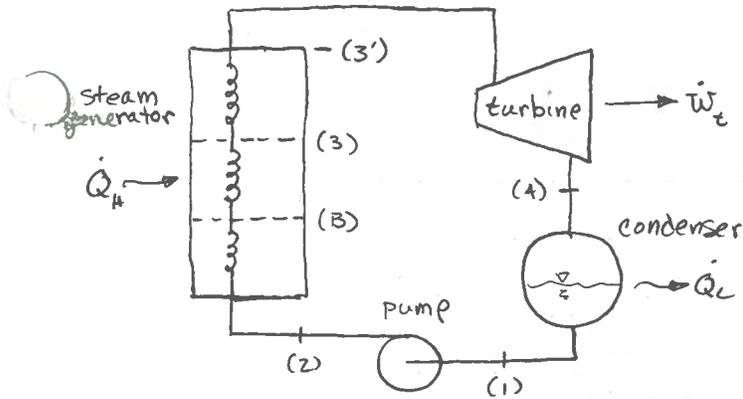


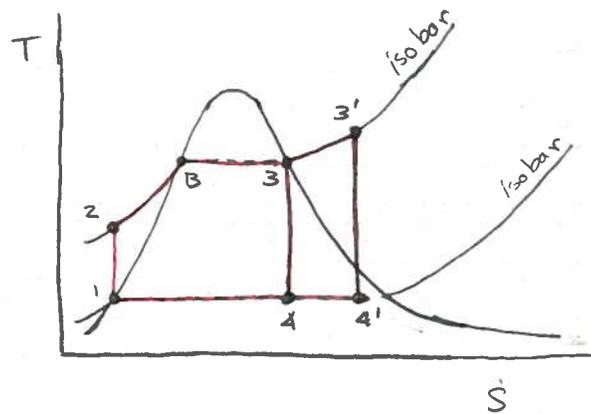
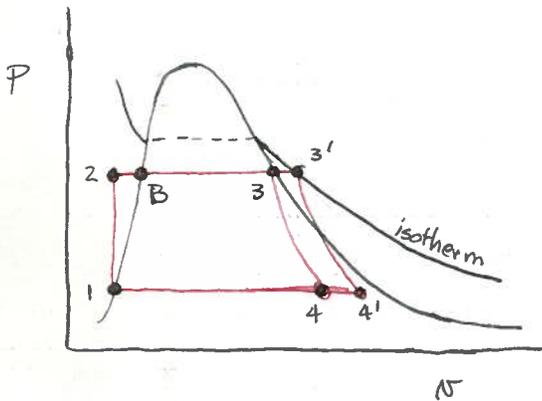
Ideal Rankine Cycle (all processes internally reversible)



- heat exchangers used for heat addition & rejection (exception is steam locomotives)
- external heating/combustion
 - generally less polluting than internal
 - can use lower grade fuels
 - can use solar, nuclear & geothermal thermal energy sources

- common working fluids:
 - water (most common)
 - potassium
 - sodium
 - rubidium
 - ammonia
 - chain & aromatic hydrocarbons
 - fluorocarbons

- 1-2: Isentropic Compression (Pump)
- 2-3: Isobaric Heat Addition (Steam Generator)
- 3-4: Isentropic Expansion (Turbine)
- 4-1: Isobaric Heat Rejection (Condenser)



Saturated Rankine Cycle: 1-2-B-3-4-1

Superheated Rankine Cycle: 1-2-B-3-3'-4'-1

Steam Generator:

- State (B) is saturated liquid
- 2-B Economizer → brings subcooled liquid up to saturation temperature
- B-3 Boiler/Evaporator → saturated liquid to saturated vapor
- 3-3' Superheater → saturated vapor to superheated vapor

Analyzing Rankine Cycle requires determination of enthalpies:

$$1^{st} \text{ Law: } q - w = \Delta e = \left[\underbrace{(u + Pv)}_h + \frac{1}{2}v^2 + gz \right] \Big|_{out} - \left[(u + Pv) + \frac{1}{2}v^2 + gz \right] \Big|_{in}$$

• generally can neglect kinetic & potential energy changes

$$\boxed{q - w = \Delta h}$$

Processes:

1-2 isentropic compression (pump): $-w_p = h_2 - h_1$

2-3,3' isobaric heat addition (steam generator): $q_H = h_{3,3'} - h_2$

3,3'-4,4' isentropic expansion (turbine): $-w_t = h_{4,4'} - h_{3,3'}$

4,4'-1 isobaric heat rejection (condenser): $q_L = h_1 - h_{4,4'}$ ← not present on steam locomotives

net work, $w_{net} = w_t - |w_p| = (h_{3,3'} - h_{4,4'}) - (h_2 - h_1)$

thermal efficiency, $\eta_{th} = \frac{w_{net}}{q_H} = \frac{(h_{3,3'} - h_{4,4'}) - (h_2 - h_1)}{(h_{3,3'} - h_2)}$

work ratio $\equiv \frac{\text{net work}}{\text{gross work}} = \frac{w_{net}}{w_t}$

pump work (1-2)

• For small units, where P_2 is not too large compared to P_1 , $h_2 \approx h_3$
 \Rightarrow pump work is negligible compared to turbine work

• For modern power plants, $P_2 \geq 1000 \text{ psia (70 bar)}$ & $P_1 \sim 1 \text{ psia (0.07 bar)}$
 h_1 is the saturated liquid enthalpy at P_1 (h_{1f})

h_2 is found from subcooled liquid tables at T_2 & P_2

• generally, $T_2 \approx T_1$ and the latter is used in lieu of T_2 which is difficult to obtain

A good approximation for the pump work can be obtained from the change in "flow work"

$$|w_p| = v_1 (P_2 - P_1)$$

↑ $v_2 = v_1$ for incompressible fluid

$$h_2 - h_1 = \underbrace{(u_2 - u_1)}_{\approx 0 \text{ since } T_2 \approx T_1} + \underbrace{(P_2 v_2 - P_1 v_1)}_{v_1 (P_2 - P_1)}$$

Simple Rankine Cycle

Find the thermal efficiency and the specific work of a simple Rankine cycle if the maximum temperature and pressure are 540°C and 7.0 MPa and the minimum pressure is 10 kPa . The turbine and pump efficiencies are both 85% .

State (1) - saturated liquid

$$h_1 = h_f(P_1 = 10\text{ kPa}) = 191.53\text{ kJ/kg}$$

$$v_1 = v_f(P_1 = 10\text{ kPa}) = 0.0010102\text{ m}^3/\text{kg}$$

State (2) - subcooled (compressed) liquid

$$h_2 = h_1 + (w_p)$$

$$|w_p| \approx v_1 (P_2 - P_1) \cdot \left(\frac{1}{0.85} \right)$$

$$h_2 = 191.53\text{ kJ/kg} + 8.31\text{ kJ/kg} = 199.84\text{ kJ/kg}$$

State (3) - superheat vapor

$$h_3 = h(T = 540^\circ\text{C}, P = 7\text{ MPa}) = 3506.9\text{ kJ/kg}$$

$$s_3 = 6.9193\text{ kJ/kgK}$$

State (4) - saturated liquid & vapor ($0 < x < 1$)

$$\eta_t = \frac{h_3 - h_4}{h_3 - h_{4s}} \rightarrow h_4 = h_{4s} + (1 - \eta_t)(h_3 - h_{4s})$$

$$s_{4s} = s_3$$

$$M_{4s} = \frac{s_{4g}(P = 10\text{ kPa}) - s_{4s}}{s_{4g}(P = 10\text{ kPa}) - s_{4f}(P = 10\text{ kPa})} = \frac{8.1502\text{ kJ/kgK} - 6.9193\text{ kJ/kgK}}{7.5009\text{ kJ/kgK}} = 0.1641 \Rightarrow x_{4s} = 1 - M_{4s}$$

$$h_{4s} = h_{4g} - x_{4s}(h_{4g} - h_{4f}) = 2584.7\text{ kJ/kg} - 0.1641(2392.8\text{ kJ/kg}) = 2192.0\text{ kJ/kg}$$

$$h_4 = 2192.0\text{ kJ/kg} + (0.15)(3506.9\text{ kJ/kg} - 2192.0\text{ kJ/kg}) = 2389.2\text{ kJ/kg}$$

• actual moisture content at turbine

$$M_A = \frac{h_{4g}(P = 10\text{ kPa}) - h_4}{h_{4g}(P = 10\text{ kPa}) - h_{4f}(P = 10\text{ kPa})} = \frac{2584.7\text{ kJ/kg} - 2389.2\text{ kJ/kg}}{2392.8\text{ kJ/kg}} = 0.0817$$

• net specific work

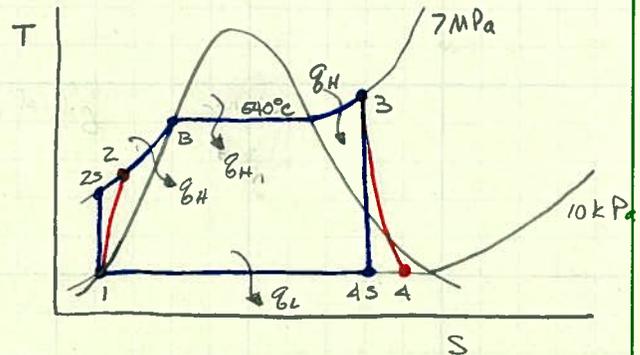
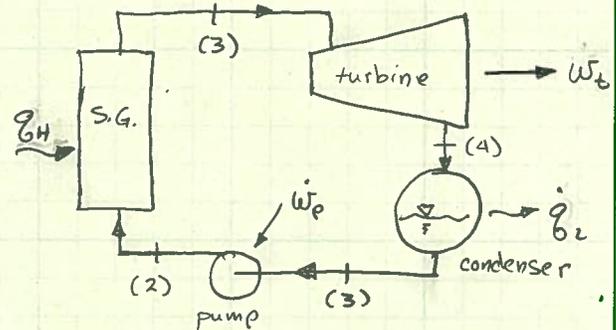
$$w_{\text{net}} = \underbrace{(h_3 - h_4)}_{\text{turbine}} - \underbrace{(h_2 - h_1)}_{\text{pump}} = \left[3506.9\frac{\text{kJ}}{\text{kg}} - 2389.2\frac{\text{kJ}}{\text{kg}} \right] - \left[199.84\frac{\text{kJ}}{\text{kg}} - 191.8\frac{\text{kJ}}{\text{kg}} \right]$$

$$w_{\text{net}} = 1109.7\text{ kJ/kg}$$

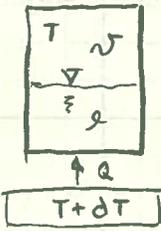
• thermal efficiency

$$\eta_{\text{th}} = 1 - \frac{q_L}{q_H} = 1 - \frac{(h_1 - h_4)}{(h_3 - h_2)} = 1 - \left[\frac{191.8\frac{\text{kJ}}{\text{kg}} - 2389.2\frac{\text{kJ}}{\text{kg}}}{3506.9\frac{\text{kJ}}{\text{kg}} - 199.84\frac{\text{kJ}}{\text{kg}}} \right]$$

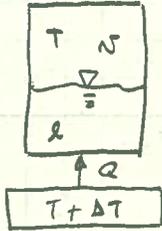
$$\eta_{\text{th}} = 33.55\%$$



Rankine Cycle - External Irreversibilities



reversible heat transfer, external & internal

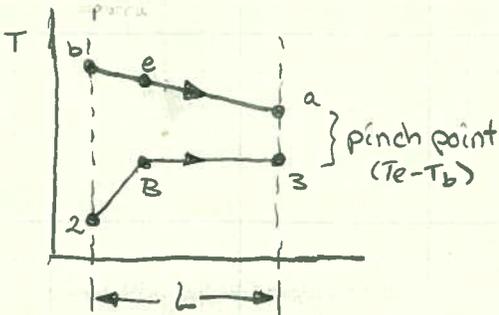
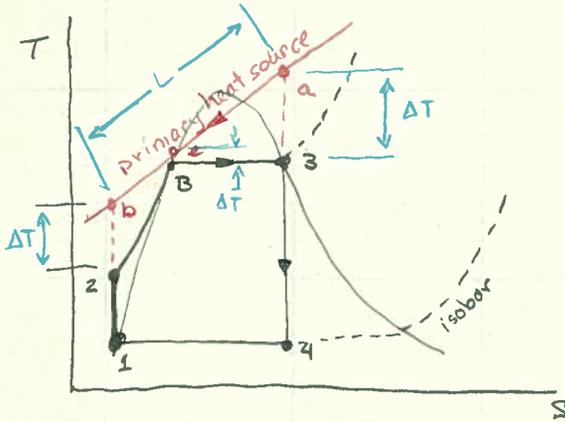


internally reversible heat transfer, but not externally

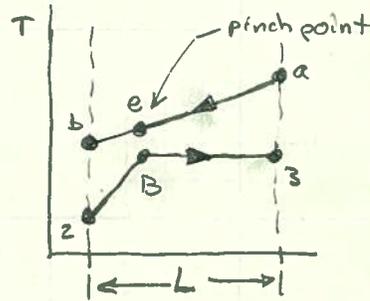
evaporation/condensation is a reversible process
external irreversibilities primarily due to finite temperature difference, ΔT_{ext}

For steam Generator, external irreversibilities primarily due to ΔT between primary heat sources & working fluid.

- combustion gases
- primary coolant from nuclear reactor
- solar concentrator
- steam
- ammonia
- refrigerants



parallel flow (co-flow) steam generator



counter flow steam generator

- The net ΔT is greater for parallel flow than for counterflow.
- Heat transfer process is also more efficient with counterflow.

- Use Counter Flow -

Pinch-Point Temperature Difference, $T_e - T_B$:

- small pinch-point \rightarrow • lower overall temperature difference between steam lines & primary heat source lines; results in lower external irreversibilities
 - requires large, costly steam generators
- large pinch-point \rightarrow • small, inexpensive steam generators
 - higher overall temperature difference & external irreversibilities; reduction in plant efficiency
- most economical pinch-point ΔT is obtained by considering both capital costs and recurring costs.

\uparrow price of steam generator

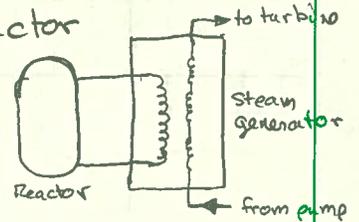
\uparrow operating & fuel costs

Types of Heat Source Fluids for Steam Generator :

- gas \rightarrow • combustion gases
 - primary coolant from a gas-cooled reactor, CO_2 or He
- liquid \rightarrow • water; pressurized-water reactor (PWR)
 - liquid metal; liquid-metal fast-breeder reactor

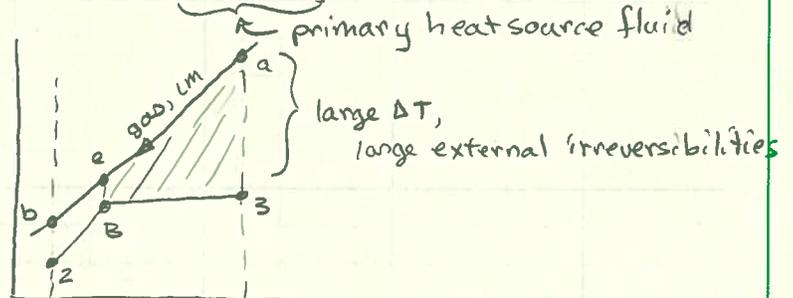
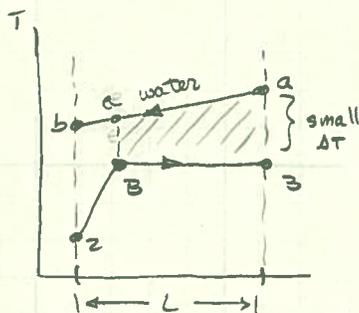
Heat Transfer: $\dot{Q} = \dot{m} c_p \Delta T$

\uparrow Heat Source Fluids have a variety of c_p & \dot{m} required to achieve \dot{Q}

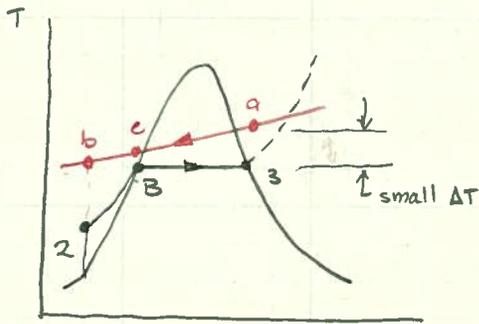


(PWR) water has a higher c_p than gas, but also has a much higher \dot{m} in order to limit temperature rise in water flowing through the reactor; need to maintain uniform moderation of neutrons
 $\therefore \dot{m} c_p$ much greater for water than for gases

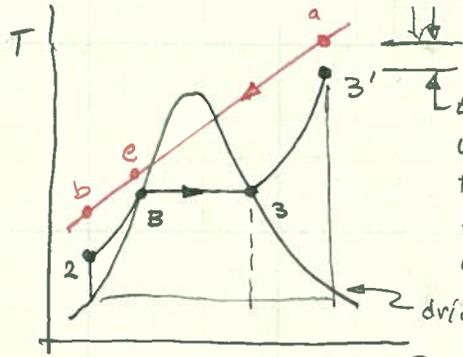
$\frac{dQ}{dz} = \dot{m} c_p \frac{dT}{dz} \rightarrow$ slope of line a-b is proportional to $\frac{1}{\dot{m} c_p}$
 or $\frac{dT}{dz} \sim \frac{1}{\dot{m} c_p}$ for a fixed \dot{Q}



To Superheat or Not To Superheat (& Reheat)?



water as primary heat source fluid
 → PWR, BWR (?)

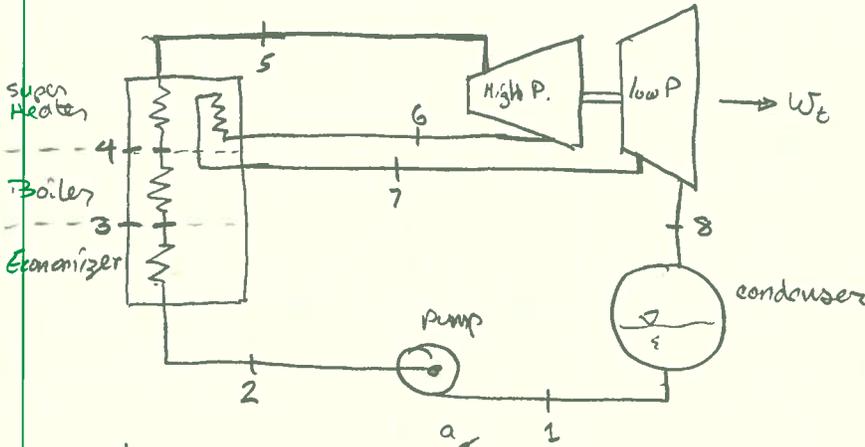


gas or liquid metal as primary heat source fluid
 → fossil fuel
 gas-cooled reactors
 liquid-metal-cooled reactors

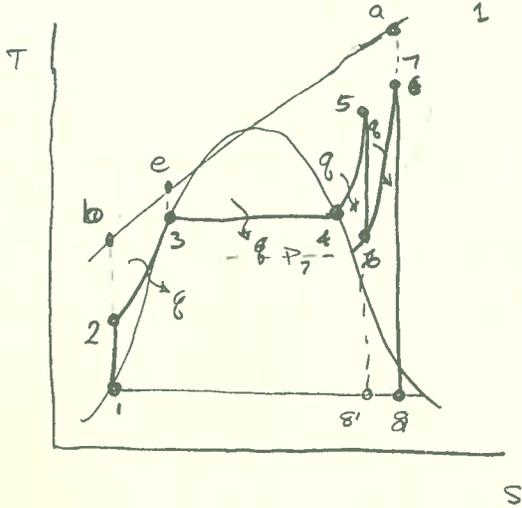
↑ more efficient & less prone to blade damage

ΔT reduced by using superheat; therefore, external irreversibilities also reduced

REHEAT



- Improved efficiency
- drier steam at exhaust
- all modern power plants use steam heat & at least 1 stage of reheat



$$-w_t = (h_6 - h_5) + (h_8 - h_7)$$

$$-w_p = (h_2 - h_1)$$

$$w_{net} = [h_5 - h_6 + h_8 - h_7] - [h_2 - h_1]$$

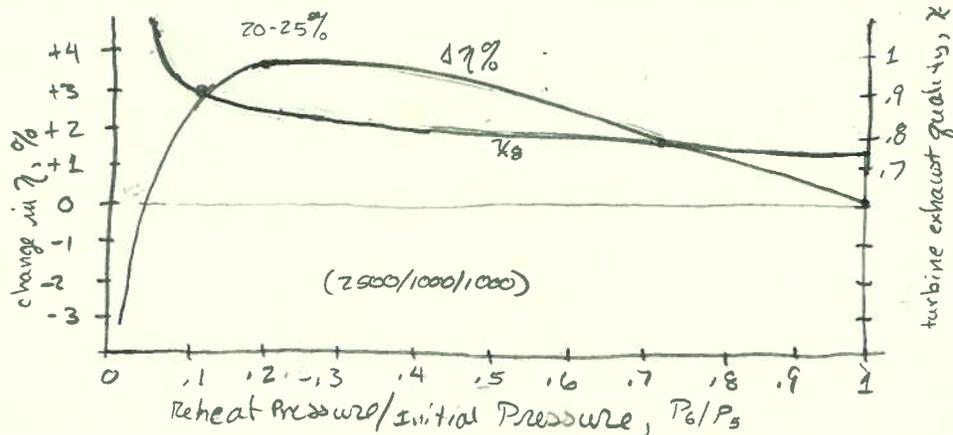
$$q_h = (h_5 - h_2) + (h_7 - h_6)$$

$$\eta_{th} = \frac{w_{net}}{q_h} = \frac{h_5 - h_6 + h_8 - h_7 - h_2 + h_1}{h_5 - h_2 + h_7 - h_6}$$

Superheat Reheat Power Plant often designated as $P_5/T_5/T_6$

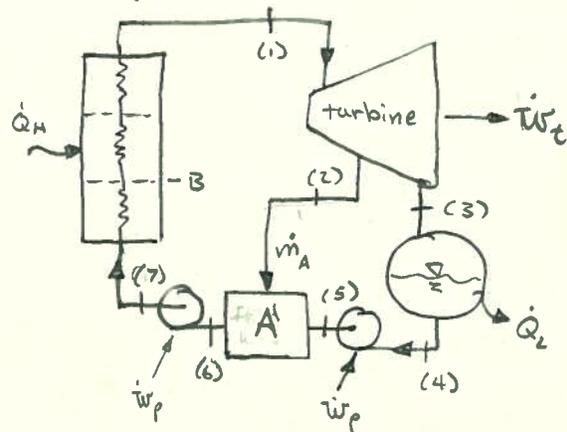
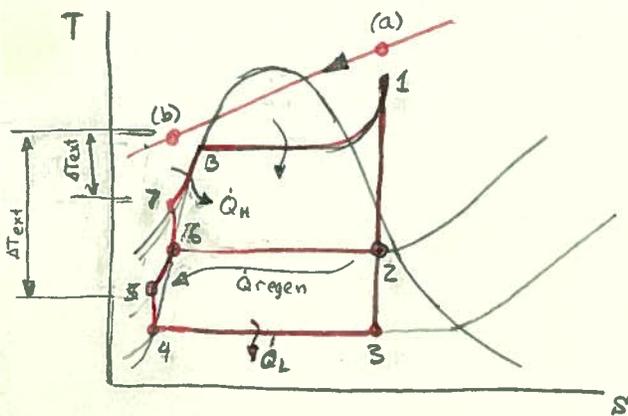
turbine inlet

- The pressure at which the steam is reheated is critical. A reheat pressure ($P_6 = P_7$) too close to the initial pressure ($P_2 = P_3 = P_4 = P_5$) results in little increase to the cycle efficiency. The efficiency increase due to reheat improves as the reheat pressure ($P_6 = P_7$) is lowered, reaching a peak at a pressure ratio P_6/P_5 of 0.2 to 0.25.
- Too low of a pressure ratio could result in superheated exhaust steam, unfavorable for condensers.



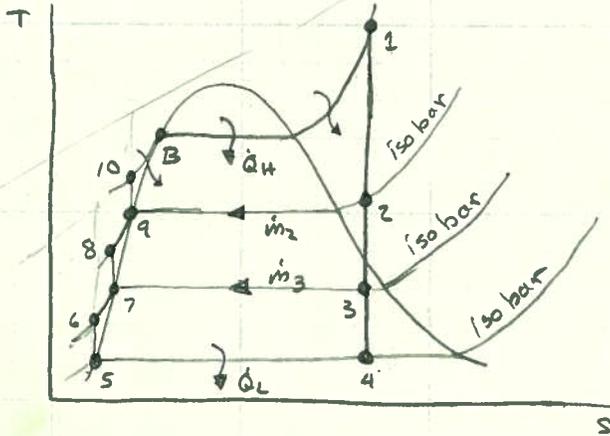
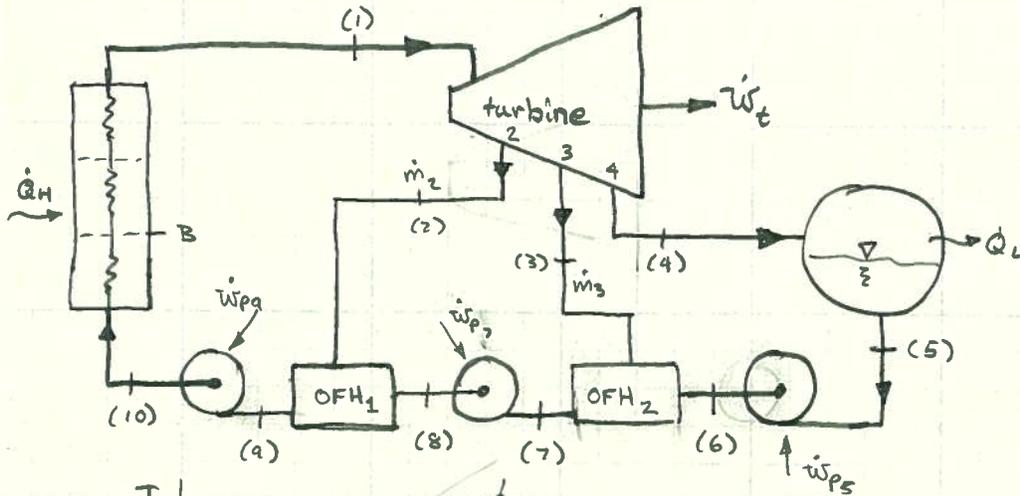
Regeneration - Feedwater Heating

- Superheat addresses external irreversibilities in the boiler.
- Regeneration would address external irreversibilities in the economizer;
 - transfer heat from exhaust side of cycle to compressed liquid prior to heat addition process.
 - For the Rankine cycle, heat rejection at the condenser is isothermal and below the temperature of the compressed liquid; can not transfer thermal energy from a cold fluid to a warm fluid.
- Regeneration is therefore accomplished by "bleeding" a small amount of steam from the turbine and heating the "feedwater" (compressed liquid) prior to the feedwater entering the steam generator.



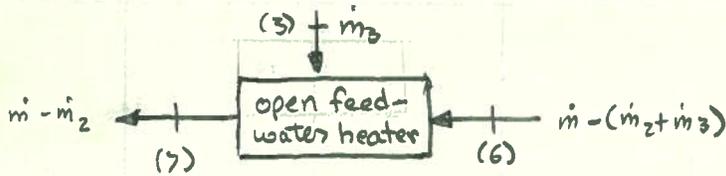
- Feedwater is the compressed liquid flowing towards the steam generator
- compressed liquid at (5) is heated with small amount of steam (2) to bring the feedwater back to saturated liquid state (6)
- heat transfer takes place in a heat exchanger \rightarrow "feedwater heater"
- dates back to the 1920's, same time period that steam temperatures reached 725°F
- modern power plants use between 5 and 8 feedwater heaters
- still need an economizer (7-B), but a much smaller one than without feedwater heating; compare the temperature difference from (b) to (7) and from (b) to (5).
- three types of feedwater heaters:
 - open (direct contact) \rightarrow steam is mixed with feedwater
 - closed with drains cascading backwards
 - closed with drains cascading forwards
 } steam is run through a tube-and-shell heat exchanger.

Open Feedwater Heater



- Large amount of energy released from extracted steam as it condenses; small amount of energy required to bring feedwater back to saturated liquid
- Therefore, m_2 and m_3 are small fractions of the total m through the turbine.

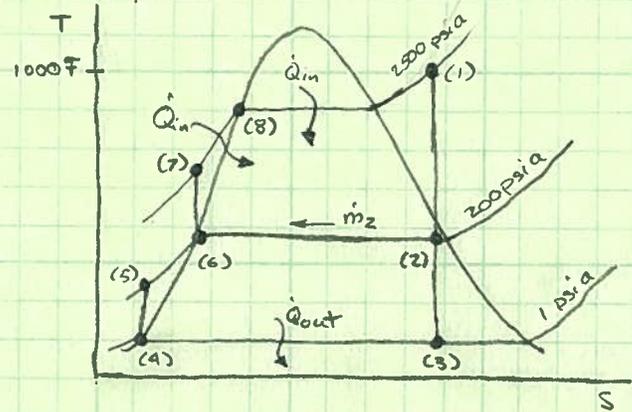
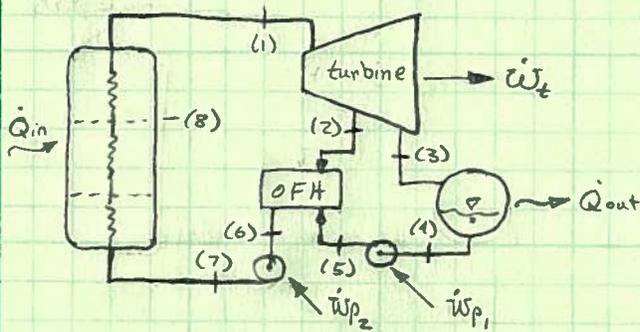
- Steam is "bled off" at (2) and (3) and mixed with subcooled water at (6) and (8) to produce saturated water at the temperature of the bled steam.



- state (3) & state (6) mix to become state (7)

- The amount of steam bled, m_3 , is that which will saturate the subcooled liquid, $m - (m_2 + m_3)$
- $P_6 = P_7 \leq P_3$ (reverse flow into turbine would occur if $P_7 > P_3$)
- requires a pump after the feedwater heater & each pump has to handle nearly all of the mass flow in cycle
- open feedwater heaters also serve as deaerators to remove non-condensable gases such as O_2 , H_2 , and CO_2 .

An ideal Rankine cycle operates between 2500 psia and 1000°F at throttle and 1 psia in the condenser. One open feed water heater is placed at 200 psia. Calculate the net work and heat input in Btu/lbm.



(1) $T_1 = 1000^\circ\text{F}$ } superheated
 $P_1 = 2500 \text{ psia}$ } steam
 $h_1 = 1457.5 \text{ Btu/lbm}$
 $s_1 = 1.15269 \text{ Btu/lbm}^\circ\text{R}$

(2) $P_2 = 200 \text{ psia}$ } saturated
 $s_2 = s_1 = 1.15269 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}} < s_g(200 \text{ psia})$ } steam

$s_{2f} = 0.5438 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}}$
 $s_{2g} = 1.5454 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}}$ } $x_2 = 0.9815$

$h_{2f} = 355.5 \text{ Btu/lbm}$
 $h_{2g} = 1198.3 \text{ Btu/lbm}$ } $h_2 = 1182.7 \frac{\text{Btu}}{\text{lbm}}$

- \dot{m}_2 is drawn off at this point -

(3) $\dot{m}_3 = \dot{m} - \dot{m}_2$ saturated steam
 $P_3 = 1 \text{ psia}$
 $s_3 = s_1 = 1.15269 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}} < s_g(1 \text{ psia})$

$s_{3f} = 0.1326 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}}$
 $s_{3g} = 1.9781 \frac{\text{Btu}}{\text{lbm}^\circ\text{R}}$ } $x_3 = 0.7555$

$h_{3f} = 69.73 \text{ Btu/lbm}$
 $h_{3g} = 1105.83 \text{ Btu/lbm}$ } $h_3 = 852.2 \frac{\text{Btu}}{\text{lbm}}$

(4) saturated liquid
 $h_4 = h_{3f} = 69.73 \text{ Btu/lbm}$
 $v_4 = 0.016136 \text{ ft}^3/\text{lbm}$
 $s_4 = s_{3f} = 0.1326 \text{ Btu/lbm}^\circ\text{R}$
 $P_4 = 1 \text{ psia}$

(5) Subcooled liquid

$$P_5 = 200 \text{ psia}$$

$$N_5 \approx N_4 = 0.016136 \text{ ft}^3/\text{lbm}$$

$$S_5 = S_4$$

$$W_{p_1} = |-_4 W_5| = h_5 - h_4 = N_4 (P_5 - P_4) = 0.59 \frac{\text{Btu}}{\text{lbm}}$$

$$h_5 = h_4 + W_{p_1} = 70.32 \text{ Btu/lbm}$$

(6) Saturated liquid

$$P_6 = 200 \text{ psia}$$

$$h_6 = h_{2f} = 355.5 \text{ Btu/lbm}$$

$$N_6 = 0.01839 \text{ ft}^3/\text{lbm}$$

(7) Subcooled liquid

$$P_7 = 2500 \text{ psia}$$

$$W_{p_2} = |-_6 W_7| = h_7 - h_6 = N_6 (P_7 - P_6) = 7.83 \text{ Btu/lbm}$$

$$h_7 = h_6 + W_{p_2} = 363.33 \text{ Btu/lbm}$$

(8) Saturated liquid

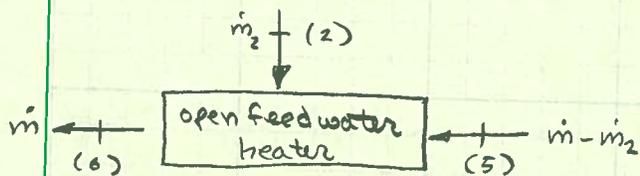
$$P_8 = 2500 \text{ psia}$$

$$h_8 = \underline{1699.92} \text{ Btu/lbm}$$

Summary of States

	<u>h [Btu/lbm]</u>
(1) Superheated Vapor	1457.5
(2) $x = 0.9815$	1182.7
(3) $x = 0.7555$	857.2
(4) Saturated liquid	69.73
(5) Subcooled liquid	70.32
(6) Saturated liquid	355.5
(7) Subcooled liquid	363.33
(8) Saturated liquid	1699.92

what is the mass fraction at (2)?



Applying Cons. of Energy: $\dot{Q} - \dot{W} = \dot{m}_6 h_6 - \dot{m}_5 h_5 - \dot{m}_2 h_2$

- steady
- uniform
- incompressible

$$\dot{m}_6 = \dot{m}$$

$$\dot{m}_5 = \dot{m} - \dot{m}_2$$

$$\begin{aligned} \dot{m} h_6 &= (\dot{m} - \dot{m}_2) h_5 + \dot{m}_2 h_2 \\ &= \dot{m} h_5 + \dot{m}_2 (h_2 - h_5) \end{aligned}$$

$$\dot{m}_2 = \dot{m} \left(\frac{h_6 - h_5}{h_2 - h_5} \right) = \dot{m} \left[\frac{355.5 - 70.32 \text{ Btu/lbm}}{1182.7 - 70.32 \text{ Btu/lbm}} \right] = \underline{\underline{0.2564 \dot{m}}}$$

Turbine work:

$$\dot{W}_t = \dot{m} \left\{ (h_1 - h_2) + \left(1 - \frac{\dot{m}_2}{\dot{m}}\right) (h_2 - h_3) \right\}$$

$$\begin{aligned} \dot{W}_t &= \left(1457.5 \frac{\text{Btu}}{\text{lbm}} - 1182.7 \frac{\text{Btu}}{\text{lbm}} \right) + (1 - 0.2564) \left(1182.7 \frac{\text{Btu}}{\text{lbm}} - 852.2 \frac{\text{Btu}}{\text{lbm}} \right) \\ &= 274.80 \text{ Btu/lbm} + 245.76 \text{ Btu/lbm} \\ &= \underline{\underline{520.56 \text{ Btu/lbm}}} \end{aligned}$$

Pump Work:

$$\dot{W}_p = (\dot{m} - \dot{m}_2) w_5 + \dot{m}_6 w_7 = \dot{m} \left\{ \left(1 - \frac{\dot{m}_2}{\dot{m}}\right) w_5 + w_7 \right\}$$

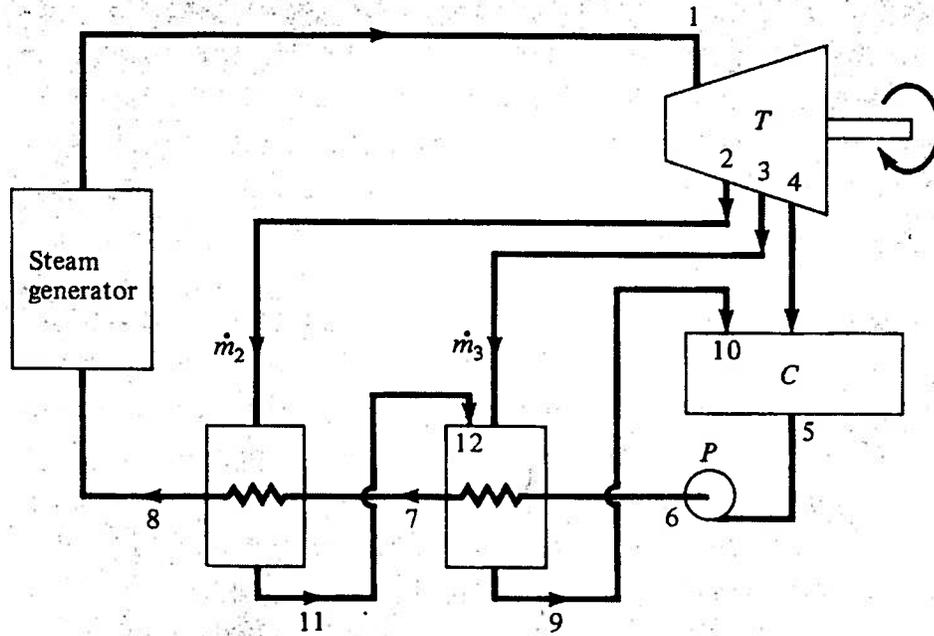
$$\begin{aligned} \dot{W}_p &= (1 - 0.2564) (0.59 \text{ Btu/lbm}) + (7.83 \text{ Btu/lbm}) = \\ &= 0.44 \text{ Btu/lbm} + 7.83 \text{ Btu/lbm} \\ &= \underline{\underline{8.27 \text{ Btu/lbm}}} \end{aligned}$$

Net Work: $\dot{W}_{net} = \dot{W}_t - \dot{W}_p = \underline{\underline{512.29 \text{ Btu/lbm}}}$

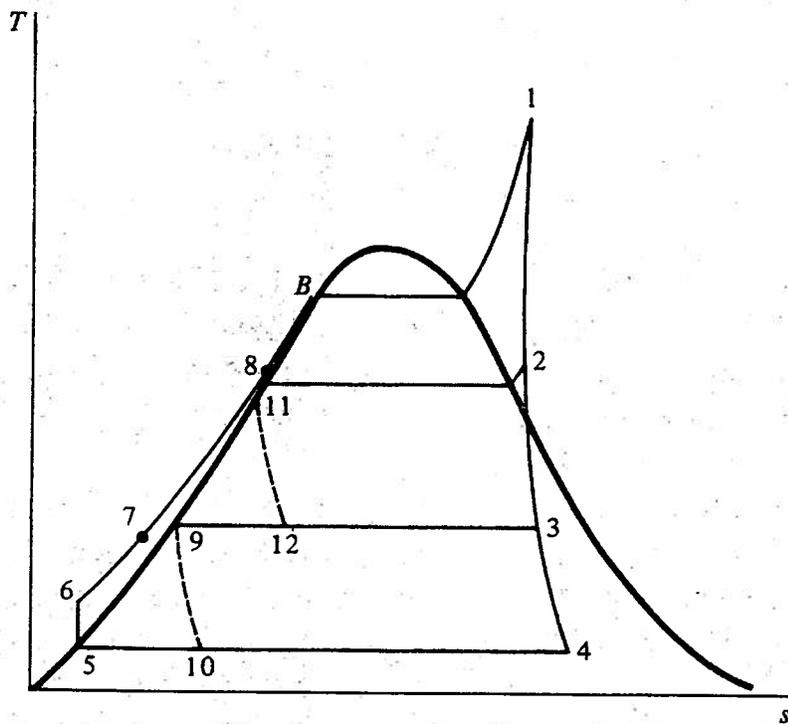
Heat Input: $\dot{q}_H = h_1 - h_7 = \underline{\underline{1094.2 \text{ Btu/lbm}}}$

$$\left. \begin{array}{l} \dot{W}_{net} = 512.29 \text{ Btu/lbm} \\ \dot{q}_H = 1094.2 \text{ Btu/lbm} \end{array} \right\} \underline{\underline{\eta_{th} = 46.8\%}}$$

Closed Feedwater Heater - Cascading Drain Backwards



(a)



(b)

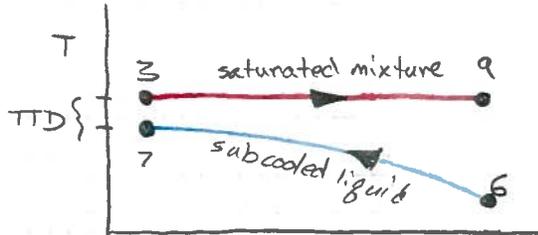
Figure 1: Closed feedwater heaters cascading backwards. From Figure 2-15, page 52, El-Wakil.

Terminal Temperature Difference (TTD or TD)

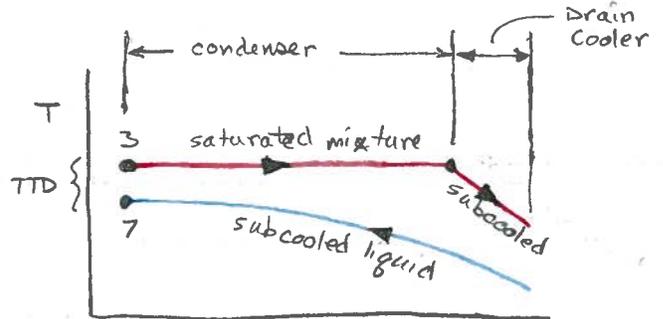
$$TTD = \left\{ \begin{array}{l} \text{saturation temperature} \\ \text{of bled steam} \end{array} \right\} - \left\{ \begin{array}{l} \text{exit temperature} \\ \text{of feed water} \end{array} \right\}$$

- measure of feedwater heater effectiveness

Low Pressure Feedwater Heater

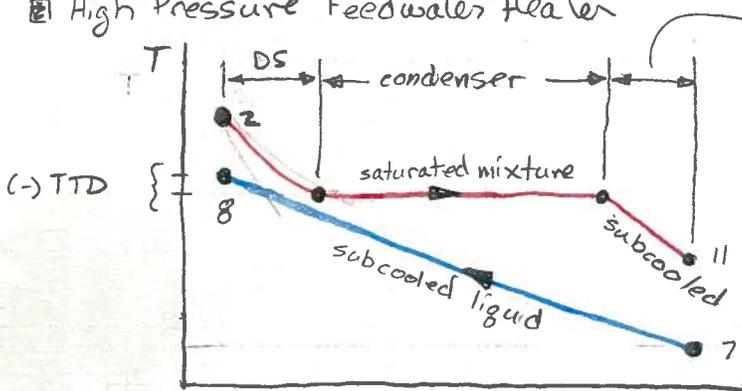


heat exchanger length



heat exchanger length

High Pressure Feedwater Heater

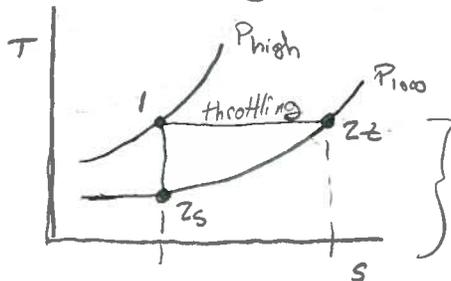


Drain Cooler

DS \equiv desuperheater

The drains must be throttled ("trapped") to lower the pressure back to the pressure of the earlier stage. This throttling results in a loss of availability

- recall for a gas



loss of potential to extract work
availability

- also a loss of availability due to finite temperature difference during heat transfer & external irreversibilities

Closed Feedwater Heater - Pumped Forward

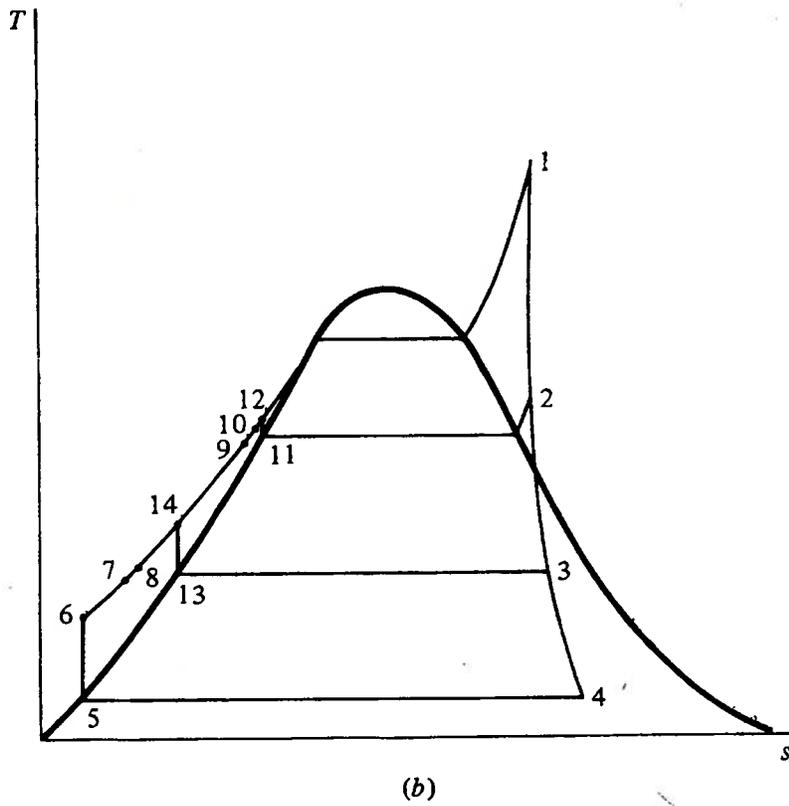
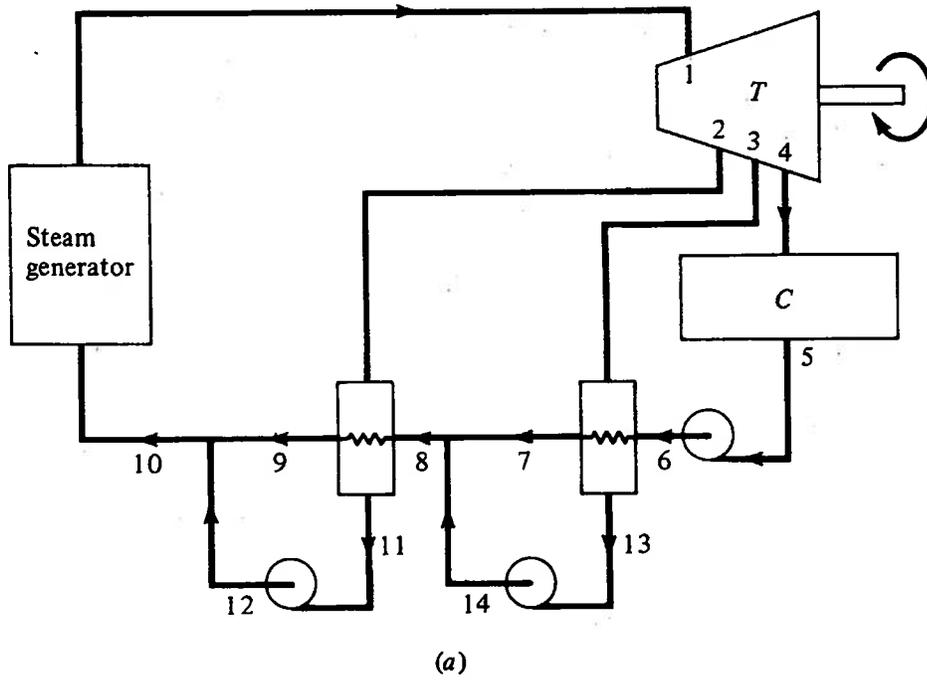
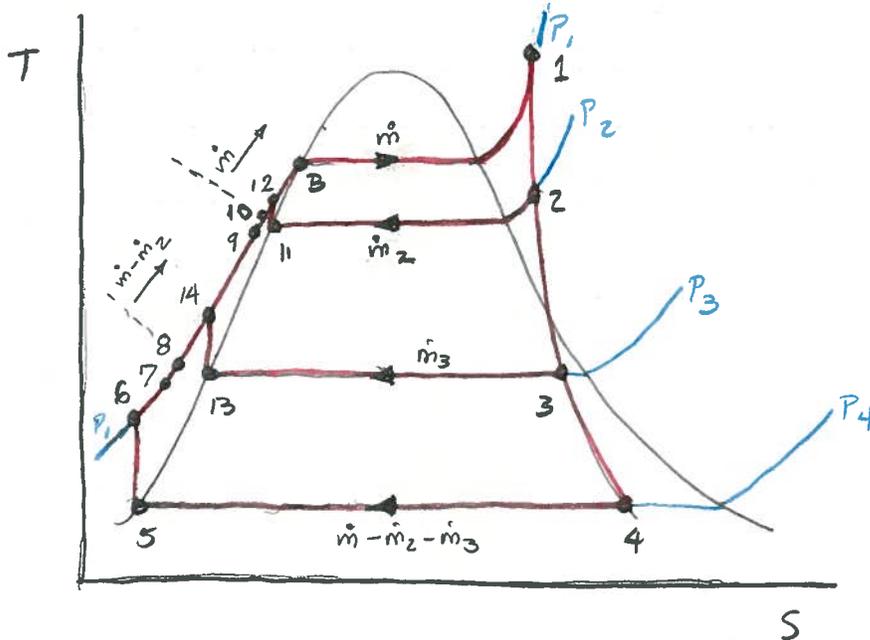
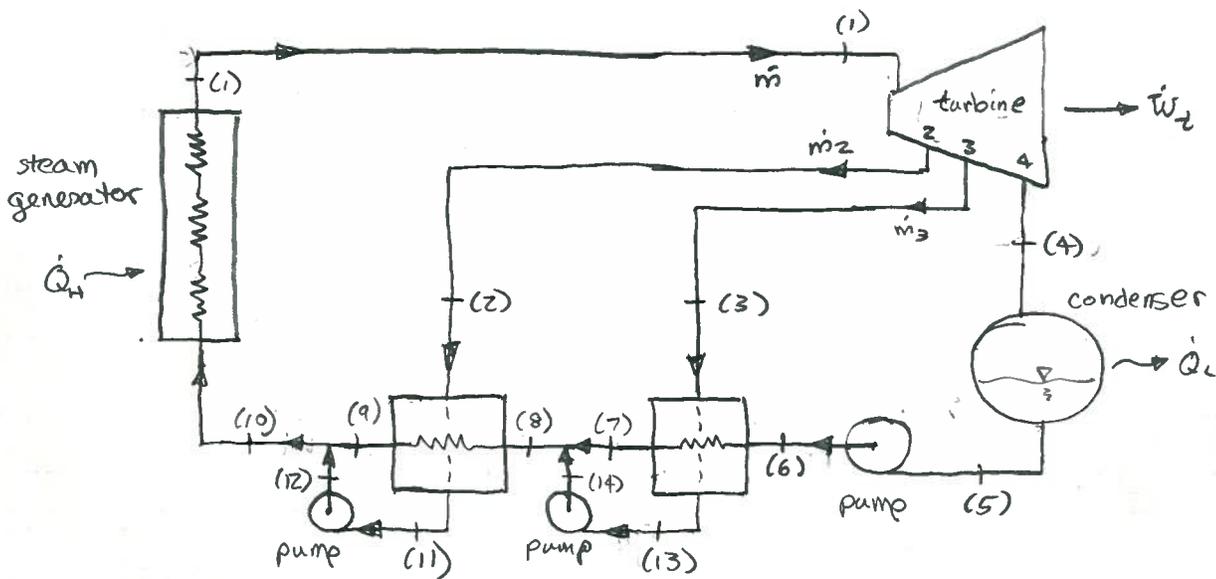


Figure 2: Closed feedwater heaters pumped forward. From Figure 2-18, page 58, El-Wakil.

Closed Feedwater Heater - Drains Pumped Forward

- to reduce loss of availability
- adds some complexity with small pumps
- drains pumped forward into the main feedwater line
 - pumps only handle a fraction of the flow, not all of the flow as with the open feed water heaters
- efficiency is better without drain coolers
- typically used at the lowest feedwater pressure point in order to prevent throttling of a large flow (combined cascaded drains) into the condenser



$$\begin{aligned}
 (1) \quad & P_1 = 15.5 \text{ MPa} \\
 & T_1 = 540^\circ\text{C} \\
 & h_1 = 3446 \text{ kJ/kg} \\
 & s_1 = 6.466 \text{ kJ/kgK}
 \end{aligned}
 \left. \vphantom{\begin{aligned} P_1 \\ T_1 \\ h_1 \\ s_1 \end{aligned}} \right\} \text{superheated vapor}$$

Property Data obtained from EES.

$$\begin{aligned}
 (2) \quad & P_2 = 8.0 \text{ MPa} \\
 & s_{2s} = 6.466 \text{ kJ/kgK} \\
 & h_{2s} = 3209 \text{ kJ/kg}
 \end{aligned}
 \left. \vphantom{\begin{aligned} P_2 \\ s_{2s} \\ h_{2s} \end{aligned}} \right\} \text{isentropic expansion}$$

$$h_2 = h_1 - \eta_t (h_1 - h_{2s}) = 3242 \text{ kJ/kg}$$

$$s_2 = 6.513 \text{ kJ/kgK}$$

$$T_2 = 711.4 \text{ K}$$

actual expansion, superheated vapor

$$\begin{aligned}
 (3) \quad & P_3 = 8.0 \text{ MPa} \\
 & T_3 = 590^\circ\text{C} \\
 & h_3 = 3617 \text{ kJ/kg} \\
 & s_3 = 6.992 \text{ kJ/kgK}
 \end{aligned}
 \left. \vphantom{\begin{aligned} P_3 \\ T_3 \\ h_3 \\ s_3 \end{aligned}} \right\} \text{superheated vapor}$$

$$\begin{aligned}
 (4) \quad & P_4 = 3.4 \text{ MPa} \\
 & T_4 = 462.1^\circ\text{C}
 \end{aligned}
 \left. \vphantom{\begin{aligned} P_4 \\ T_4 \end{aligned}} \right\} \text{superheated vapor}$$

$$h_{4s} = 3318 \text{ kJ/kg for } s = s_3$$

$$h_4 = h_3 - \eta_t (h_3 - h_{4s}) = 3366 \text{ kJ/kg}$$

$$s_4 = 7.058 \text{ kJ/kgK}$$

$$(5) \quad P_5 = 170 \text{ kPa}$$

$$h_{5s} = 2625 \text{ kJ/kg for } s = s_3$$

$$h_5 = h_3 - \eta_t (h_3 - h_{5s}) = 2784 \text{ kJ/kg}$$

$$s_5 = 7.389 \text{ kJ/kgK}$$

$$T_5 = 456.4^\circ\text{C}$$

superheated vapor

$$(6) \quad P_6 = 13 \text{ kPa}$$

$$h_{6s} = 2248 \text{ kJ/kg for } s = s_3$$

$$h_6 = h_3 - \eta_t (h_3 - h_{6s}) = 2467 \text{ kJ/kg}$$

$$\eta_6 = 0.947$$

$$T_6 = 51^\circ\text{C}$$

$$s_6 = 7.668 \text{ kJ/kgK}$$

$\eta < 1$

$$(7) \quad P_7 = 13 \text{ kPa} \leftarrow \text{saturated liquid, } \eta = 0$$

$$h_7 = 213.6 \text{ kJ/kg}$$

$$T_7 = 51^\circ\text{C}$$

$$s_7 = 0.7169 \text{ kJ/kgK}$$

$$v_7 = 0.001013 \text{ m}^3/\text{kg}$$

- (8) $P_8 = 170 \text{ kPa}$ ← subcooled liquid
 $N_8 \approx N_7$, incompressible fluid

$$|W_p|_{\text{ideal}} = h_{g8} - h_7 = N_7 (P_8 - P_7) = 0.159 \text{ kJ/kg}$$

$$|W_p|_{\text{actual}} = \frac{1}{\eta_p} |W_p|_{\text{ideal}} = 0.1893 \text{ kJ/kg} = h_8 - h_7$$

$$h_8 = 213.8 \text{ kJ/kg}$$

- (9) $P_9 = P_8 = 170 \text{ kPa}$ ← saturated liquid

$$T_9 = 115^\circ\text{C}$$

$$h_9 = 483.2 \text{ kJ/kg}$$

$$S_9 = 1.475 \text{ kJ/kg K}$$

$$N_9 = 0.001056 \text{ m}^3/\text{kg}$$

- (10) $P_{10} = 15,500 \text{ kPa}$ ← subcooled liquid

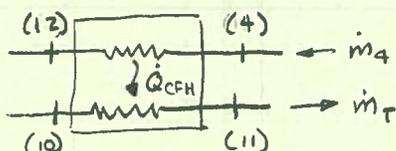
$$N_{10} \approx N_a$$
, incompressible fluid

$$|W_p|_{\text{ideal}} = h_{10s} - h_9 = N_a (P_{10} - P_a) = 19.27 \text{ kJ/kg}$$

$$|W_p|_{\text{actual}} = \frac{1}{\eta_p} |W_p|_{\text{ideal}} = 19.27 \text{ kJ/kg} = h_{10} - h_9$$

$$h_{10} = 502.5 \text{ kJ/kg}$$

Closed Feedwater Heater



$$\dot{Q}_{\text{CFH}} = \dot{m}_4 (h_4 - h_{12}) = \dot{m}_T (h_{11} - h_{10})$$

$$m_4^* = \frac{\dot{m}_4}{\dot{m}_T} = \frac{h_{11} - h_{10}}{h_4 - h_{12}}$$

$$\text{Terminal Temperature Difference: } T_{12} - T_{11} = 3^\circ\text{C}$$

- (11) $P_{11} = 15,500 \text{ kPa}$ ← subcooled liquid

$$T_{11} = T_{12} - 3^\circ\text{C}$$

$$\uparrow T_{\text{sat},12} = 241^\circ\text{C} \quad \left. \vphantom{\begin{matrix} T_{11} = T_{12} - 3^\circ\text{C} \\ \uparrow T_{\text{sat},12} = 241^\circ\text{C} \end{matrix}} \right\} T_{11} = 239^\circ\text{C}$$

$$h_{11} = 1029 \text{ kJ/kg}$$

- (12) $P_{12} = 3400 \text{ kPa}$ } ← saturated liquid

$$T_{12} = 241^\circ\text{C}$$

$$h_{12} = 1042 \text{ kJ/kg}$$

$$S_{12} = 2.71 \text{ kJ/kg K}$$

$$m_4^* = \frac{h_{11} - h_{10}}{h_4 - h_{12}} = 0.2267 \text{ kg/kg}$$

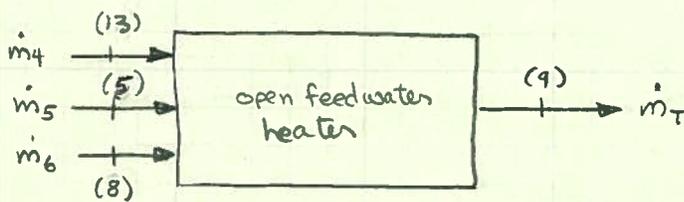
- (13) Throttling Process $h_{13} = h_{12} = 1042 \text{ kJ/kg}$

$$P_{13} = 170 \text{ kPa}$$

$$T_{13} = 115^\circ\text{C}$$

$$x = 0.252$$

Open Feedwater Heater



Conservation of Mass: $\dot{m}_6 = \dot{m}_T - (\dot{m}_4 + \dot{m}_5)$

$$m_6^* = 1 - (m_4^* + m_5^*)$$

Conservation of Energy:

$$\dot{m}_4 h_{13} + \dot{m}_5 h_5 + \dot{m}_6 h_8 = \dot{m}_T h_9$$

$$m_4^* h_{13} + m_5^* h_5 + (1 - m_4^* - m_5^*) h_8 = h_9$$

$$m_4^* (h_{13} - h_8) + m_5^* (h_5 - h_8) = (h_9 - h_8)$$

$$m_5^* = \frac{(h_9 - h_8) - m_4^* (h_{13} - h_8)}{(h_5 - h_8)} = 0.03181 \text{ kg/kg}$$

$$m_6^* = 0.7415 \text{ kg/kg}$$

Turbine Work:

$$\dot{W}_t = \dot{m}_T (h_1 - h_2) + \dot{m}_T (h_3 - h_4) + (\dot{m}_T - \dot{m}_4) (h_4 - h_5) + (\dot{m}_T - \dot{m}_4 - \dot{m}_5) (h_5 - h_6)$$

$$w_t = (h_1 - h_2) + (h_3 - h_4) + (1 - m_4^*) (h_4 - h_5) + (1 - m_4^* - m_5^*) (h_5 - h_6)$$

$$= (170 \text{ kJ/kg}) + (251 \text{ kJ/kg}) + 0.7733 (582 \text{ kJ/kg}) + 0.7415 (317 \text{ kJ/kg})$$

$$= 1106 \text{ kJ/kg}$$

Turbine Power: $\dot{W}_t = 200 \text{ MW} = \dot{m}_T w_t$

$$\dot{m}_T = 180.8 \text{ kg/s}$$

Pump Work: $|w_{p1}| = (\dot{m}_T - \dot{m}_4 - \dot{m}_5) (h_8 - h_7) + \dot{m}_T (h_{10} - h_9)$

$$|w_{p1}| = (1 - m_4^* - m_5^*) (h_8 - h_7) + (h_{10} - h_9) = 19.45 \text{ kJ/kg}$$

Net work: $w_{\text{net}} = w_t - |w_{p1}| = 1090.55 \text{ kJ/kg}$

Heat Input: $\dot{Q}_{\text{in}} = \dot{m}_T (h_1 - h_{11}) + \dot{m}_T (h_3 - h_2)$

$$q_{\text{in}} = (h_1 - h_{11}) + (h_3 - h_2) = 2762 \text{ kJ/kg}$$

Thermal Efficiency: $\eta_{\text{th}} = \frac{w_{\text{net}}}{q_{\text{in}}} = 39.5\%$

Moisture Content at Turbine Exhaust = $1 - x_6 = 5.3\%$

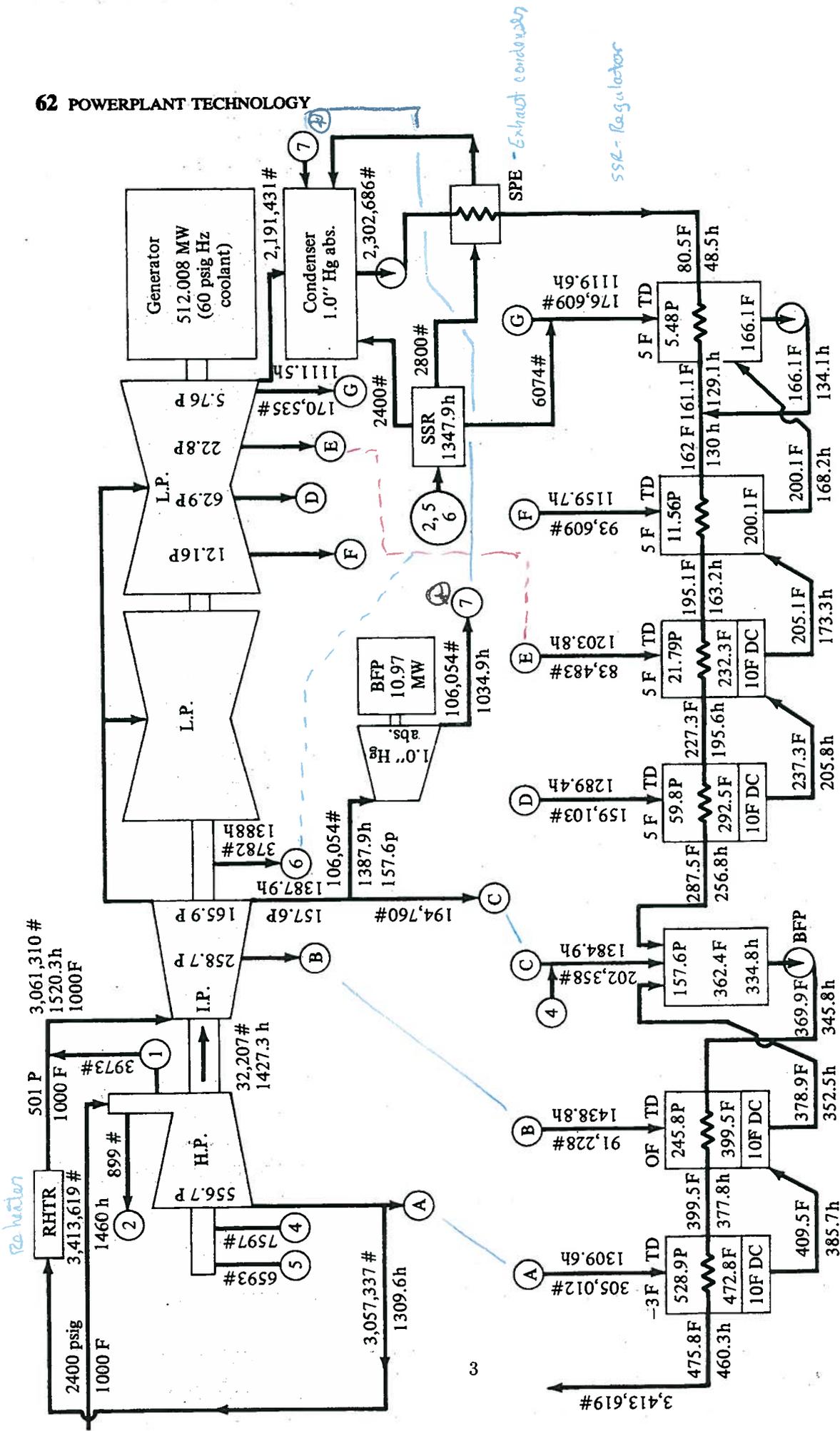
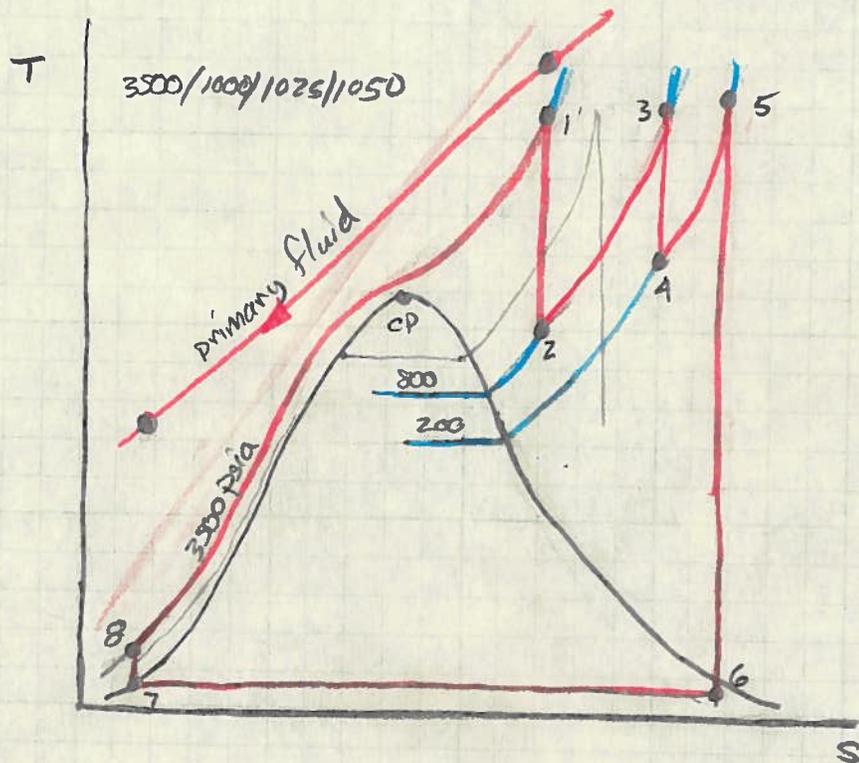


Figure 2-20 Flow diagram of an actual 512-MW 2400 psig/1000°F/1000°F reheat powerplant with seven feedwater heaters. (Courtesy Wisconsin Power & Light Co.)

Supercritical Cycle



- critical point of steam is 3208 psia
- gradual change in temperature and density, but no phase change
- smaller external irreversibilities
 - no abrupt change in temperature in 2-phase region
- supercritical cycle receives more thermal energy at high temperatures than a subcritical cycle with the same turbine inlet temperature
- single stage expansion would result in very wet steam; often use reheat or two reheats
- The temperature can be increased in reheat sections (3500/1000/1025/1050) because the pressures are lower (steam tube materials limit).

Co-Generation

Simultaneous generation of electricity and process steam in a single plant. The process steam may be used as a source of superheated vapor or as a source of heat. Co-generation is commonly used by the chemical industry, paper mills, municipalities for district heating, and in cement kilns.

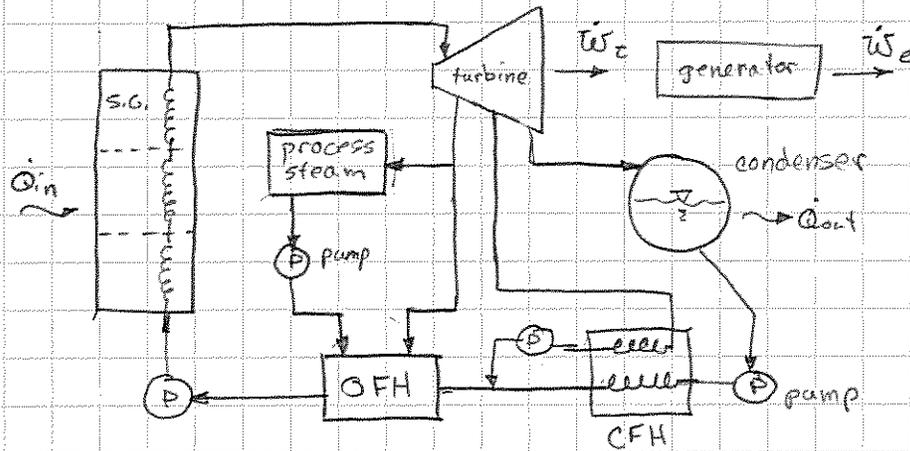


Fig. Co-Generation shown as a topping cycle with extraction and regeneration.

Topping Cycle: High pressure steam is used to generate electrical power, and low pressure steam (0.5 - 40.0 bar) is used for process. The process steam may be extracted from the turbine or from the turbine exhaust. The latter is known as a back pressure turbine since the turbine exhaust is at a higher pressure than the condenser.

Bottoming Cycle: High pressure steam is used for process, and low pressure steam is used to produce electrical power. An example of a bottoming cycle is certain types of cement kilns. Typically, co-generation with a bottoming cycle has very low overall efficiency.

Co-Generation Efficiency

Co-generation is only beneficial if less energy is used as compared to separate generation of electricity and process steam (heat).

$$\text{Powerplant Efficiency, } \eta_{\text{plant}} = \frac{\dot{w}_e}{\Delta H_{\text{fuel}}} = \eta_{\text{sg}} \eta_{\text{th}} \eta_e$$

$$\left\{ \text{Heat Rate, HR} = \frac{\Delta H_{\text{fuel}}}{\dot{w}_e} \right\}$$

$$\text{Co-Generation Plant Efficiency, } \eta_{\text{cogen}} = \frac{\dot{w}_e + \Delta H_{\text{process}}}{\Delta H_{\text{fuel}}}$$

In order to assess whether or not co-generation is an improvement over separate generation, the fraction of fuel used to generate electrical power must be determined. Define the fraction of electrical power output to total power output as:

$$\chi = \frac{\dot{w}_e}{\dot{w}_e + \Delta H_{\text{process}}}$$

and

$$1 - \chi = \frac{\Delta H_{\text{process}}}{\dot{w}_e + \Delta H_{\text{process}}}$$

The reciprocal of the co-generation plant efficiency, which is similar to a Heat Rate, allows for an assessment of efficiency improvements.

$$\frac{1}{\eta_{\text{cogen}}} = \frac{\Delta H_{\text{fuel}}}{\dot{w}_e + \Delta H_{\text{process}}} = \frac{\Delta H_{\text{fuel}}}{\dot{w}_e} \chi + \frac{\Delta H_{\text{fuel}}}{\Delta H_{\text{process}}} (1 - \chi)$$

$$\eta_{\text{cogen}} = \frac{\dot{w}_e + \Delta H_{\text{process}}}{\Delta H_{\text{fuel}}} \quad (1)$$

$$\eta_{\text{sep}} = \frac{1}{\frac{\chi}{\eta_{\text{sg}} \eta_{\text{th}} \eta_e} + \frac{1 - \chi}{\eta_{\text{sg}}}} \quad (2)$$

Co-generation may be beneficial if $\eta_{\text{cogen}} > \eta_{\text{sep}}$

(eq. 1) (eq. 2)