

1/29/2020 ①
Cm3120

Example:

When will my pipes freeze?

The temperature has been 35°F for a while now, sufficient to chill the ground to this temperature for many tens of feet below the surface. Suddenly the temperature drops to -20°F . How long will it take for freezing temperatures (32°F) to reach my pipes, which are 8 ft under ground?



6:21 AM	Houghton	-5°
7:21 AM	New Effington	-29°
7:21 AM	Minneapolis	-27°
7:21 AM	Madison	-26°
7:21 AM	Iron Mountain	-16°
7:21 AM	Ironwood	-18°
8:21 AM	Copper Harbor	-4°

1/30/19



At first...

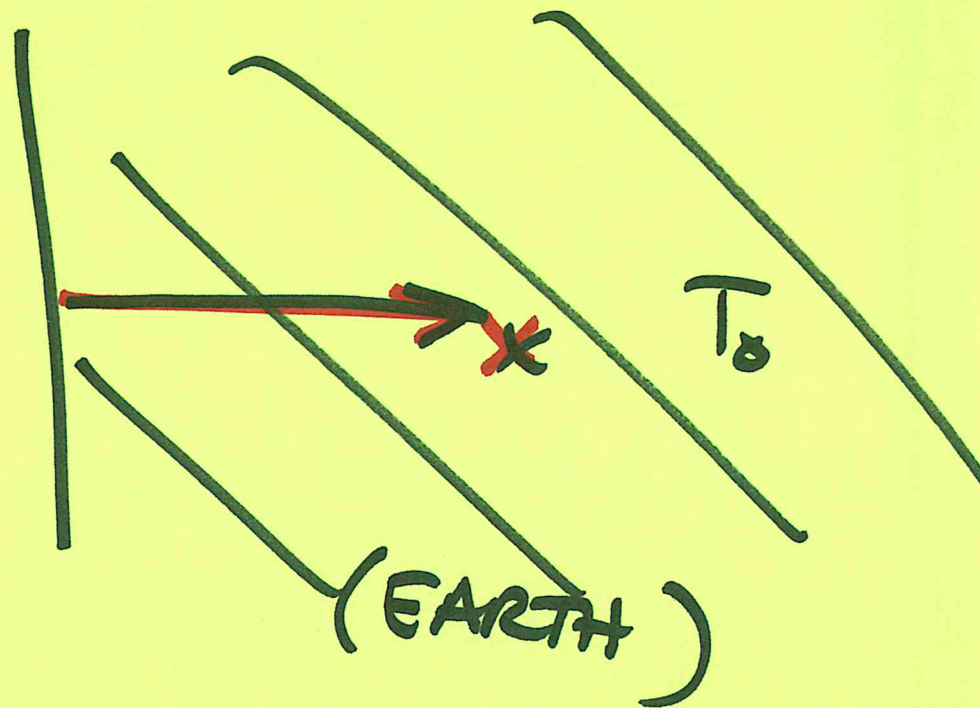
(BUILDING THE MODEL)

(2)

$$t < 0$$

$$\forall x$$

$$T = T_0 = 35^\circ\text{F}$$

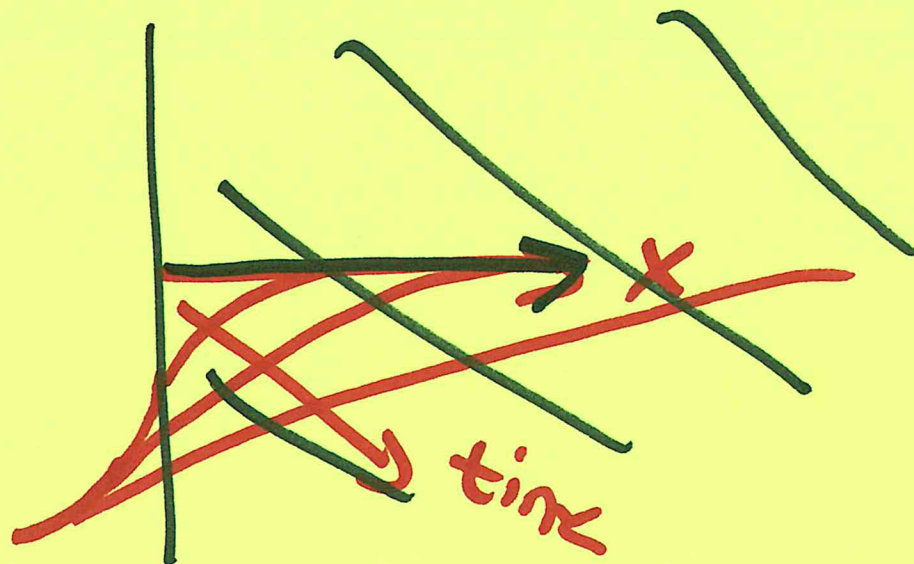


Then suddenly...

$$t \geq 0 \quad x = 0$$

$$-k \frac{dT}{dx} \Big|_{x=0} = \frac{q_x}{A} \Big|_{x=0} = h(T_b - T) \Big|_{x=0}$$

(< 0)
Newton's Law of Cooling T_b





The Equation of Energy for systems with constant k

Microscopic energy balance, constant thermal conductivity; Gibbs notation

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \mathbf{v} \cdot \nabla T \right) = k \nabla^2 T + S_e$$

Microscopic energy balance, constant thermal conductivity; Cartesian coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S_e$$

Microscopic energy balance, constant thermal conductivity; cylindrical coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right) + S_e$$

Microscopic energy balance, constant thermal conductivity; spherical coordinates

$$\begin{aligned} \rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) \\ = k \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 T}{\partial \phi^2} \right) + S_e \end{aligned}$$

$$\rho \hat{c}_p \frac{\partial T}{\partial t} = k \frac{\partial^2 T}{\partial x^2}$$

$$\frac{\partial T}{\partial t} = \frac{k}{\rho \hat{c}_p} \frac{\partial^2 T}{\partial x^2}$$

α thermal diffusivity

Reference: F. A. Morrison, "Web Appendix to An Introduction to Fluid Mechanics," Cambridge University Press, New York, 2013 (pages.mtu.edu/~fmorriso/cm310/IFMWWebAppendixDMicroBalanceMorrison.pdf). This worksheet is on the web at pages.mtu.edu/~fmorriso/cm310/energy.pdf.

④

$$\left| \frac{q_{bx}}{A} \right| = h \left| T_b - T_w \right|$$

$\frac{q_{bx}}{A}$ is negative

arrange
the temps
to get a
negative #

"initial" (one)

~~$t=0$~~

$t < 0$

$\forall x$

Need:

① initial \neq
② Boundary conditions
Two

$T(t, x)$ (3)

$$T = T_0 = 35^\circ\text{F}$$

"boundary" (two)

1) $t \geq 0$

$x=0$

$$-k \left. \frac{dT}{dx} \right|_{x=0} = h (T_b - T) \Big|_{x=0}$$

2) $\forall t$

$x=L$

$$T = T_0$$

guess ∞

Ⓞ

we know this

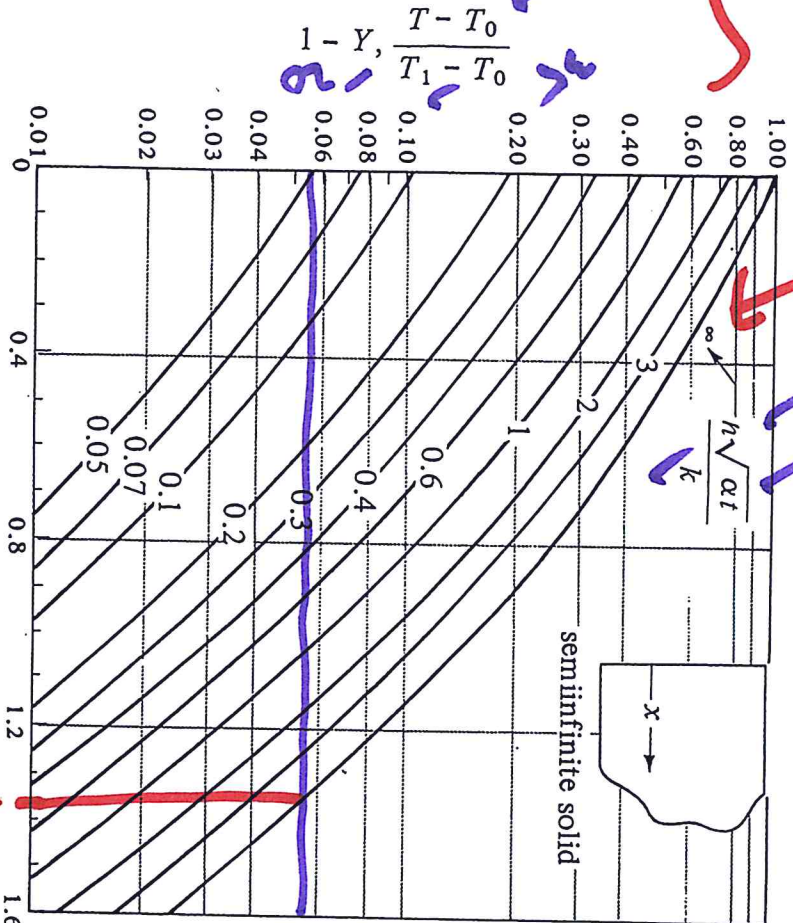


FIGURE 5.3-3. Unsteady-state heat conduction in a semi-infinite solid with surface convection. Calculated from Eq. (5.3-7)(SI).

$1 - Y = 0.05$
 32%
 35%
 35%

PURPOSE:
 Learn to use solns
 from the literature - quick.
 Grankep 15, 4th edition

see if
 this
 value
 gives
 the
 same
 T