

CM3120 Transport/Unit Operations 2

STEADY Heat Transfer—Review



Professor Faith A. Morrison

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www.chem.mtu.edu/~fmorriso/cm3120/cm3120.html

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Why study transport/unit ops?

“... detailed, analytical understanding of how physics/nature works ...”

How do we apply this understanding?

- Organize with dimensional analysis
- Make operational by performing experiments leading to data correlations.

Why study transport/unit ops? Michigan Tech

• Modern engineering systems are complex and often cannot be operated and maintained without analytical understanding

• Design of new systems will come from high-tech innovation, which can only come from detailed, analytical understanding of how physics/nature works



Image: wikipedia.org



Image: planetforward.ca

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Why study transport/unit ops?

“... detailed, analytical understanding of how physics and nature works ...”

Topics:

How nature works:

- **Mass** is conserved
- **Momentum** is conserved
- **Energy** is conserved

How engineering operations work:

- Transform one type of **mass mixture** to others
- Often in a **flow** environment
- Often with the input or output of **energy**
- **Energy** flows are often transient (unsteady)

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Where are we?

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Why study transport/unit ops?

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Topics:

How nature works:

- Mass is conserved
- Momentum is conserved
- Energy

How engineering:

- Trans
- Offer
- Offer
- Energy flows are often transient (unsteady)

Where are we?

How far did we get in CM3110 and other prerequisite courses?

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Why study transport/unit ops?

Where are we?

How far did we get on in CM3110 and other prerequisite courses?


Michigan Tech

CM3110
Transport Processes and Unit Operations I

Part 2:

Professor Faith Morrison
Department of Chemical Engineering
Michigan Technological University

CM3110 - Momentum and Heat Transport
CM3120 - Heat and Mass Transport



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CM3110: Heat Transfer

**CM3110
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CM3110
Transport Processes and Unit Operations I
Part 2:
Professor Faith Morrison
Department of Chemical Engineering
Michigan Technological University
CM3110 - Momentum and Heat Transport
CM3120 - Heat and Mass Transport

Summary

Within homogeneous phases:

- Microscopic Energy Balances
- 1D **Steady** solutions

rectangular: $\frac{q_x}{A} = C_1$
 $T = ax + b$

cylindrical: $\frac{q_r}{A} = \frac{C_1}{r}$
 $T = a \ln x + b$

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S_e$$

- Temperature and *Newton's law of cooling* boundary conditions (if *h* is supplied; or obtain from literature correlation)

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CM3110: Heat Transfer

**CM3110
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CM3110
Transport Processes and Unit Operations I
Part 2:
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Department of Chemical Engineering
Michigan Technological University
CM3110 - Momentum and Heat Transport
CM3120 - Heat and Mass Transport

Summary

Across phase boundaries:

- Newton's law of cooling $\frac{q_x}{A} = h |T_{bulk} - T_{wall}|$
- Microscopic Energy, Momentum, and Mass Balances

Micro momentum:

Micro energy:

$$\rho \left(\frac{\partial \underline{v}}{\partial t} + \underline{v} \cdot \nabla \underline{v} \right) = -\nabla p + \mu \nabla^2 \underline{v} + \rho \underline{g}$$

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S_e$$

- Simultaneous effects (complex)
- Solutions are difficult to obtain (and often not really necessary)
→ use **dimensional analysis** to obtain **h**
- h* Data correlations for:
 - ✓ forced convection (Sieder-Tate)
 - ✓ natural convection
 - ✓ evaporation/condensation
 - ✓ radiation

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CM3110: Heat Transfer

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 Transport Processes and Unit Operations I
 Part 2:
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 CM3110 - Momentum, Heat Transfer
 CM3120 - Heat and Mass Transfer

Summary

Applied Heat Transfer (including Unit Operations)

- Macroscopic energy balances
- Heat Exchangers
 - ✓ double pipe (ΔT_{lm})
 - ✓ Shell-and-tube ($F_T \Delta T_{lm}$)
 - ✓ Heat exchanger effectiveness (NTU, $Q = \epsilon(mC_p)_{min}(T_{hi} - T_{ci})$)
- Evaporators/ Condensers
- Ovens (radiation and convection)
- Heat Shields

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Where are we?

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Topics:

How nature works:

- ✓ Mass is conserved
- ✓ Momentum is conserved
- ✓ Energy is conserved

How engineering operations work:

- Transform one type of mass mixture to others
- ✓ Often in a fluid environment
- ✓ Often with the input or output of energy
- Energy flows are often **transient (unsteady)**


We are here

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Unsteady State Heat Transfer

CM3120 Transport/Unit Operations 2

Unsteady State Heat Transfer



We begin with a Review:

- Microscopic energy balance
- Fourier’s law of heat conduction (k , homogeneous)
- Newton’s law of cooling (h , at the boundary)
- Resistances due to k and h
- Solving for the *steady* temperature field $T(x, y, z)$

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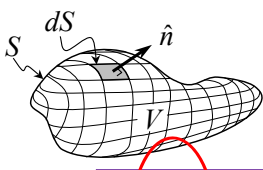
Microscopic Energy Balance Review
CM3110
REVIEW

The **microscopic** energy balance is an expression of the law of conservation of energy.

It includes consideration of **unsteady** energy flows.

Microscopic Energy Balance:

Equation of Thermal Energy



Microscopic **energy** balance written on an arbitrarily shaped volume, V, enclosed by a surface, S

Gibbs notation: $\rho \left(\frac{\partial \hat{E}}{\partial t} + \mathbf{v} \cdot \nabla \hat{E} \right) = -\nabla \cdot \bar{q} + S_e$ general conduction

Gibbs notation: $\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \mathbf{v} \cdot \nabla T \right) = k \nabla^2 T + S_e$ Only Fourier conduction

(incompressible fluid, constant pressure, neglect \hat{E}_k, \hat{E}_p , viscous dissipation)

http://pages.mtu.edu/~fmorriso/cm310/energy_equation.html
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Microscopic Energy Balance Review
CM3110
REVIEW

What **physics** determines how rapidly (the *rate*) the heat transfers from the outside stream to the inside stream?

Energy Transport law

Fourier's Law of Heat Conduction

$$\frac{q_x}{A} = -k \frac{dT}{dx}$$

(for a homogeneous phase)

$\frac{q_x}{A}$ – heat flux=energy/area time)
 k – thermal conductivity
 $\frac{dT}{dx}$ –temperature gradient

(the driving physics of Fourier's law is *Brownian motion*: energy transports down ∇T due to Brownian motion)

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Microscopic Energy Balance Review CM3110 REVIEW

Heat Transfer Rate law:

Fourier's law of Heat Conduction

makes reference to a coordinate system

$$\frac{q_x}{A} = -k \frac{dT}{dx}$$

Allows you to solve for temperature profiles (also known as temperature distributions or fields)

Gibbs notation: $\frac{q}{A} = -k \nabla T$

Fourier's law in three dimensions

$$\tilde{q} = \frac{q}{A} = \begin{pmatrix} -k \frac{\partial T}{\partial x} \\ -k \frac{\partial T}{\partial y} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}_{xyz}$$

- Heat flows **down** a temperature gradient
- Flux is proportional to the magnitude of temperature gradient

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Microscopic Energy Balance Review CM3110 REVIEW

Equation of Energy
(microscopic energy balance)

$$\underbrace{\rho \hat{C}_p}_{\text{rate of change}} \left(\underbrace{\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T}_{\text{convection}} \right) = \underbrace{k \nabla^2 T}_{\text{conduction (all directions)}} + \underbrace{S_e}_{\text{source (energy generated per unit volume per time)}}$$

velocity must satisfy equation of motion, equation of continuity

see handout for component notation

http://pages.mtu.edu/~fmorriso/cm310/energy_equation.html © Faith A. Morrison, Michigan Tech U.

The Equation of Energy in Cartesian, cylindrical, and spherical coordinates for Newtonian fluids of constant density, with source term S . Source could be electrical energy due to current flow, chemical energy, etc. Two cases are presented: the general case where thermal conductivity may be a function of temperature (vector flux $\underline{\tilde{q}} = \underline{g}/A$ appears in the equations); and the more usual case, where thermal conductivity is constant.

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Microscopic energy balance, in terms of flux; Gibbs notation

$$\rho c_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = -\nabla \cdot \underline{\tilde{q}} + S$$

Microscopic energy balance, in terms of flux; Cartesian coordinates

$$\rho c_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = - \left(\frac{\partial \tilde{q}_x}{\partial x} + \frac{\partial \tilde{q}_y}{\partial y} + \frac{\partial \tilde{q}_z}{\partial z} \right) + S$$

Microscopic energy balance, in terms of flux; cylindrical coordinates

$$\rho c_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = - \left(\frac{1}{r} \frac{\partial (r \tilde{q}_r)}{\partial r} + \frac{1}{r} \frac{\partial \tilde{q}_\theta}{\partial \theta} + \frac{\partial \tilde{q}_z}{\partial z} \right) + S$$

Microscopic energy balance, in terms of flux; spherical coordinates

$$\rho c_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) = - \left(\frac{1}{r^2} \frac{\partial (r^2 \tilde{q}_r)}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial (\tilde{q}_\theta \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial \tilde{q}_\phi}{\partial \phi} \right) + S$$

Fourier's law of heat conduction, Gibbs notation: $\underline{\tilde{q}} = -k \nabla T$

Fourier's law of heat conduction, Cartesian coordinates: $\begin{pmatrix} \tilde{q}_x \\ \tilde{q}_y \\ \tilde{q}_z \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial x} \\ -k \frac{\partial T}{\partial y} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}$

Fourier's law of heat conduction, cylindrical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_z \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}$

Fourier's law of heat conduction, spherical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_\phi \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -\frac{k}{r \sin \theta} \frac{\partial T}{\partial \phi} \end{pmatrix}$

Reference: F. A. Morrison, "Web Appendix to An Introduction to Fluid Mechanics," Cambridge University Press, New York, 2013. On the web at www.chem.mtu.edu/~fmorriso/IFM_WebAppendixCD2013.pdf

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http://pages.mtu.edu/~fmorriso/cm310/energy_equation.html

The Equation of Energy in Cartesian, cylindrical, and spherical coordinates for Newtonian fluids of constant density, with source term S . Source could be electrical energy due to current flow, chemical energy, etc. Two cases are presented: the general case where thermal conductivity may be a function of temperature (vector flux $\underline{\tilde{q}} = \underline{g}/A$ appears in the equations); and the more usual case, where thermal conductivity is constant.

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Microscopic energy balance, in terms of flux; Gibbs notation

$$\rho c_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = -\nabla \cdot \underline{\tilde{q}} + S$$

Microscopic energy balance, in terms of flux; Cartesian coordinates

$$\rho c_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = - \left(\frac{\partial \tilde{q}_x}{\partial x} + \frac{\partial \tilde{q}_y}{\partial y} + \frac{\partial \tilde{q}_z}{\partial z} \right) + S$$

Microscopic energy balance, in terms of flux; cylindrical coordinates

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Microscopic energy balance, in terms of flux; spherical coordinates

$$\rho c_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) = - \left(\frac{1}{r^2} \frac{\partial (r^2 \tilde{q}_r)}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial (\tilde{q}_\theta \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial \tilde{q}_\phi}{\partial \phi} \right) + S$$

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Fourier's law of heat conduction, Cartesian coordinates: $\begin{pmatrix} \tilde{q}_x \\ \tilde{q}_y \\ \tilde{q}_z \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial x} \\ -k \frac{\partial T}{\partial y} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}$

Fourier's law of heat conduction, cylindrical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_z \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}$

Fourier's law of heat conduction, spherical coordinates: $\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_\phi \end{pmatrix} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -\frac{k}{r \sin \theta} \frac{\partial T}{\partial \phi} \end{pmatrix}$

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Front side:

- Micro E-balance in terms of flux $\underline{\tilde{q}} \equiv \frac{\underline{q}}{A}$
- Fourier's law, $\underline{\tilde{q}} = -k \nabla T$

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Microscopic energy balance, in terms of flux; Gibbs notation

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = -\nabla \cdot \tilde{q} + S$$

Microscopic energy balance, in terms of flux; Cartesian coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = - \left(\frac{\partial \tilde{q}_x}{\partial x} + \frac{\partial \tilde{q}_y}{\partial y} + \frac{\partial \tilde{q}_z}{\partial z} \right) + S$$

Microscopic energy balance, constant thermal conductivity; Cartesian coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

Microscopic energy balance, constant thermal conductivity; cylindrical coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

Microscopic energy balance, constant thermal conductivity; spherical coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) = k \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 T}{\partial \phi^2} \right) + S$$

The Equation of Energy for systems with **constant k**

Microscopic energy balance, constant thermal conductivity; Gibbs notation

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S$$

Microscopic energy balance, constant thermal conductivity; Cartesian coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

Microscopic energy balance, constant thermal conductivity; cylindrical coordinates

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Front side:

- Micro E-balance in terms of flux $\tilde{q} \equiv \frac{q}{A}$
- Fourier's law, $\tilde{q} = -k \nabla T$

Back side:

- Micro E-balance in terms of temperature (Fourier's law incorporated)

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Microscopic Energy Balance

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The Equation of Energy for systems with **constant k**

Microscopic energy balance, constant thermal conductivity; Gibbs notation

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S$$

Microscopic energy balance, constant thermal conductivity; Cartesian coordinates

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

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$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) = k \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 T}{\partial \phi^2} \right) + S$$

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Fourier's Law of Heat Conduction

Fourier's law of heat conduction, Gibbs notation: $\tilde{q} = \underline{q}/A = -k\nabla T$

Fourier's law of heat conduction, Cartesian coordinates: (constant thermal conductivity k)

$$\begin{pmatrix} \tilde{q}_x \\ \tilde{q}_y \\ \tilde{q}_z \end{pmatrix}_{xyz} = \begin{pmatrix} q_x/A \\ q_y/A \\ q_z/A \end{pmatrix}_{xyz} = \begin{pmatrix} -k \frac{\partial T}{\partial x} \\ -k \frac{\partial T}{\partial y} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}_{xyz}$$

$\tilde{q}_x = \frac{q_x}{A}$

Fourier's law of heat conduction, cylindrical coordinates: (constant thermal conductivity k)

$$\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_z \end{pmatrix}_{r\theta z} = \begin{pmatrix} q_r/A \\ q_\theta/A \\ q_z/A \end{pmatrix}_{r\theta z} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -k \frac{\partial T}{\partial z} \end{pmatrix}_{r\theta z}$$

Fourier's law of heat conduction, spherical coordinates: (constant thermal conductivity k)

$$\begin{pmatrix} \tilde{q}_r \\ \tilde{q}_\theta \\ \tilde{q}_\phi \end{pmatrix}_{r\theta\phi} = \begin{pmatrix} q_r/A \\ q_\theta/A \\ q_\phi/A \end{pmatrix}_{r\theta\phi} = \begin{pmatrix} -k \frac{\partial T}{\partial r} \\ -\frac{k}{r} \frac{\partial T}{\partial \theta} \\ -\frac{k}{r \sin \theta} \frac{\partial T}{\partial \phi} \end{pmatrix}_{r\theta\phi}$$

<https://pages.mtu.edu/~fmorriso/cm310/energy.pdf>

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Microscopic Energy Balance Review

Unsteady State Heat Transfer

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Unsteady State Heat Transfer



Now, *Boundary Conditions and Resistances*

We begin with a Review:

- Microscopic energy balance
- Fourier's law of heat conduction (k , homogeneous)
- Newton's law of cooling (h , at the boundary)
- Resistances due to k and h
- Solving for the steady temperature field $T(x, y, z)$

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Microscopic Energy Balance Review—Boundary Conditions

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We will need **boundary conditions** on temperature to solve the microscopic balances for the temperature distribution.

Example 1: Heat flux in a rectangular solid – Temperature BC

What is the steady state temperature profile in a rectangular slab if one side is held at T_1 and the other side is held at T_2 ?

Assumptions:

- wide, tall slab
- steady state

$\frac{q_x}{A}$

HOT SIDE

$T_1 > T_2$
COLD SIDE

We may know the temperature at the boundary.

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Microscopic Energy Balance Review—Boundary Conditions

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We will need **boundary conditions** on temperature to solve the microscopic balances for the temperature distribution.

Example 1: Heat flux in a rectangular solid – Temperature BC

What is the steady state temperature profile in a rectangular slab if one side is held at T_1 and the other side is held at T_2 ?

Assumptions:

- wide, tall slab
- steady state

What if we don't know the wall temperature?

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Microscopic Energy Balance Review—Boundary Conditions

CM3110 REVIEW

The interface between the solid and the fluid calls for a new type of **boundary condition**, *Newton's Law of Cooling*.

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There may exist a resistance to heat transfer due to fluid at the boundary

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Microscopic Energy Balance Review—Boundary Conditions

CM3110 REVIEW

The interface between the solid and the fluid calls for a new type of **boundary condition**, *Newton's Law of Cooling*.

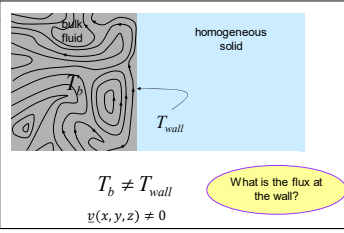
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The temperature difference at the fluid-wall interface is caused by complex phenomena that are lumped together into the heat transfer coefficient, h

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Microscopic Energy Balance Review

The flux at the wall is given by the empirical expression known as **Newton's Law of Cooling** (a linear driving-force model)



$T_b \neq T_{wall}$
 $v(x,y,z) \neq 0$

What is the flux at the wall?

This expression serves as the definition of the **heat transfer coefficient**.

$$\left| \frac{q_x}{A} \right| = h |T_{bulk} - T_{wall}|$$

h depends on:

- geometry
- fluid velocity field
- fluid properties
- temperature difference

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Microscopic Energy Balance Review

Review so far...

- **Microscopic energy balance**

$$\rho \hat{c}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S_e$$

rate of change (pink), convection (green), source (blue), conduction (all directions) (black)

- **Fourier's law of heat conduction**

$$\frac{q_x}{A} = -k \frac{dT}{dx}$$

(for a homogeneous phase)

- **Newton's law of cooling** (h , at the boundary)

This expression serves as the definition of the **heat transfer coefficient**.

$$\left| \frac{q_x}{A} \right| = h |T_{bulk} - T_{wall}|$$

Pending →

- **Resistances** due to k and h
- **Solving** for the steady temperature field $T(x, y, z)$

→

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Microscopic Energy Balance Review—Resistance to Heat Transfer

\mathcal{R} = Resistance to Heat Transfer

The language of **resistance** to describe the physics of heat transfer will be handy in our study of unsteady state temperature profiles. We encountered this language in CM3110, and we review and summarize now.

$$\frac{q_x}{A} = \frac{\text{driving force}}{\sum \text{resistances}}$$

Two limitations create resistance:

1. *Limited* conductivity within the homogeneous phase (k)
2. *Limited* heat transfer to the homogeneous phase through the boundary (h)

Also, resistances:

1. Are affected by geometry (rectangular versus radial)
2. Can be stacked (that is, added together like electrical resistances)

Note: Geankoplis uses a slightly different definition of resistance; we follow Bird et al. 2002.

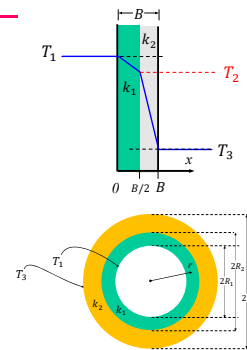
1D Heat Transfer – Resistance

Thermal conductivity k and heat transfer coefficient h may be thought of as sources of **resistance** \mathcal{R} to heat transfer.

These resistances \mathcal{R} **stack up** in a logical way, allowing us to quickly and accurately determine the effect of adding insulating layers, encountering pipe fouling, and other applications.

Using the microscopic energy balance on a test problem, we can solve for the temperature profile and then the heat flux, which is the driving force/resistance.

We can then identify the resistances for each test case considered.



$$\frac{q_x}{A} = \frac{\text{driving force}}{\sum \text{resistances}}$$

1D Rectangular Heat Transfer – Resistance

1

1D Rectangular: Composite Door

For an outside door, a metal is used (k_1) for strength, and a cork (k_2) is used for insulation. Both are the same thickness $B/2$. What is the temperature profile in the door at steady state? What is the energy flux? The inside temperature of the metal is T_1 and the outside temperature of the cork is T_3 .

Temperature boundary conditions

See handwritten notes.

https://pages.mtu.edu/~fmorriso/cm310/lectures/handnotes_lec14-15_1d_rect%20temp%20composite.pdf

Note: in the hand notes the temperatures from left to right are T_1, T_3, T_2 .

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1D Rectangular Heat Transfer – Resistance

1D Rectangular Slab, Composite Door, two equal width layers

SOLUTION:

$$\frac{q_x}{A} = \frac{(T_1 - T_3)}{\frac{B/2}{k_1} + \frac{B/2}{k_2}}$$

Let: $\mathcal{R}_i \equiv \frac{\text{thickness}}{k_i}$

Resistance due to finite thermal conductivity (rectangular)

$$\frac{q_x}{A} = \frac{(T_1 - T_3)}{\mathcal{R}_1 + \mathcal{R}_2} = \frac{\text{driving force}}{\text{resistance}}$$

Each of the layers contributes a resistance, added in *series* (like in electricity).

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1D Rectangular Heat Transfer – Resistance

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1D Rectangular: Slab with Exposed to Two Different Fluids

What is the steady state temperature profile in a rectangular slab if the **fluid** on one side is held at T_{b1} and the fluid on the other side is held at T_{b2} ?

Newton's law of cooling boundary conditions

Assumptions:

- h_1 and h_2 are the heat transfer coefficients of the left and right walls

See handwritten notes.

https://pages.mtu.edu/~fmorriso/cm310/lectures/handnotes_lec14-15_1d_rect_h.pdf

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1D Rectangular Heat Transfer – Resistance

1D Rectangular, Slab with Exposed to Two Different Fluids

SOLUTION:

$$\frac{q_x}{A} = \frac{T_{b1} - T_{b2}}{\left(\frac{1}{h_1} + \frac{B}{k} + \frac{1}{h_2}\right)}$$

Let: $\mathcal{R}_i \equiv \frac{1}{h_i}$

Let: $\mathcal{R}_i \equiv \frac{B}{k}$

Resistance due to heat transfer at boundary

Resistance due to finite thermal conductivity

$$\frac{q_x}{A} = \frac{(T_1 - T_3)}{\mathcal{R}_1 + \mathcal{R}_2 + \mathcal{R}_3} = \frac{\text{driving force}}{\text{resistance}}$$

Each of the effects contributes a resistance, added in *series*.

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1D Radial Heat Transfer – Resistance

3

1D Radial: Heat flux in a cylindrical shell – Temp BC

What is the steady state temperature profile in a cylindrical shell (pipe) if the inner wall is at T_1 and the outer wall is at T_2 , $T_1 > T_2$?

Temperature boundary conditions

See handwritten notes.
https://pages.mtu.edu/~fmorriso/cm310/lectures/handnotes_lec14-15_1d_radial_both%20bc.pdf

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1D Radial Heat Transfer – Resistance

1D Radial: Heat flux in a cylindrical shell – Temp BC

SOLUTION:

$$\frac{q_r}{A} = \left(\frac{(T_1 - T_2)}{\frac{1}{k} \ln \frac{R_2}{R_1}} \right) \frac{1}{r}$$

$\frac{1}{r}$

Radial flux NOT constant

Let: $\mathcal{R}_1 \equiv \frac{1}{k} \ln \frac{R_2}{R_1}$

Resistance due to finite thermal conductivity (radial)

$$\frac{q_r}{A} = \left(\frac{(T_1 - T_2)}{\mathcal{R}_1} \right) \frac{1}{r} = \frac{\text{driving force}}{\text{resistance}}$$

Geometry has an impact on heat-transfer resistance.

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1D Radial Heat Transfer – Resistance

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1D Radial: Insulated Pipe (Composite, radial conduction)

For a metal pipe carrying a hot liquid (k_1) an insulation layer is added with thermal conductivity k_2 . What is the temperature profile in the composite pipe at steady state? What is the flux? The inside temperature of the metal pipe is T_1 and the outside temperature of the insulation is T_3 .

Temperature boundary conditions

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1D Radial Heat Transfer – Resistance

1D Radial: Insulated Pipe

SOLUTION:

$$\frac{q_r}{A} = \left(\frac{(T_1 - T_3)}{\frac{1}{k_1} \ln \frac{R_2}{R_1} + \frac{1}{k_2} \ln \frac{R_3}{R_2}} \right) \frac{1}{r}$$

Let: $\mathcal{R}_i \equiv \frac{1}{k_i} \ln \frac{R_{i+1}}{R_i}$

Resistance due to finite thermal conductivity (radial)

$$\frac{q_r}{A} = \left(\frac{(T_1 - T_3)}{\mathcal{R}_1 + \mathcal{R}_2} \right) \frac{1}{r} = \frac{\text{driving force}}{\text{resistance}}$$

Radial flux NOT constant

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1D Radial Heat Transfer – Resistance

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1D Radial: Heat flux in a cylindrical shell – Newton's Law BC

What is the steady state temperature profile in a cylindrical shell (pipe) if the fluid on the inside is at T_{b1} and the fluid on the outside is at T_{b2} ? ($T_{b1} > T_{b2}$)

Newton's law of cooling boundary conditions

See handwritten notes.
https://pages.mtu.edu/~fmorriso/cm310/lectures/handnotes_lec14-15_1d_radial_both%20bc.pdf

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1D Radial Heat Transfer – Resistance

1D Radial: Heat flux in a cylindrical shell – Newton's Law BC

SOLUTION:

$$\frac{q_r}{A} = \frac{(T_{b1} - T_{b2})}{\frac{1}{h_2 R_2} + \frac{1}{k} \ln\left(\frac{R_2}{R_1}\right) + \frac{1}{h_1 R_1}} \left(\frac{1}{r}\right)$$

Radial flux NOT constant

Let: $\mathcal{R}_i \equiv \frac{1}{h_i R_i}$ Resistance due to heat transfer at boundary

Let: $\mathcal{R}_3 \equiv \frac{1}{k} \ln\left(\frac{R_2}{R_1}\right)$ Resistance due to finite thermal conductivity

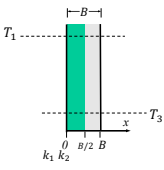
$$\frac{q_r}{A} = \left(\frac{(T_{b1} - T_{b2})}{\mathcal{R}_1 + \mathcal{R}_2 + \mathcal{R}_3} \right) \frac{1}{r} = \frac{\text{driving force}}{\text{resistance}}$$

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1D Heat Transfer – Resistance

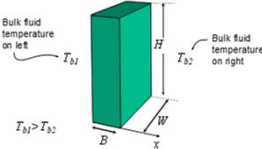
RESISTANCE SUMMARY:

1D Rectangular: Door ($k_1 = k_2$), and Composite Door



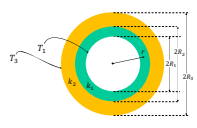
$$\frac{q_x}{A} = \frac{(T_1 - T_3)}{\left(\frac{B/2}{k_1} + \frac{B/2}{k_2}\right)}$$

1D Rectangular: Slab with Newton's law BC



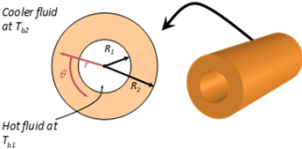
$$\frac{q_x}{A} = \frac{T_{b1} - T_{b2}}{\left(\frac{1}{h_1} + \frac{B}{k} + \frac{1}{h_2}\right)}$$

1D Radial: Pipe ($k_1 = k_2$) and Composite Pipe



$$\frac{q_r}{A} = \frac{(T_1 - T_3)}{\frac{1}{k_1} \ln \frac{R_2}{R_1} + \frac{1}{k_2} \ln \frac{R_3}{R_2}} \left(\frac{1}{r}\right)$$

1D Radial: Pipe with Newton's law BC

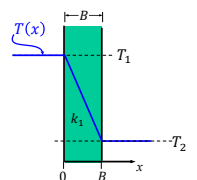


$$\frac{q_r}{A} = \frac{(T_{b1} - T_{b2})}{\frac{1}{h_2 R_2} + \frac{1}{k} \ln \left(\frac{R_2}{R_1}\right) + \frac{1}{h_1 R_1}} \left(\frac{1}{r}\right)$$

1D Heat Transfer – Resistance

1D Rectangular

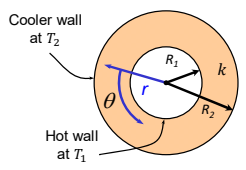
Let: $\mathcal{R} \equiv \frac{B}{k}$

$$\frac{q_x}{A} = \frac{(T_1 - T_2)}{\mathcal{R}} = \frac{\text{driving force}}{\text{resistance}}$$


Note: Geankoplis uses a different resistance. For rectangular heat flux:
 $R_{\text{Geankoplis}} = \mathcal{R}/LW$

1D Radial

Let: $\mathcal{R} \equiv \frac{1}{k} \ln \frac{R_2}{R_1}$

$$\frac{q_r}{A} = \left(\frac{T_1 - T_2}{\mathcal{R}}\right) \frac{1}{r} = \frac{\text{driving force}}{\text{resistance}}$$


Note: Geankoplis uses a different resistance. For radial heat flux:
 $R_{\text{Geankoplis}} = \mathcal{R}/2\pi L$

Temperature Boundary Conditions

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1D Heat Transfer – Resistance

Let:

1D Rectangular

$$\mathcal{R}_i \equiv \frac{1}{h_i} \text{ for } i = 1, 2$$

$$\mathcal{R}_3 \equiv \frac{B}{k}$$

$$\frac{q_x}{A} = \frac{(T_{b1} - T_{b2})}{\mathcal{R}_1 + \mathcal{R}_2 + \mathcal{R}_3} = \frac{\text{driving force}}{\text{resistance}}$$

Newton's Law of Cooling Boundary Conditions

Let:

1D Radial

$$\mathcal{R}_i \equiv \frac{1}{R_i h_i} \text{ for } i = 1, 2$$

$$\mathcal{R}_3 \equiv \frac{1}{k} \ln \frac{R_2}{R_1}$$

$$\frac{q_r}{A} = \frac{(T_{b1} - T_{b2})}{\mathcal{R}_1 + \mathcal{R}_2 + \mathcal{R}_3} \left(\frac{1}{r} \right) = \frac{\text{driving force}}{\text{resistance}}$$

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Microscopic Energy Balance—Solve for Temperature Field

**CM3110
REVIEW**

For review, let's carry out an example of 1D, steady heat transfer

CM3110
Transport I
Part II: Heat Transfer

One-Dimensional Heat Transfer
(part 1: rectangular slab)

Professor Faith Morrison
Department of Chemical Engineering
Michigan Technological University

Simple problems that allow us to identify the physics

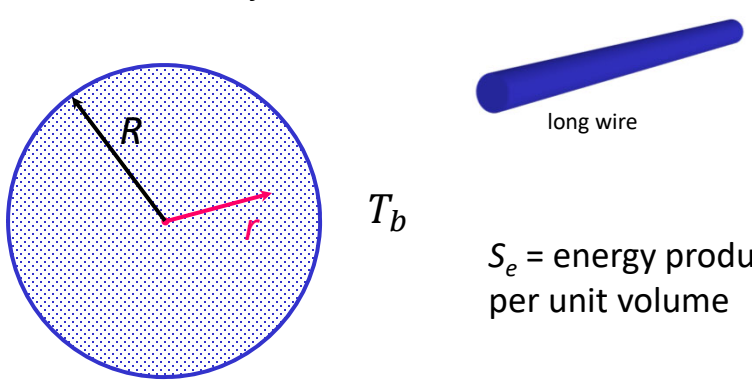
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Microscopic Energy Balance—Solve for Temperature Field

Example 3: Heat Conduction with Generation

What is the steady state temperature profile in a wire if heat is generated uniformly throughout the wire at a rate of S_e W/m³ and the bulk fluid surrounding the wire is at T_b ? What is the heat flux?



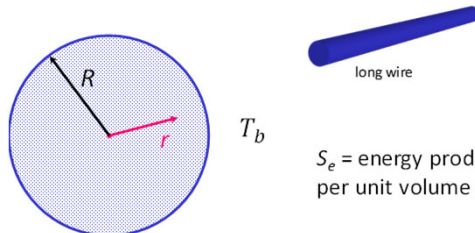
T_b

$S_e =$ energy production per unit volume

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Microscopic Energy Balance—Solve for Temperature Field

Example: Heat conduction with generation



T_b

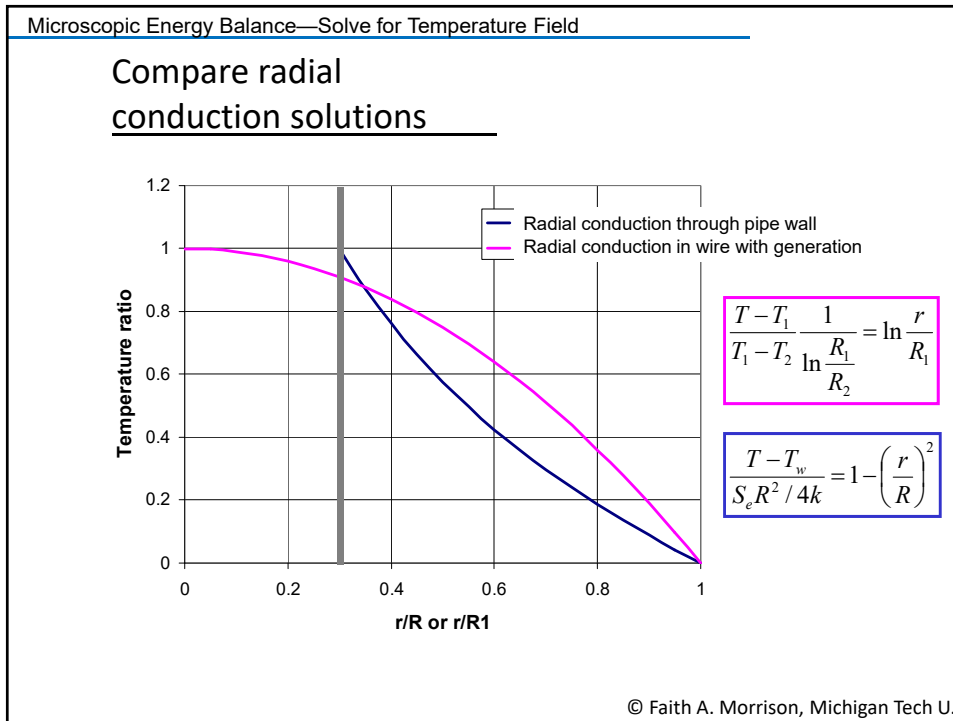
$S_e =$ energy production per unit volume

$h =$ heat transfer coefficient

Let's try.

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Unsteady State Heat Transfer

CM3110 REVIEW

Heat-Transfer Review Summary:

- Microscopic energy balance
- Fourier’s law of heat conduction (k , homogeneous)
- Newton’s law of cooling (h , at the boundary)
- **Resistances** due to k and h ; vary with boundary conditions (BC) and geometry

	T BC	h BC
1D rectangular	$\frac{B}{k}$	$\frac{1}{h}$
1D radial	$\frac{1}{k} \ln \frac{R_2}{R_1}$	$\frac{1}{Rh}$

Sneak peak: The ratio of T (internal) and h (external) resistances is the **Biot** number:

$$Bi = \frac{B/k}{1/h} = \frac{hB}{k}$$

This is important in unsteady heat transfer.


- Solving for the *steady* temperature field $T(x, y, z)$, a.k.a. “*Slash and Burn*”


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NEXT: Unsteady State Heat Transfer

CM3120 Transport/Unit Operations 2

Unsteady State Heat Transfer



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www.chem.mtu.edu/~fmorriso/cm3120/cm3120.html

