right) &= -\frac{\partial P}{\partial z} + \mu\left(\frac{\partial^2 v_z}{\partial x^2} + \frac{\partial^2 v_z}{\partial y^2} + \frac{\partial^2 v_z}{\partial z^2}\right) + \rho g_z \end{split}$$
 $= -\frac{1}{r}\frac{\partial P}{\partial \theta} + \mu\left(\frac{1}{r^2}\frac{\partial}{\partial r}\left(r^2\frac{\partial v_\theta}{\partial r}\right) + \frac{1}{r^2}\frac{\partial}{\partial \theta}\left(\frac{1}{\sin\theta}\frac{\partial}{\partial \theta}\left(v_\theta\sin\theta\right)\right) + \frac{1}{r^2\sin^2\theta}\frac{\partial^2 v_\theta}{\partial \phi^2}$ $\rho\left(\frac{\partial v_{\phi}}{\partial t} + v_{r}\frac{\partial v_{\phi}}{\partial r} + \frac{v_{\theta}}{r}\frac{\partial v_{\phi}}{\partial \theta} + \frac{v_{\phi}}{r\sin\theta}\frac{\partial v_{\phi}}{\partial \phi} + \frac{v_{r}v_{\phi}}{r} + \frac{v_{\phi}v_{\phi}}{r} + \frac{v_{\phi}v_{\theta}}{r}\right)$ $\rho\left(\frac{\partial v_{\theta}}{\partial t} + v_r \frac{\partial v_{\theta}}{\partial r} + \frac{v_{\theta}}{r} \frac{\partial v_{\theta}}{\partial \theta} + \frac{v_{\phi}}{r \sin\theta} \frac{\partial v_{\theta}}{\partial \phi} + \frac{v_r v_{\theta}}{r} - \frac{v_{\phi}^2 \cot\theta}{r}\right)$ $\rho\left(\frac{\partial v_r}{\partial t} + v_r\frac{\partial v_r}{\partial r} + \frac{v_\theta}{r}\frac{\partial v_r}{\partial \theta} + \frac{v_\theta}{r}\frac{\partial v_r}{\partial \theta} + \frac{v_\phi}{r\sin\theta}\frac{\partial v_r}{\partial \phi} - \frac{v_\theta^2 + v_\phi^2}{r}\right)$ Wiley: NY, 1987. drical coordinates References: coordinates coordinates $\rho\left(\frac{\partial v_{\theta}}{\partial t} + v_{r}\frac{\partial v_{\theta}}{\partial r} + \frac{v_{\theta}}{r}\frac{\partial v_{\theta}}{\partial \theta} + \frac{v_{\theta}v_{r}}{r} + v_{z}\frac{\partial v_{\theta}}{\partial z}\right) = -\frac{1}{r}\frac{\partial P}{\partial \theta} + \left(\frac{1}{r^{2}}\frac{\partial(r^{2}\tilde{x}_{\theta})}{\partial r} + \frac{1}{r}\frac{\partial\tilde{x}_{\theta\theta}}{\partial\theta} + \frac{\partial\tilde{x}_{z}}{\partial z} + \frac{\tilde{x}_{\theta}r - \tilde{x}_{r\theta}}{r}\right) + \rho_{\theta\theta}$ The Equation of Continuity and the Equation of Motion in Cartesian, cylindrical, and spherical coordinates $= -\frac{1}{r\sin\theta}\frac{\partial P}{\partial\phi} + \left(\frac{1}{r^3}\frac{\partial(r^3\bar{r}_{\gamma\phi})}{\partial r} + \frac{1}{r\sin\theta}\frac{\partial(\bar{r}_{\theta\phi}\sin\theta)}{\partial\theta} + \frac{1}{r\sin\theta}\frac{\partial\bar{r}_{\theta\phi}}{\partial\phi} + \frac{1}{\bar{r}}\frac{\bar{r}_{\bar{r}}r_{\bar{r}}}{r} + \frac{\bar{r}_{\phi\sigma}\cot\theta}{r}\right) + \rho_{\theta\phi}^{2}$ $= -\frac{1}{r}\frac{\partial P}{\partial \theta} + \left(\frac{1}{r^3}\frac{\partial(r^3\tilde{\tau}_{r\theta})}{\partial r} + \frac{1}{r\sin\theta}\frac{\partial(\tilde{\tau}_{r\theta}\sin\theta)}{\partial \theta} + \frac{1}{r\sin\theta}\frac{\partial\tilde{\tau}_{r\theta}}{\partial \phi} + \frac{\tilde{\tau}_{\theta}r - \tilde{\tau}_{r\theta}}{r} - \frac{\tilde{\tau}_{\phi\phi}\cot\theta}{r}\right) + \rho_{\theta\theta}$ $\rho\left(\frac{\partial v_r}{\partial t} + v_r \frac{\partial v_r}{\partial r} + \frac{v_\theta}{r} \frac{\partial v_r}{\partial \theta} - \frac{v_\theta^2}{r} + v_z \frac{\partial v_r}{\partial z}\right) = -\frac{\partial P}{\partial r} + \left(\frac{1}{r} \frac{\partial (r\tilde{\tau}_{rr})}{\partial r} + \frac{1}{r} \frac{\partial \tilde{\tau}_{\theta r}}{\partial \theta} - \frac{\tilde{\tau}_{\theta \theta}}{r} + \frac{\partial \tilde{\tau}_{zr}}{\partial z}\right) + \rho_{\theta r}$
$$\begin{split} \rho\left(\frac{\partial v_x}{\partial t} + v_x \frac{\partial v_x}{\partial x} + v_y \frac{\partial v_x}{\partial y} + v_z \frac{\partial v_x}{\partial z}\right) &= -\frac{\partial P}{\partial x} + \left(\frac{\partial \tilde{r}_{xx}}{\partial x} + \frac{\partial \tilde{r}_{yx}}{\partial z} + \frac{\partial \tilde{r}_{xx}}{\partial z}\right) + \rho g_x \\ \rho\left(\frac{\partial v_y}{\partial t} + v_x \frac{\partial v_y}{\partial x} + v_y \frac{\partial v_y}{\partial y} + v_z \frac{\partial v_y}{\partial z}\right) &= -\frac{\partial P}{\partial y} + \left(\frac{\partial \tilde{r}_{xy}}{\partial x} + \frac{\partial \tilde{r}_{yy}}{\partial y} + \frac{\partial \tilde{r}_{xy}}{\partial z}\right) + \rho g_y \\ \rho\left(\frac{\partial v_z}{\partial t} + v_x \frac{\partial v_z}{\partial x} + v_y \frac{\partial v_z}{\partial y} + v_z \frac{\partial v_z}{\partial z}\right) &= -\frac{\partial P}{\partial z} + \left(\frac{\partial \tilde{r}_{xx}}{\partial x} + \frac{\partial \tilde{r}_{yy}}{\partial x} + \frac{\partial \tilde{r}_{xy}}{\partial z}\right) + \rho g_z \end{split}$$
Equation of Motion for an incompressible fluid, 3 components in cylindrical coordinates $\rho\left(\frac{\partial v_z}{\partial t} + v_r \frac{\partial v_z}{\partial r} + \frac{v_\theta}{r} \frac{\partial v_z}{\partial \theta} + v_z \frac{\partial v_z}{\partial z}\right) = -\frac{\partial P}{\partial z} + \left(\frac{1}{r} \frac{\partial (r\tilde{\tau}_{rz})}{\partial r} + \frac{1}{r} \frac{\partial \tilde{\tau}_{\theta z}}{\partial \theta} + \frac{\partial \tilde{\tau}_{zz}}{\partial z}\right) + \rho_{\theta z}$ $= -\frac{\partial P}{\partial r} + \left(\frac{1}{r^2}\frac{\partial(r^2\tilde{r}_{rr})}{\partial r} + \frac{1}{r\sin\theta}\frac{\partial(\tilde{r}_{\theta}\sin\theta)}{\partial\theta} + \frac{1}{r\sin\theta}\frac{\partial\tilde{r}_{\theta}\omega}{\partial\phi} + \frac{1}{r}\frac{\partial\tilde{r}_{\theta}\omega}{\partial\phi} + \frac{1}{\rho}\theta,$ Equation of Motion for an incompressible fluid, 3 components in Cartesian coordinates Equation of Motion for an incompressible fluid, 3 components in spherical coordinates $\frac{\partial\rho}{\partial t} + \left(v_x \frac{\partial\rho}{\partial x} + v_y \frac{\partial\rho}{\partial y} + v_z \frac{\partial\rho}{\partial z}\right) + \rho \left(\frac{\partial v_x}{\partial x} + \frac{\partial v_y}{\partial y} + \frac{\partial v_z}{\partial z}\right) = 0$ $\frac{\partial\rho}{\partial t} + \frac{1}{r^2} \frac{\partial(\rho r^2 v_r)}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial(\rho v_\theta \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial(\rho v_\theta)}{\partial \phi} = 0$ $\rho \left(\frac{\partial v_{\phi}}{\partial t} + v_r \frac{\partial v_{\phi}}{\partial r} + \frac{v_{\theta}}{r} \frac{\partial v_{\phi}}{\partial \theta} + \frac{v_{\phi}}{r \sin \theta} \frac{\partial v_{\phi}}{\partial \phi} + \frac{v_r v_{\phi}}{r} + \frac{v_{\phi} v_{\theta} \cot \theta}{r} \right)$ $\rho\left(\frac{\partial u_{\theta}}{\partial t} + v_{r}\frac{\partial u_{\theta}}{\partial r} + \frac{v_{\theta}}{r}\frac{\partial u_{\theta}}{\partial \theta} + \frac{v_{\phi}}{r\sin\theta}\frac{\partial u_{\theta}}{\partial \phi} + \frac{v_{r}u_{\theta}}{r} - \frac{v_{\phi}^{2}\cot\theta}{r}\right)$ $\rho\left(\frac{\partial v_r}{\partial t} + v_r\frac{\partial v_r}{\partial r} + \frac{v_\theta}{r}\frac{\partial v_r}{\partial \theta} + \frac{v_\phi}{r}\frac{\partial v_r}{\partial \theta} + \frac{v_\phi}{r\sin\theta}\frac{\partial v_r}{\partial \phi} - \frac{v_\theta^2 + v_\phi^2}{r}\right)$ CM3110 Fall 2011 Faith A. Morrison $\frac{\partial \rho}{\partial t} + \frac{1}{r} \frac{\partial (\rho r v_r)}{\partial r} + \frac{1}{r} \frac{\partial (\rho v_{\theta})}{\partial \theta} + \frac{\partial (\rho v_z)}{\partial z} = 0$ Continuity Equation, cylindrical coordinates Continuity Equation, Cartesian coordinates Continuity Equation, spherical coordinates

The Newtonian Constitutive Equation in Cartesian, Cylindrical, and Spherical coordinates

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Cartesian Coordinates

$$\begin{pmatrix} \tilde{\tau}_{xx} & \tilde{\tau}_{xy} & \tilde{\tau}_{xz} \\ \tilde{\tau}_{yx} & \tilde{\tau}_{yy} & \tilde{\tau}_{yz} \\ \tilde{\tau}_{zx} & \tau_{zy} & \tilde{\tau}_{zz} \end{pmatrix}_{xyz} = \mu \begin{pmatrix} 2\frac{\partial v_x}{\partial x} & \frac{\partial v_x}{\partial y} + \frac{\partial v_y}{\partial x} & \frac{\partial v_x}{\partial z} + \frac{\partial v_z}{\partial x} \\ \frac{\partial v_x}{\partial y} + \frac{\partial v_y}{\partial x} & 2\frac{\partial v_y}{\partial y} & \frac{\partial v_y}{\partial z} + \frac{\partial v_z}{\partial y} \\ \frac{\partial v_x}{\partial z} + \frac{\partial v_z}{\partial x} & \frac{\partial v_y}{\partial z} + \frac{\partial v_z}{\partial y} & 2\frac{\partial v_z}{\partial z} \end{pmatrix}_{xyz}$$

Cylindrical Coordinates

$$\begin{pmatrix} \tilde{\tau}_{rr} & \tilde{\tau}_{r\theta} & \tilde{\tau}_{rz} \\ \tilde{\tau}_{\theta r} & \tilde{\tau}_{\theta \theta} & \tilde{\tau}_{\theta z} \\ \tilde{\tau}_{zr} & \tau_{z\theta} & \tilde{\tau}_{zz} \end{pmatrix}_{r\theta z} = \mu \begin{pmatrix} 2\frac{\partial v_r}{\partial r} & r\frac{\partial}{\partial r} \left(\frac{v_{\theta}}{r}\right) + \frac{1}{r}\frac{\partial v_r}{\partial \theta} & \frac{\partial v_r}{\partial z} + \frac{\partial v_z}{\partial r} \\ r\frac{\partial}{\partial r} \left(\frac{v_{\theta}}{r}\right) + \frac{1}{r}\frac{\partial v_r}{\partial \theta} & 2\left(\frac{1}{r}\frac{\partial v_{\theta}}{\partial \theta} + \frac{v_r}{r}\right) & \frac{1}{r}\frac{\partial v_z}{\partial \theta} + \frac{\partial v_{\theta}}{\partial z} \\ \frac{\partial v_r}{\partial z} + \frac{\partial v_z}{\partial r} & \frac{1}{r}\frac{\partial v_z}{\partial \theta} + \frac{\partial v_{\theta}}{\partial z} & 2\frac{\partial v_z}{\partial z} \end{pmatrix}_{r\theta z}$$

Spherical Coordinates

$$\begin{pmatrix} \tilde{\tau}_{rr} & \tilde{\tau}_{r\theta} & \tilde{\tau}_{r\phi} \\ \tilde{\tau}_{\theta r} & \tilde{\tau}_{\theta \theta} & \tilde{\tau}_{\theta \phi} \\ \tilde{\tau}_{\theta r} & \tau_{\phi \theta} & \tilde{\tau}_{\phi \phi} \end{pmatrix}_{r\theta \phi} \\ = \mu \begin{pmatrix} 2 \frac{\partial v_r}{\partial r} & r \frac{\partial}{\partial r} \left(\frac{v_{\theta}}{r} \right) + \frac{1}{r} \frac{\partial v_r}{\partial \theta} & \frac{1}{r \sin \theta} \frac{\partial v_r}{\partial \phi} + r \frac{\partial}{\partial r} \left(\frac{v_{\phi}}{r} \right) \\ r \frac{\partial}{\partial r} \left(\frac{v_{\theta}}{r} \right) + \frac{1}{r} \frac{\partial v_r}{\partial \theta} & 2 \left(\frac{1}{r} \frac{\partial v_{\theta}}{\partial \theta} + \frac{v_r}{r} \right) & \frac{\sin \theta}{r} \frac{\partial}{\partial \theta} \left(\frac{v_{\phi}}{\sin \theta} \right) + \frac{1}{r \sin \theta} \frac{\partial v_{\theta}}{\partial \phi} \\ \frac{1}{r \sin \theta} \frac{\partial v_r}{\partial \phi} + r \frac{\partial}{\partial r} \left(\frac{v_{\phi}}{r} \right) & \frac{\sin \theta}{r} \frac{\partial}{\partial \theta} \left(\frac{v_{\phi}}{\sin \theta} \right) + \frac{1}{r \sin \theta} \frac{\partial v_{\theta}}{\partial \phi} & 2 \left(\frac{1}{r \sin \theta} \frac{\partial v_{\phi}}{\partial \phi} + \frac{v_r}{r} + \frac{v_{\theta} \cot \theta}{r} \right) \end{pmatrix}_{r\theta \phi}$$

These expressions are general and are applicable to three-dimensional flows. For unidirectional flows they reduce to Newton's law of Viscosity. Reference: Faith A. Morrison, *An Introduction to Fluid Mechanics* (Cambridge University Press: New York, 2013)

$ \begin{array}{llllllllllllllllllllllllllllllllllll$				215 (200°C)	109 (200°C)	372 (200°C)	31 (200°C)	45 (200°C)		18.9 (300°C)	57 (200°C)	G. Eckert and Chilton, Chemical and Critical Tables					
$Table A.3-16 Geankoplis, 2003$ $P = Re_{2n}f_{nr}$ $P = Re_{2n}f$		ties of Metals	$k(W/m \cdot K)$	206 (100°C)	104 (100°C)	377 (100°C)	33 (100°C)	45 (100°C)	21.6 (500°C)	16.3 (100°C)	59 (100°C)	ok Company, 1951; E. R. 9; R. H. Perry and C. H. arch Council Internation					
$P = Re_{nf}r_{n}$ $P = Re_{nf}$		and Heat Capaci		202 (0°C) 230 (300°C)	97 (0°C)	388 (0°C)	35 (0°C)	45.3 (18°C)	43 (300°C) 15.2 (100°C)	13.8 (0°C)	62 (0°C)	York: McGraw-Hill Boo Hill Book Company. 1956					
tanter (general), $D_{ij} = \frac{4A_{ij}}{p}$ $P = Re_{ij}/h_{ij}$ $P = Re_{ij}/h_{ij}$ A - 16 Thermal Conductivities, $1A - 16 Thermal Conductivities, 1A - 10 Thermal Conductivities, 1$, 2003	Densities,	$\frac{c_p}{\left(\frac{kJ}{kg\cdot K}\right)}$	0.896	0.385	0.383	0.130	0.473	0.461	0.461	0.227	ok. 5th ed. New York: McGraw-J Hill Book Comm					
tiameter (general), $D_{ii} = \frac{4A_{52}}{p}$ $P_0 = Re_{D_i} f_{D_i}$ $P_0 = Re_{D_i} f_{D_i}$ $A.3-16$ Thermal Condition ($e_{D_i} = \frac{4A_{52}}{126}$) $A.3-16$ Thermal Condition ($e_{D_i} = \frac{4A_{52}}{126}$) $A.3-16$ Thermal Condition ($e_{D_i} = \frac{1}{12}$) $A.3-120$ The Area Condition ($e_{D_i} = \frac{1}{12}$) A.3-120 The Area Conditio	ankoplis	activities, l	$\left(\frac{\rho}{m^3}\right)$	2707	8522 7503	8954	11 370	7801	7849	7817	7304	ineers' Handbo r, 2nd ed. New '	pany, 1929.				
diameter (general), $D_{ij} = \frac{4A_{ss}}{p}$ $Po = Re_{ij}f_{in}$ $Po = Re_{ij}f_{in}$ $A.0 \log\left(\frac{Re_{ij}\sqrt{f_{1n}}}{D_{10}}\right) - 0.40$ $A.0 \log\left(\frac{Re_{ij}\sqrt{f_{1n}}}{D_{10}}\right) - 0.40$ A.3-16 Therm A.3-16 Therm A.3-10 Therm A.3-	-16 Ge	al Condi	t (°C)	20	20	20	20	20	20	0	20	chanical Eng Mass Transfe 5th ed. New Y	ll Book Com				
diameter (general), $D_{ii} \equiv \frac{4A_{sc}}{p}$ $Po = Re_{Dif}D_{ii}$ $4.0 \log\left(\frac{Re_{Dif}J_{Dif}}{\overline{D0}_{dist}}\right) - 0.40$ $4.0 \log\left(\frac{Re_{Dif}J_{Dif}}{\overline{D0}_{dist}}\right) - 0.40$ $\frac{1}{2} \frac{1}{2} \frac{1}{2}$	Table A.3	A.3-16 Therm	Material	Aluminum	Brass (70–30) Cast iron	Copper	Lead	Steel 1%C	308 stainless	304 stainless	Tin	Source: L. S. Marks, Me R. M. Drake, Heat and Environet' Hondbook *	New York: McGraw-Hi				

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The Equation of Energy in Cartesian, cylindrical, and spherical coordinates for Newtonian fluids of constant density, with source term S. Source could be electrical energy due to	The Equation of Energy for systems with constant K
current flow, chemical energy, etc. Two cases are presented: the general case where thermal conductivity may be a function of temperature (vector flux $\tilde{q} = q/A$ appears in the equations); and the more usual case, where thermal conductivity is constant.	Microscopic energy balance , constant thermal conductivity; Gibbs notation
Fall 2013 Faith A. Morrison, Michigan Technological University	$ ho \hat{\mathcal{L}}_p\left(rac{\partial T}{\partial t} + \underline{v} \cdot \nabla T ight) = k \nabla^2 T + S$ Microscopic energy balance, constant thermal conductivity; Cartesian coordinates
Microscopic energy balance, in terms of flux, Gibbs notation $\rho\hat{C}_p\left(\frac{\partial T}{\partial t}+\underline{\nu}\cdot\nabla T\right)=-\nabla\cdot\bar{q}+S$	$\rho \hat{\mathcal{L}}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$
Microscopic energy balance, in terms of flux, Cartesian coordinates $\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_x \frac{\partial T}{\partial x} \right) = - \left(\frac{\partial \hat{q}_x}{\partial x} + \frac{\partial \hat{q}_y}{\partial y} + \frac{\partial \hat{q}_x}{\partial z} \right) + S$	$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$
Microscopic energy balance, in terms of flux; cylindrical coordinates $\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = -\left(\frac{1}{r} \frac{\partial (r \hat{q}_r)}{\partial r} + \frac{1}{r} \frac{\partial \hat{q}_\theta}{\partial z} + \frac{\partial \hat{q}_z}{\partial z} \right) + S$	$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{v} \frac{\partial T}{\partial \theta} + \frac{v_\theta}{r \sin \theta} \frac{\partial T}{\partial \phi} \right)$
Microscopic energy balance, in terms of flux; spherical coordinates $\alpha^{\hat{n}} \left(\frac{\partial T}{\partial t} + n \frac{\partial T}{\partial t} + \frac{v_{\theta}}{2} \frac{\partial T}{\partial t} + \frac{v_{\phi}}{2} \frac{\partial T}{\partial t} \right)_{= -} - \left(\frac{1}{2} \frac{\partial (r^2 \tilde{q}_{\tau})}{\partial t} + \frac{1}{2} \frac{\partial (\tilde{q}_{\theta} sin\theta)}{\partial sin\theta} + \frac{1}{2} \frac{\partial \tilde{q}_{\phi}}{\partial t} \right)_{+} \varsigma$	$= k \left(\frac{r^2}{r^2} \frac{\partial r}{\partial r} \left(r^2 \frac{\partial r}{\partial r} \right) + \frac{r^2}{r^2} \frac{\sin \theta}{\sin \theta} \frac{\partial \theta}{\partial \theta} \left(\sin \theta \frac{\partial \theta}{\partial \theta} \right) + \frac{r^2}{r^2} \frac{\partial \theta}{\sin^2 \theta} \frac{\partial \phi^2}{\partial \phi^2} \right) + S$
$\rho^{cp}(\partial t^{-v}\partial r^{-v}\partial r^{-v}r\partial \theta^{-v}r\partial \theta^{-v}r\sin\theta\partial\phi)^{-v}$	
Fourier's law of heat conduction, <code>Gibbs</code> notation: $ar{q}=-k abla T$	
Fourier's law of heat conduction, Cartesian coordinates: $\begin{pmatrix} \widehat{q}_x \\ \widehat{q}_y \\ \widehat{q}_z \end{pmatrix}_{xyz} = \begin{pmatrix} -k \frac{\partial T}{\partial x} \\ -k \frac{\partial T}{\partial y} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}_{xyz}$	
Fourier's law of heat conduction, cylindrical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_r \\ \tilde{q}_z \end{pmatrix}_{yyz} = \begin{pmatrix} -k \frac{\partial r}{\partial r} \\ -k \frac{\partial r}{\partial \sigma} \\ -k \frac{\partial r}{\partial \sigma} \end{pmatrix}_{r\thetaz}$	
Fourier's law of heat conduction, spherical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_{\theta} \\ \tilde{q}_{\theta} \end{pmatrix}_{xyz} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k \frac{\partial T}{\partial \theta}}{\sigma \theta} \end{pmatrix}_{r\theta\phi}$	Reference: F. A. Morrison, "Web Appendix to <i>An Introduction to Fluid Mechanics,</i> " Cambridge University Press, New York, 2013. On the web at <u>www.chem.mtu.edu/~fmorriso/IFM_WebAppendixCD2013.pdf</u>
1	2

			Specific	Volume	En	thalpy	E	ntropy		
	Temper-	Vapor	(m ³ /	/kg)	(k	J/kg)	(k)	/kg·K)		
an early states and a many strength of the state of the states of the states of the states of the states of the	ature (°C)	Pressure (kPa)	Liquid	Sat'd Vapor	Liquid	Sat'd Vapor	Liquid	Sat'd Vapor	Antiper (ana ang ang ang ang ang ang ang ang ang
	0.01	0.6113	0.0010002	206 136	÷ 0.00	2501.4	0 0000	0 1562		
	2	0.0115	0.0010002	168 132	12 57	2501.4	0.0000	9.1302		
	5	0.7377	0.0010001	137 734	25.20	2500.7	0.0437	9.0003		
	0	1 1/77	0.0010001	113 386	37.80	2512.4	0.0912	8 9253		
e en la segura de la calega de la	12	1 /022	0.0010005	03 784	50.41	2523.4	0.1302	8 8524		
	12	1.4022	0.0010005	77 926	62.99	2525.4	0.2245	8 7814		
	18	2 0640	0.0010014	65.038	75 58	2534.4	0.2679	8.7123		
	21	2.0040	0.0010020	54,514	88.14	2539.9	0.3109	8.6450		
	24	2.985	0.0010027	45.883	100.70	2545.4	0.3534	8.5794		
	24 25	3 169	0.0010029	43 360	104.89	2547.2	0.3674	8,5580		
	25	3 567	0.0010035	38 774	113.25	2550.8	0.3954	8 51 56		
	30	4 246	0.0010043	32 894	125.79	2556.3	0.4369	8.4533		
	33	5 034	0.0010053	28 011	138 33	2561.7	0.4781	8.3927		
	36	5 947	0.0010055	23,940	150.86	25671	0 5188	8 3336		
	<u>4</u> 0	7 384	0.0010078	19 523	167 57	2574.3	0.5725	8 2570		
	40	0 503	0.0010000	15 258	188.45	2583.2	0.6387	8,1648		
3	50	12 3/10	0.0010033	12.032	209.33	2505.2	0.7038	8 0763		
	55	15 758	0.0010121	9 568	230.23	2600.9	0.7679	7 9913		
	22	10.750	0.0010140	7 671	250.25	2600.2	0.7072	7 9096	3	
	65	25.03	0.0010172	6 197	272.06	2618.3	0.0012	7.8310		
	70	23.05	0.0010102	5.042	202.00	2626.8	0.0535	7.0510		
	70	39.59	0.0010228	1 131	313.03	2625.3	1 0155	7 6824		
	80	17 30	0.0010253	3 /07	334 01	2643.7	1.0155	7.0024		<u>8</u>
	85	57.83	0.0010291	2,407	355.90	2651.0	1 1 3 4 3	7.5445		
	00	70.1 <i>/</i>	0.0010325	2.020	376.92	2660.1	1.1975	7 4701		
	90	84.55	0.0010300	1 0810	307.06	2668 1	1.1525	7 4150		
	100	101 35	0.0010337	1.5015	419.04	2676.1	1 3060	7 3540		
	100	101.55	0.0010455	1.0725	440 15	2670.2	1 3630	7 2058	10	
	105	1/2 27	0.0010475	1.7107	461 30	2601.5	1 4185	7.2250		
<u>,</u>	115	140.27	0.0010510	1.2102	487.48	2600 0	1.4105	7 1833		
	120	109.00	0.0010553	0.8010	503 71	2000.0	1.5276	7.1000		
	120	190.33	0.0010640	0.0919	52/ 00	2700.5	1.5270	7.1200		
Ţ.	120	232.1	0.0010049	0.6685	5/6 31	2713.5	1.5015	7.0269		
	135	212.0	0.0010097	0.5822	567.69	2720,5	1.6970	6 9777		
5	ن 140 [%]	316.3	0.0010740	0.5022	580 13	2727.5	1 7301	6 9299		
	140 👷	510.5 415 4	0.0010797	0.2009	610 63	2735.9	1.7591	6 8833		
	145	415.4	0.0010850	0.3028	632.20	2746.5	1 8418	6 8379		
	155	47J.0 572 1	0.0010905	0.3468	653.84	2740.J 2752 A	1.0410	6 7035		
	160	617 Q	0.0010901	0.3400	675 55	2752.4	1 9427	6 7502		
	165	700 5	0.0011020	0.3071	607 3/	2750.1	1 0025	6 7078		
	170	701.5	0.0011000	0.2121	710 71	2769.2	2 0/10	6 6663		
	175	191.1	0.0011143	0.2420	7/1 17	2100.1 2772 6	2.0419	6 6256		
	190	092.U 1002 1	0.0011207	0.2100	762 22	2113.0 2778 2	2.0309	6 5 8 57		
	100	1002.1	0.0011274	0.19403	807 67	27786 1	2.1350	6 5070		
	190	1204.4	0.0011414	0.10004	857 15	2100.4	7 2200 7 2200	6 1272		
	200	2540	0.0011000	0.12/30	046 70	2173.2	2.2209	67507		
	223	2072	0.0011992	0.07849	900.78	2003.3	צנטנ.ג ררחד ר	6 0720		
	230	3772 ⁰	0.0012172	0.02013	1010 07	2001.3	2.1921	5 2020		
	213	3942 0501	0.0010426	0.05279	1210,07	2700.0	2.0208	5 7045		
	300	8281	0.0010436	0.02167	1544.0	2749.0	3.2334	5.7045		

A.2-9 Properties of Saturated Steam and Water (Steam Table), SI Units

Source: Abridged from J. H. Keenan, F. G. Keyes, P. G. Hill, and J. G. Moore, Steam Tables-Metric Units. New York: John Wiley & Sons, Inc., 1969. Reprinted by permission of John Wiley & Sons, Inc.

Т (°С)	Т (К)	р (kg/m³)	$(kJ/kg \cdot K)$	$\mu \times 10^{3}$ (Pa · s, or kg/m · s)	k (W/m • K)	Npr	$eta imes 10^4$ (1/K)	$(g\beta\rho^{2}/\mu^{2}) \times 10^{-8} \ (1/K \cdot m^{3})$
0	273.2	999.6	4.229	1.786	0.5694	13.3	-0.630	
15.6	288.8	998.0	4.187	1.131	0.5884	8.07	1.44	10.93
26.7	299.9	996.4	4.183	0.860	0.6109	5.89	2.34	30.70
37.8	311.0	994.7	4.183	0.682	0.6283	4.51	3.24	68.0
65.6	338.8	981.9	4.187	0.432	0.6629	2.72	5.04	256.2
93.3	366.5	962.7	4.229	0.3066	0.6802	1.91	6.66	642
121.1	394.3	943.5	4.271	0.2381	0.6836	1.49	8.46	1300
148.9	422.1	917.9	4.312	0.1935	0.6836	1.22	10.08	2231
204.4	477.6	858.6	4.522	0.1384	0.6611	0.950	14.04	5308
260.0	533.2	784.9	4.982	0.1042	0.6040	0.859	19.8	11 030
315.6	588.8	679.2	6.322	0.0862	0.5071	1.07	31.5	19 260

A.2-11 Heat-Transfer Properties of Liquid Water, SI Units

A.2-11 Heat-Transfer Properties of Liquid Water, English Units

Т (°F)	$\frac{\rho}{\left(\frac{lb_m}{ft^3}\right)}$	$ \begin{pmatrix} c_p \\ \frac{btu}{lb_m \cdot {}^\circ F} \end{pmatrix} $	$\frac{\mu \times 10^3}{\left(\frac{lb_m}{ft \cdot s}\right)}$	$\frac{k}{\left(\frac{btu}{h\cdot ft\cdot {}^\circ F}\right)}$	N _{Pr}	$eta imes 10^4 \ (1/^{\circ}R)$	$(g\beta\rho^2/\mu^2) \times 10^{-6} \ (1/^{\circ}R \cdot ft^3)$
32	62.4	1.01	1.20	0.329	13.3	-0.350	
60	62.3	1.00	0.760	0.340	8.07	0.800	17.2
80	62.2	0.999	0.578	0.353	5.89	1.30	48.3
100	62.1	0.999	0.458	0.363	4.51	1.80	107
150	61.3	1.00	0.290	0.383	2.72	2.80	403
200	60.1	1.01	0.206	0.393	1.91	3.70	1010
250	58.9	1.02	0.160	0.395	1.49	4.70	2045
300	57.3	1.03	0.130	0.395	1.22	5.60	3510
400	53.6	1.08	0.0930	0.382	0.950	7.80	8350
500 .	49.0	1.19	0.0700	0.349	0.859	11.0	17 350
600	42.4	1.51	0.0579	0.293	1.07	17.5	30 300

Geankoplis, 4th edition

NOTE: Equate the label to the provided quantity in the supplied units. For example, for <u>water</u> at $0^{o}C$:

$$\mu \times 10^3 = 1.786 Pa s$$

 $\mu = 1.786 \times 10^{-3} Pa s$

T (°C)	Т (К)	ρ (kg/m³)	c _p (kJ/kg·K)	µ. × 10 ⁵ (Pa · s, or kg/m · s)	k (W/m · K)	Npr	$\beta \times 10^3$ $(1/K)$	$\frac{g\beta\rho^2/\mu^2}{(1/K\cdot m^3)}$
-17.8	255.4	1.379	1.0048	1.62	0.02250	0.720	3.92	2.79×10^{8}
0	273.2	1.293	1.0048	1.72	0.02423	0.715	3.65	2.04×10^{8}
10.0	283.2	1.246	1.0048	1.78	0.02492	0.713	3.53	1.72×10^{8}
37.8	311.0	1.137	1.0048	1.90	0.02700	0.705	3.22	1.12×10^{8}
65.6	338.8	1.043	1.0090	2.03	0.02925	0.702	2.95	0.775×10^{8}
93.3	366.5	0.964	1.0090	2.15	0.03115	0.694	2.74	0.534×10^{8}
121.1	394.3	0.895	1.0132	2.27	0.03323	0.692	2.54	0.386×10^{8}
148.9	422.1	0.838	1.0174	2.37	0.03531	0.689	2.38	0.289×10^{8}
176.7	449.9	0.785	1.0216	2.50	0.03721	0.687	2.21	0.214×10^{8}
204.4	477.6	0.740	1.0258	2.60	0.03894	0.686	2.09	0.168×10^{8}
232.2	505.4	0.700	1.0300	2.71	0.04084	0.684	1.98	0.130×10^{8}
260.0	533.2	0.662	1.0341	2.80	0.04258	0.680	1.87	0.104×10^{8}

A.3-3 Physical Properties of Air at 101.325 kPa (1 Atm Abs), SI Units

A.3-3 Physical Properties of Air at 101.325 kPa (1 Atm Abs), English Units

	ρ	c_{ρ}		k			
T (°F)	$\left(\frac{lb_m}{ft^3}\right)$	$\left(\frac{btu}{lb_m\cdot {}^{\circ}F}\right)$	μ (centipoise)	$\left(\frac{btu}{h \cdot ft \cdot {}^{\circ}F}\right)$	N _{Pr}	$\beta \times 10^{3}$ $(1/^{\circ}R)$	$g\beta\rho^2/\mu^2$ $(1/^{\circ}R\cdot ft^3)$
0	0.0861	0.240	0.0162	0.0130	0.720	2.18	4.39 × 10 ⁶
32	0.0807	0.240	0.0172	0.0140	0.715	2.03	3.21×10^{6}
50	0.0778	0.240	0.0178	0.0144	0.713	1.96	2.70×10^{6}
100	0.0710	0.240	0.0190	0.0156	0.705	1.79	1.76×10^{6}
150	0.0651	0.241	0.0203	0.0169	0.702	1.64	1.22×10^{6}
200	0.0602	0.241	0.0215	0.0180	0.694	1.52	0.840×10^{6}
250	0.0559	0.242	0.0227	0.0192	0.692	1.41	0.607×10^{6}
300	0.0523	0.243	0.0237	0.0204	0.689	1.32	0.454×10^{6}
350	0.0490	0.244	0.0250	0.0215	0.687	1.23	0.336×10^{6}
400	0.0462	0.245	0.0260	0.0225	0.686	1.16	0.264×10^{6}
450	0.0437	0.246	0.0271	0.0236	0.674	1.10	0.204×10^{6}
500	0.0413	0.247	0.0280	0.0246	0.680	1.04	0.163×10^{6}

Source: National Bureau of Standards. Circular 461C, 1947; 564, 1955: NBS-NACA. Tables of Thermal Properties of Gases. 1949; F. G. Keyes, Trans. A.S.M.E., 73, 590, 597 (1951); 74, 1303 (1952); D. D. Wagman, Selected Values of Chemical Thermodynamic Properties. Washington, D.C.: National Bureau of Standards. 1953.

Geankoplis, 4th edition

NOTE: Equate the label to the provided quantity in the supplied units. For example, for <u>air</u> at $0^{\circ}C$:

 $\mu \times 10^5 = 1.72 Pa s$ $\mu = 1.72 \times 10^{-5} Pa s$

Heat Transfer Data Correl	ations for Examinations		
CM3110 Transport Phenomena I Michigan Technological University Professor Faith A. Morrison 1 December 2020			Log mean driving force
l. Forced Convection Through	Pipes		II. Forced Conve
In forced convection, we determined function of at most Re, Pr, L/D , and v	from dimensional analysis that the Nusselt number is a viscosity ratio.		In heat transfer takin cylinder with wall ten
Prandtl number (fluid properties)	$\Pr \equiv \frac{\hat{c}_p \mu}{k}$	(1)	Film tempera
In pipe flow with heat transfer taking at T_{bo} . T_{w} is the temperature of the n pipes, all fluid material properties exc. The mean bulk temperature is given t	place, the fluid enters at bulk fluid temperature T_{bi} and exiwall. For Nu data correlations in forced convection through cept $\mu_w=\mu(T_w)$ are evaluated at the mean bulk temperatuby	s: -5	The data correlation I Outside Cylinder
Mean bulk temperature	$\bar{T}_b \equiv \frac{T_{bi} + T_{bo}}{2}$	(2)	Wall-bulk driving force
A. Laminar Flow in Pipes			The values of C and π
Sieder and Tate's correlation (Geankc	oplis, p260) for laminar flow is		values are valid for P
Laminar flow $Nu_a =$	$= \frac{h_a D}{k} = 1.86 \left(\text{RePr} \frac{D}{L} \right)^{\frac{3}{2}} \left(\frac{\mu_b}{\mu_w} \right)^{0.14}$	(3)	
	$q = h_a A \Delta T_a$	(4)	
Arithmetic mean Δ^{\prime}	$T_a = \frac{(T_w - T_{bi}) + (T_w - T_{bo})}{2}$	(2)	
B. Turbulent Flow in Pipes			
Sieder and Tate's correlation (Geankc	pplis, p261) for turbulent flow is		
Turbulent flow $Nu_{lm} =$	$=\frac{\hbar_{im}D}{k}=0.027 \mathrm{Re}^{0.8}\mathrm{Pr}_{3}^{\frac{1}{2}} \left(\frac{\mu_{b}}{\mu_{w}}\right)^{0.14}$	(6)	
	$q = h_{tm} A \Delta T_{tm}$	(7)	
		Ļ	



he data correlation for Nusselt number in this case is

utside Cylinder Nu =
$$\frac{\hbar D}{k} = C R e^m P r^{\frac{1}{3}}$$
 (10)

$$q = hA(T_w - T_b) \tag{11}$$

The values of C and m depend on the Reynolds number (Geankoplis, Table 4.6-1, p272). These values are valid for \Pr > 0.6.

U E	0.330 0.989	0.385 0.911	0.466 0.683	0.618 0.193	0.805 0.0266
Re	1 - 4	4 - 40	40 - 4,000	$4,000 - 4 \times 10^4$	$4 \times 10^4 - 2.5 \times 10^5$

2

TABLE 4.7-2. Simplified Equations for Natural Convection from Various Surfaces ,		$\begin{array}{cccc} h = b u / h^2 \cdot F & h = W/m^2 \cdot K \\ L = \beta h \Delta T = {}^\circ F & L = m, \Delta T = K \\ Physical Geometry & N_{Gr} N_{\rm Pr} & D = \beta & D = m \\ \end{array}$	A is at 101 23 1-De (1 atm) also arrange	Vertical planes and $10^{4}-10^{9}$ h = $0.28(\Lambda TL)^{14}$ h = $1.37(\Lambda TL)^{14}$ (P1) cylinders $>10^{9}$ h = $0.18(\Lambda TL)^{13}$ h = $1.24 \Delta T^{13}$ (P1)	Horizontal cylinders $10^{2}-10^{9}$ $h = 0.27(\Delta T/D)^{1/4}$ $h = 1.32(\Delta T/D)^{1/4}$ (M1) $>10^{9}$ $h = 0.18(\Delta T)^{1/3}$ $h = 1.24 \Delta T^{1/3}$ (M1)	Horizontal plates Horizontal plates Heated plate facing $10^5 - 2 \times 10^7$ $h = 0.27 (\Delta T/L)^{14}$ $h = 1.32 (\Delta T/L)^{14}$ (M1) upward or cooled $2 \times 10^7 - 3 \times 10^{10}$ $h = 0.22 (\Delta T)^{13}$ $h = 1.52 \Delta T^{13}$ (M1) nlate facing $0.22 (\Delta T)^{13}$ $h = 1.52 \Delta T^{13}$ (M1)	part atom downward	Heated plate facing $3 \times 10^{5} - 3 \times 10^{10}$ $h = 0.12(\Delta T/L)^{14}$ $h = 0.59(\Delta T/L)^{14}$ (M1)	downward or cooled plate	facing upward Water at 70°F (294 K)	Vertical planes and $10^4 - 10^9$ $h = 26(\Delta T/L)^{1/4}$ $h = 127(\Delta T/L)^{1/4}$ (P1)	cylinders	Organic liquids at 70°F (294 K) \oplus	Vertical planes and $10^{-10^{\circ}}$ $n = 1.2(\Delta I/L)^{\circ}$ $n = 3.9(\Delta I/L)^{\circ}$ (r1) cylinders									Reference : C. I. Geankonlis. Transport Processes and Generation Process Principles. 4 th Edition	reference: c. J. Geannopins, inansport modesses and separation modes a minicipites, 4 – curron, Prentice Hall, 2003.	
	,	isional	(12)	(13)		ations			Ref.		(B3)	(cr) (IM)	(M1)			(P3)	(P3)	(P3)	(P3)	(M1)	(сл)	(M1) (M1)	(++++)	(F1)	
		en found by dimer.			, elder ni silnodnee	earing of the corre		ion	a m		36 1	.59 5 4	.13 1			.49 0	$.71 \frac{1}{25}$	$\frac{1}{10}$.09 ⁵	.53 4 ⁴	<u>£</u> cr.	.54 14		.58 <u>1</u> 3	
ometries		s from various surfaces have be ollows:	$\frac{hL}{r} = a(\text{Gr Pr})^m$	$\kappa = \frac{L^3 \rho^2 g \beta \Delta T}{2}$	μ^{z}	recty, values may be round in o kt pages) provides simplified ve anic liquids).		q. (4.7-4) for Natural Conveci	$N_{ m Gr} N_{ m Pr}$		104	$\sim 10^{4} - 10^{9}$ 0.	>10 ⁹ 0.			<10 ⁻⁵ 0.	$10^{-5} - 10^{-3}$ 0.	10 ⁻³ -1 1.	1-104 1.	10 ⁴ -10 ⁷ 0.	-01~	$10^{5}-2 \times 10^{7} \qquad 0.$ 2 × 10 ⁷ -3 × 10 ¹⁰ 0		$10^{5}-10^{11}$ 0.	
III. Natural Convection from Various Ge		Natural convection heat transfer coefficients analysis and experimentally to correlate as ft	Natural convection Nu =	(various geometries) Grashof number Gr	The valuet for a and m denend on the server	ine values for us and increpted on the geometry of (p278, shown below). Table 4.7-2 (p280, ne) specialized to common fluids (air, water, org	; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;	TABLE 4.7-1. Constants for Use with Eq.	Physical Geometry	Vertical planes and cylinders [vertical height $I < 1 \text{ m} (3 \text{ ft})$]				Horizontal cylinders Idiameter D used for	L and D < 0.20 m (0.66 ft)]	-					Horizontal nlates	Upper surface of heated plates or lower surface of cooled nlates	I ower surface of heated plates or	upper surface of cooled plates	





(b)

FIGURE 4.9-4. Correction factor F_T to log mean temperature difference: (a) 1–2 and 1–4 exchangers, (b) 2–4 exchangers. [From R. A. Bowman, A. C. Mueller, and W. M. Nagle, Trans. A.S.M.E., **62**, 284, 285 (1940). With permission.]



FIGURE 4.9-5. Correction factor F_T to log mean temperature difference for cross-flow exchangers $[Z = (T_{hi} - T_{ho})/(T_{co} - T_{ci})]$: (a) single pass, shell fluid mixed, other fluid unmixed, (b) single pass, both fluids unmixed. [From R. A. Bowman, A. C. Mueller, and W. M. Nagle, Trans. A.S.M.E., **62**, 288, 289 (1940). With permission.]



exchanger.

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<u>Material</u>	<u>8</u>
Aluminum foil	0.04
Asbestos board	0.96
Polished brass	0.03
Cast iron, turned and heated	0.60-0.70
Concrete	0.85
Ice, smooth	0.966
Ice, rough	0.985
Plaster	0.98
Roofing paper	0.91
Sand	0.76
Steel, Oxidized	0.79
Wrought Iron	0.94

Table 1: Emissivity ε of solids (300K)

$$\sigma = 0.1712 \times 10^{-8} \frac{BTU}{h f t^2 R^4}$$
$$\sigma = 5.676 \times 10^{-8} \frac{W}{m^2 K^4}$$

Reference: Engineering Toolbox, <u>www.engineeringtoolbox.com/emissivity-coefficients-d_447.html</u>

	h_d $(W/m^2 \cdot K)$	h_d (btu/h · ft ² · °F)
Distilled and seawater	11 350	2000
City water	5680	1000
Muddy water	1990 - 2840	350-500
Gases	2840	500
Vaporizing liquids	2840	500
Vegetable and gas oils	1990	350

TABLE 4.9-1. Typical Fouling Coefficients (P3, N1)

Reference: Geankoplis, 4th Edition