

part 2

**Example 6:** Height of a packed bed absorber

How can we use mass transfer to design a packed bed gas absorber to achieve a desired separation?

**Example 6** is presented as a series of **linked examples** that navigate around apparent “dead ends” in modeling mass-transfer units

Previously:

LECTURES

	Identify a question	Invent something	Try to use it
III	1. How can we model a large, practical device dependent on mass transfer? III	1. Apply the species A mass balance to a macroscopic C.V. III	1. Lack a system to account for A going between phases PAUSE III
LAST TIME	2. How can we account for A going between phases? IV	2. Invent $k_x$ through linear driving force (LDF) model IV	2. Gets A <u>to</u> the boundary, but not <u>across</u> PAUSE V
	3. How can we improve LDF model to cross the boundary (bulk-to-bulk transfer)? VI	3. Write LDF in both phases and combine to create overall effect of multiple resistances VI	3. Working, but can we devise a convenient shorthand? PAUSE VI
	4. Can we model a large, practical device, incorporating $K_L, K_G$ to account for mass xfer between phases? VI	4. Yes VI	

Today: Example 8 + brief into Ex 9,10

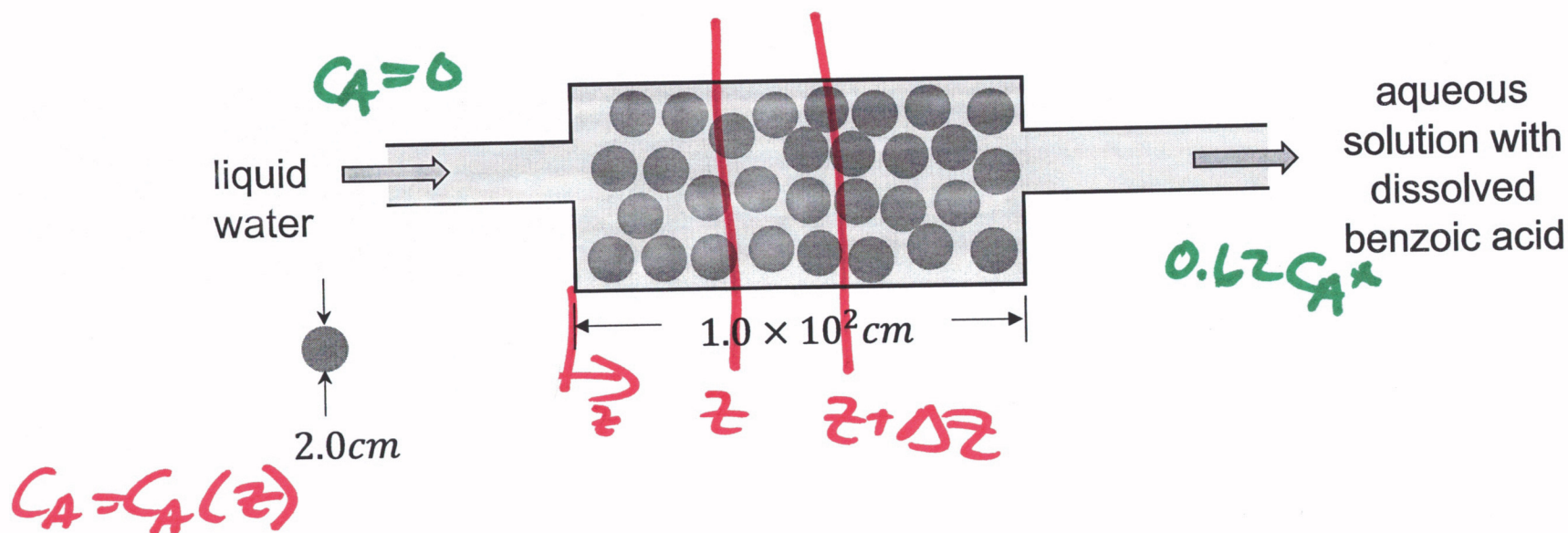
Then, Lec IV return to Ex 6

# LECTURE IV (continued)

2

## Unsteady Macroscopic Species A Mass Balance

**Example 8:** Flow through a packed bed of soluble spherical pellets.



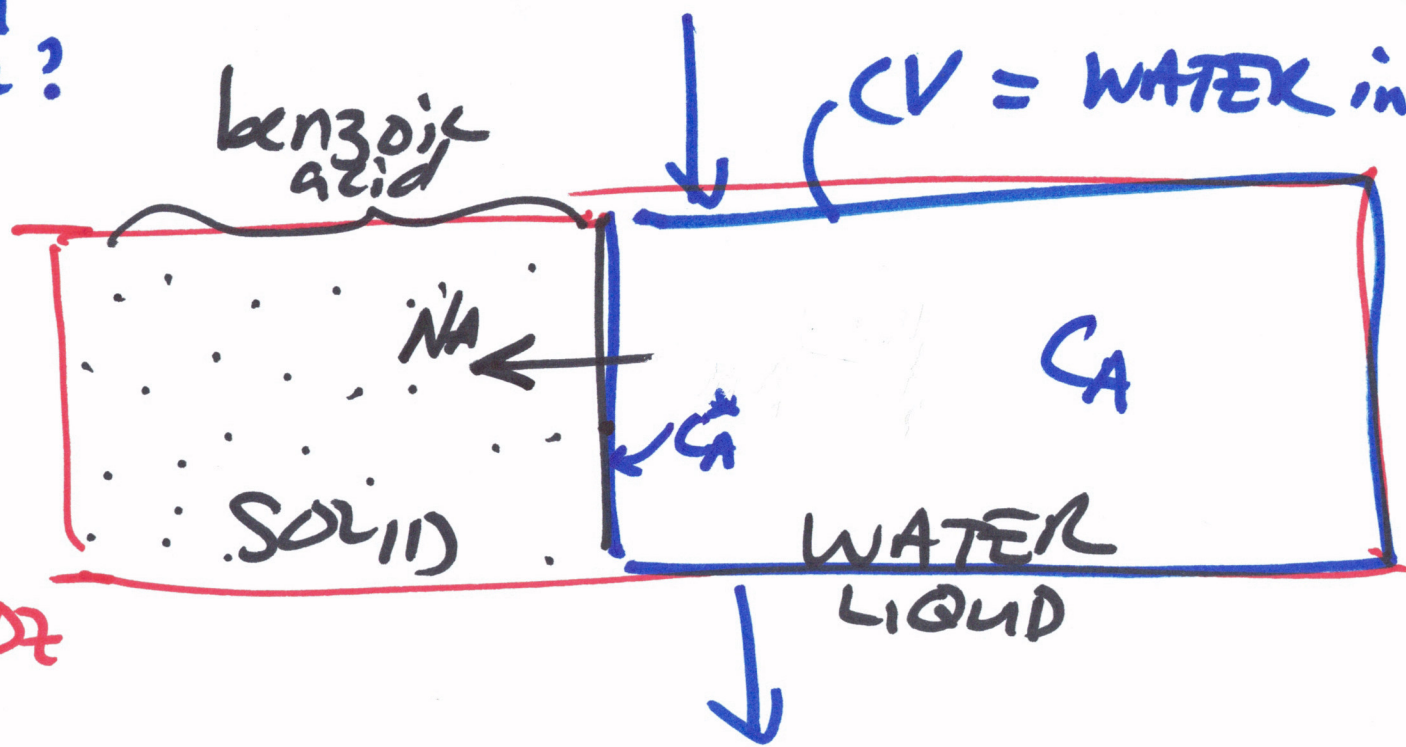
Two-centimeter diameter spheres of benzoic acid (soluble in water) are packed into a bed as shown. The spheres have  $23 \text{ cm}^2$  of surface area per  $\text{cm}^3$  volume of bed. What is the mass transfer coefficient when pure water flowing in ("superficial velocity" =  $5.0 \text{ cm/s}$ ) exits 62% saturated with benzoic acid?

Lets do a  
pseudo steady  
state solution

$$a = 23 \frac{\text{cm}^2}{\text{cm}^3}$$

Control Volume?

$z$   
 $z + \Delta z$



CV = WATER in SLICE  
WORKS

$A_{xs} \Delta z = \text{VOLUME of RED}$

cannot be C.V. since species A mass x fu between phases is not captured

$S = a (A_{xs} \Delta z) = \text{surface area of the red total volume of the slice}$

★  $N_A = \text{flux "out" according to our macro species A mass bal eqn}$

Unsteady Macroscopic Species A Mass Balance

MOLES

3

accumulation = net flow in + production + introduction

$$\frac{d}{dt} (\mathcal{M}_{A,sys}) = -\Delta \dot{\mathcal{M}}_A + R_A V_{sys} - \sum_j (N_A S)_j$$

pseudo steady state

no rxn

no rxn

Species A conc grows over length of the bed

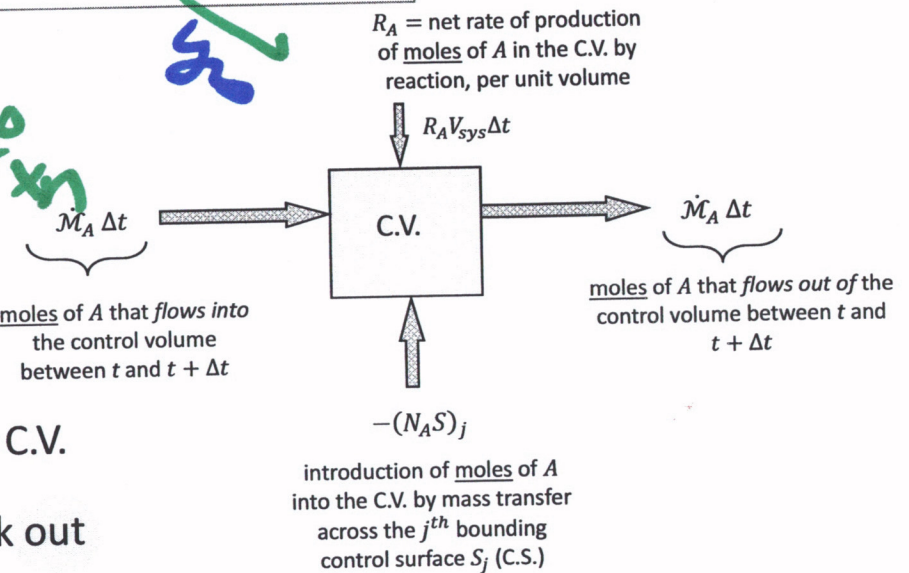
$\mathcal{M}_{A,sys} = c_A V_{sys}$  = total moles of A in the C.V.

$\Delta \dot{\mathcal{M}}_A = \sum_{j,outs} \dot{\mathcal{M}}_{A,j} - \sum_{j,ins} \dot{\mathcal{M}}_{A,j}$  = bulk out

$R_A$  = net rate of production of moles of A in the C.V. by reaction, per unit volume

$V_{sys}$  = system volume

$N_{Aj}$  = molar flux of A out through the  $j^{th}$  C.S.



$S_{sys} = \sum_j S_j$   
 Δ is "out" - "in"  
 C.S. = control surface  
 C.V. = control volume

out of C.V.

# How to characterize the mass xfr between phases?

(3k)

Overall Mass-Transfer Coefficient

concentrated regime

## Liquid-phase-units

Film Linear driving force model:

$$N_A \equiv k_x(x_{A,i} - x_{A,b})$$

$$N_A \equiv k_{cL}(c_{AL,i} - c_{AL,b})$$

## Gas-phase-units:

Film Linear driving force model:

$$N_A \equiv k_p(p_{A,b} - p_{A,i})$$

$$N_A \equiv k_{cG}(c_{AG,b} - c_{A,i})$$

$$N_A \equiv k_y(y_{A,b} - y_{A,i})$$

## Liquid-phase-units

Overall Linear driving force model:

$$N_A \equiv K_x(x_A^*( ) - x_{A,b})$$

$$N_A \equiv K_{cL}(c_{AL}^*( ) - c_{AL,b})$$

( ) =  $p_{A,b}$  or  $c_{A,b}$  or  $y_{A,b}$

## Gas-phase-units:

Overall Linear driving force model:

$$N_A \equiv K_p(p_{A,b} - p_A^*( ))$$

$$N_A \equiv K_{cG}(c_{AG,b} - c_{AG}^*( ))$$

$$N_A \equiv K_y(y_{A,b} - y_A^*( ))$$

( ) =  $x_{A,b}$  or  $c_{AL,b}$

$$K_x = \frac{1}{\frac{1}{k_x} + \frac{1}{m''k_y}}$$

$$K_y = \frac{1}{\frac{1}{k_p} + \frac{1}{m'k_x}}$$

$\tilde{N}_A = \text{flux in}$   
 $N_A = -\tilde{N}_A = \text{flux out}$   
 Let's take these tools out for a spin!

**Warning:**  
 There's even more complexity coming



$\tilde{N}_A = k_c(c_{A,i} - c_{A,b})$   
 $= k_c(c_A^* - c_A)$ , etc.

"out"

(4)

$N_{AS} = ?$

$$-N_A = k_c (C_A^* - C_A)$$

saturated at surface of solid benzoic acid

moles A  
area time  
(out of LIQUID)

mass transfer coef

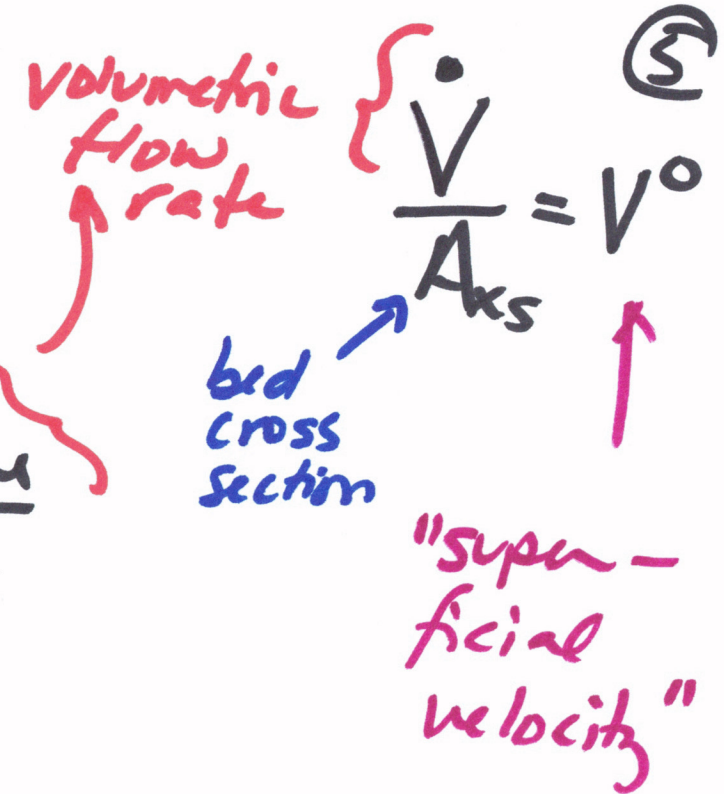
$$N_{AS} = -\underbrace{a A_{xs} \Delta z}_S k_c (C_A^* - C_A)$$

interfacial area between solid and liquid

$\Delta M_A?$   
moles in:

$C_A|_z$   
 $\frac{\text{moles } A}{\text{volume}}$

$V^0 A_{xs}$   
 $\dot{V} \frac{\text{volume}}{\text{time}}$



moles out:

$C_A|_{z+\Delta z} V^0 A_{xs}$

combine + integrate ...

(see HW)

$$0 = - \left( C_A|_{z+\Delta z} - C_A|_z \right) V^0 A_{cs} - N_{AS}$$

"outs" - "ins"

flux out  $\text{\textcircled{6}}$   
 $\downarrow$   
 $N_{AS}$   
 $\frac{a \Delta z}{V^0}$

$$0 = - \left( C_A|_{z+\Delta z} - C_A|_z \right) V^0 + a \Delta z k_c (C_A^* - C_A)$$

$$\lim_{\Delta z \rightarrow 0} \frac{C_A|_{z+\Delta z} - C_A|_z}{\Delta z} = \frac{a k_c}{V^0} (C_A^* - C_A)$$

$$\frac{dC_A}{dz} = \left( \frac{a k_c}{V^0} \right) (C_A^* - C_A)$$

solve w/  $t=0$   
 $C_A=0$  //