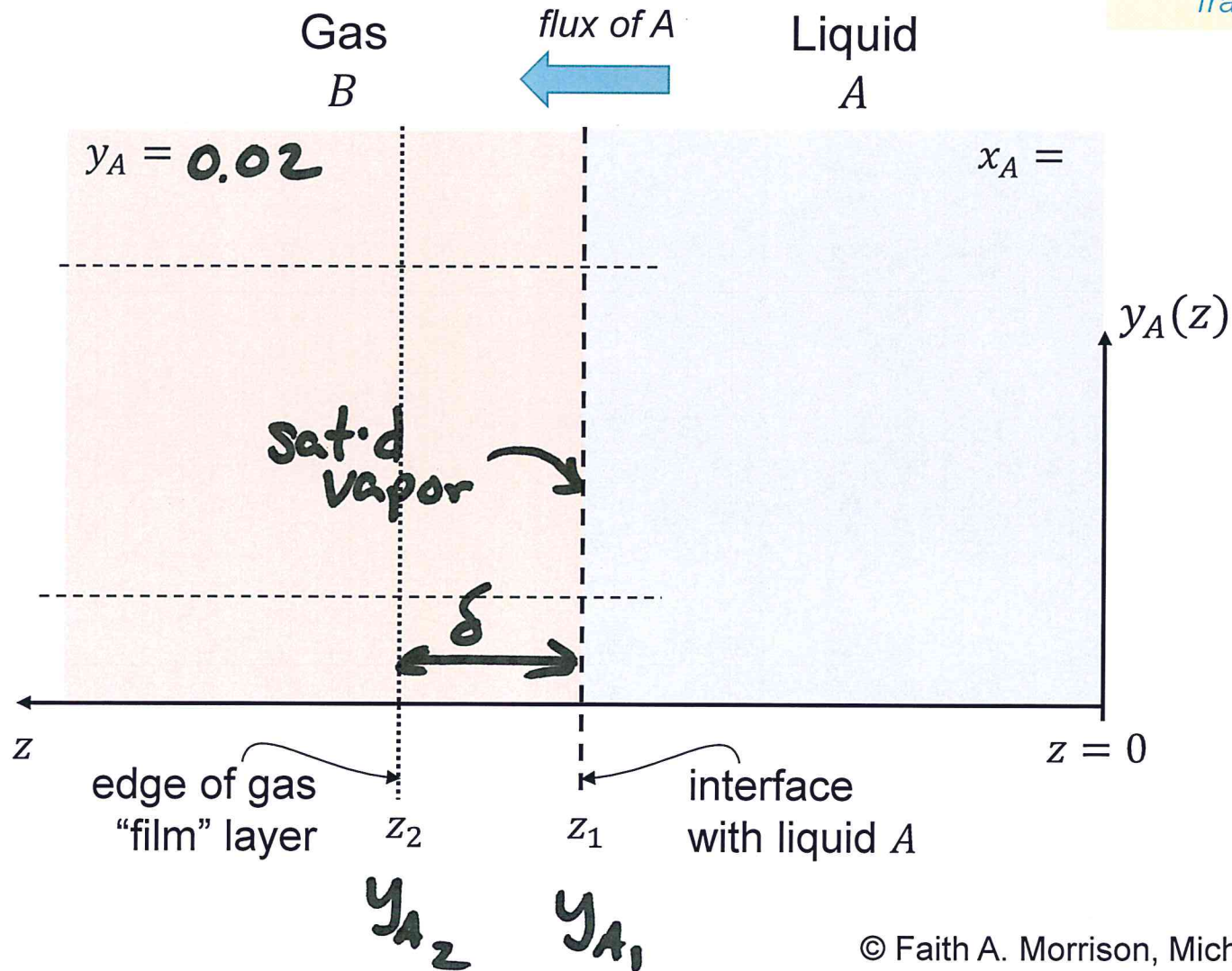


**Example:** The film model of 1D steady diffusion yields a composition distribution and an expression for the flux. What is the mass-transfer coefficient in the film model?

(we use  $x$  for liquid and  $y$  for gas mole fractions)



From Example 1:

3 April 2019 (2)

$$N_{A2} = \frac{C_{DAB}}{(z_2 - z_1)} \ln \left( \frac{1 - y_{A2}}{1 - y_{A1}} \right)$$

↑  
need  $D_{AB}$

Linear driving force model:

$$N_{A2} = k_y (y_{A1} - y_{A2})$$

↑  
need  $k_y$

Both approaches are right. (if their assumptions are met)

If we equate these two expressions, we see how the two approaches are related.

$$N_{A2} = k_y (y_{A1} - y_{A2})$$

$$= \frac{c D_{AB}}{\delta} \ln \left( \frac{1 - y_{A2}}{1 - y_{A1}} \right)$$

Since they are mole fractions

$$\begin{cases} 1 - y_{A1} = y_{B1} \\ 1 - y_{A2} = y_{B2} \end{cases}$$

NOTE

$$y_{A1} - y_{A2} = 1 - y_{B1} - (1 - y_{B2})$$

$$= 1 - y_{B1} - 1 + y_{B2}$$

$$= y_{B2} - y_{B1}$$

Substituting above + solving for  $k_y$ :

(F)

$$k_y = \frac{c D_{AB}}{\delta} \left( \frac{\ln \frac{y_{B2}}{y_{B1}}}{y_{B2} - y_{B1}} \right)$$

$\underbrace{\hspace{15em}}_{1/y_{B,lm}}$

a "log mean" concentration driving force

$$k_y = \left( \frac{c D_{AB}}{\delta y_{B,lm}} \right)$$

- $k_y \propto D_{AB}$
- we don't know  $\delta$ , however