



Last time


CM3120 Transport/Unit Operations 2

Unsteady Macroscopic Energy Balance





Professor Faith A. Morrison
Department of Chemical Engineering
Michigan Technological University



© Faith A. Morrison, Michigan Tech U. ¹

Unsteady State Heat Transfer

Summary

$$Bi_{LP} = \frac{hV_{sys}}{kA_{sys}}$$

**LP good for:
 $Bi_{LP} < 0.1$**

Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature: heat transfer is limited by rate of heat transfer to the surface.	High Bi:
When the temperature is uniform in the body, we can do a macroscopic energy balance to solve many problems of interest. This is called a "lumped parameter analysis."	dominates
At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h).	Low Bi: high k , low h

Low Bi: no internal temperature variation \Rightarrow Lumped parameter analysis (macroscopic energy balance, unsteady); $Bi_{LP} = hV/kA < 0.1$

© Faith A. Morrison, Michigan Tech U. ²

2. Negligible Surface Resistance, $\frac{1}{h}$

High Bi:
low k ,
high h

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

At moderate Bi, heat transfer is affected by both the body and the surface.

At low Bi, the surface temperature is uniform in space; heat transfer is limited by rate of heat transfer to the surface (h).

High Bi:
low k ,
high h

Moderate Bi:

Low Bi:
high k ,
low h

When the wall temperature and the bulk temperature are equal, the microscopic energy balance is easier to carry out (temperature boundary conditions).

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

© Faith A. Morrison, Michigan Tech U. ³

Negligible Surface Resistance, $\frac{1}{h}$

High Bi:
low k ,
high h

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

At moderate Bi, heat transfer is affected by both the body and the surface.

At low Bi, the surface temperature is uniform in space; heat transfer is limited by rate of heat transfer to the surface (h).

High Bi:
low k ,
high h

Moderate Bi:

Low Bi:
high k ,
low h

When the wall temperature and the bulk temperature are equal, the microscopic energy balance is easier to carry out (temperature boundary conditions).

We have done many examples with constant temperature boundary conditions.

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

$$\lim_{h \rightarrow \infty} |T_{bulk} - T_{wall}| = 0$$

⇒ $T_{wall} = T_{bulk}$

© Faith A. Morrison, Michigan Tech U. ⁴

Unsteady State Heat Transfer

Summary

High Bi: dominated by internal temperature variation \Rightarrow solve with temperature boundary conditions; $Bi = hD_{char}/k$ (D_{char} varies with the problem)

Low Bi: no internal temperature variation \Rightarrow Lumped parameter analysis (macroscopic energy balance, unsteady); $Bi_{LP} = hV/kA < 0.1$

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body. **High Bi: low k , high h**

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface. **Moderate Bi: neither process dominates**

At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h). **Low Bi: high k , low h**

When the wall temperature and the bulk temperature are equal, the microscopic energy balance is easier to carry out (temperature boundary conditions).

5
© Faith A. Morrison, Michigan Tech U.

3. No Mechanism Dominates

**Moderate Bi:
neither process dominates**

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface.

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

When both processes affect the outcomes, the full solution may be necessary; for uniform starting temperatures there are charts.

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface. **Moderate Bi: neither process dominates**

At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h). **Low Bi: high k , low h**

6
© Faith A. Morrison, Michigan Tech U.

3. No Mechanism Dominates

Moderate Bi:
neither process dominates

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface.

At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h).

This is the most complicated set of cases.

© Faith A. Morrison, Michigan Tech U.

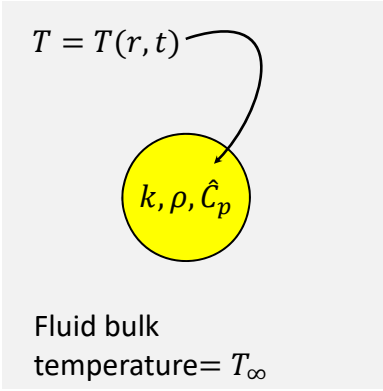
Unsteady State Heat Transfer: Intermediate Biot Number

No Mechanism Dominates

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment
- Both internal (D/k) and external ($1/h$) resistances are important
- We need to match measurable quantities with calculable quantities
- ⇒ **Microscopic** Energy Balance
- ⇒ *Uncertainty considerations*

$T = T(r, t)$



Fluid bulk temperature = T_∞

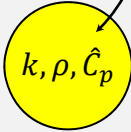
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number No Mechanism Dominates

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment
- Both internal (D/k) and external ($1/h$) resistances are important
- We need to match measurable quantities with calculable quantities
- \Rightarrow **Microscopic** Energy Balance
- \Rightarrow *Uncertainty considerations*

$T = T(r, t)$



Fluid bulk temperature = T_∞

?


© Faith A. Morrison, Michigan Tech U. ⁹

Unsteady State Heat Transfer: Intermediate Biot Number No Mechanism Dominates

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment
- Both internal (D/k) and external ($1/h$) resistances are important
- We need to match measurable quantities with calculable quantities
- \Rightarrow **Microscopic** Energy Balance
- \Rightarrow *Uncertainty considerations*

$T = T(r, t)$



Fluid bulk temperature = T_∞

?

You try.

© Faith A. Morrison, Michigan Tech U. ¹⁰

Thinking

- Create an unsteady state heat transfer situation...
- Measure ...?
- Compare ...?
- Consider uncertainty in measurements ... ?

Unsteady State Heat Transfer: Intermediate Biot Number

Experiment: Measure $T(t)$ at the center of a sphere ($r = 0$):

Initially:

$t < t_0$
 $T = T_0$

T-couple measures $T(t)$ at the center of the sphere

Suddenly:

$t \geq t_0$
 $T = T(r, t) = T(0, t)$

Unsteady state heat transfer takes place.

Unsteady State Heat Transfer: Intermediate Biot Number

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment.
- Both internal (x) and external (x) resistances are important.
- We need to match measurable quantities with calculable quantities.
- = Microscopic Energy Balance
- = Uncertainty considerations

$T = T(r, t)$

k, ρ, c_p

Fluid bulk temperature = T_∞

© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number

Experiment: Measure $T(t)$ at the center of a sphere ($r = 0$):

Initially:

$t < t_0$
 $T = T_0$

T-couple measures $T(t)$ at the center of the sphere

Suddenly:

$t \geq t_0$
 $T = T(r, t) = T(0, t)$

Excel:

$t(s)$	$T(^{\circ}C)$
9.50E-02	7.46E+00
2.11E-01	7.44E+00
3.09E-01	7.44E+00
4.09E-01	7.57E+00
5.24E-01	7.46E+00
6.23E-01	7.49E+00
7.39E-01	7.53E+00
8.37E-01	7.46E+00
9.54E-01	7.59E+00
1.05E+00	7.53E+00
1.15E+00	7.58E+00
1.27E+00	7.48E+00
1.37E+00	7.57E+00
...	...

© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number

Modeling

What are the modeling equations?

Experiment: Measure $T(t)$ at the center of a sphere ($r = 0$):

Initially:

$t < t_0$
 $T = T_0$

Suddenly:

$t \geq t_0$
 $T = T(r, t) = T(0, t)$

$t(s)$	$T(^{\circ}C)$
9.50E-02	7.46E+00
2.11E-01	7.44E+00
3.09E-01	7.44E+00
4.02E-01	7.57E+00
5.24E-01	7.46E+00
6.23E-01	7.49E+00
7.39E-01	7.53E+00
8.37E-01	7.46E+00
9.94E-01	7.59E+00
1.05E+00	7.53E+00
1.15E+00	7.58E+00
1.27E+00	7.48E+00
1.37E+00	7.57E+00
...	...

13
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment
- Both internal (k) and external (h) resistances are important
- We need to match measurable quantities with calculable quantities
- \Rightarrow **Microscopic** Energy Balance
- \Rightarrow *Uncertainty considerations*

$T = T(r, t)$

Fluid bulk temperature = T_{∞}

Initially:

$t < t_0$
 $T = T_0$

Suddenly:

$t \geq t_0$
 $T = T(t)$

Can we meet our objective?

To determine h :

- Measure center-point temperature as a function of time
- Compare with model predictions, accounting for uncertainty in measurements
- Deduce h

14
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number

Modeling

What are the modeling equations?

Experiment: Measure $T(t)$ at the center of a sphere ($r = 0$):

Initially:

$t < t_0$
 $T = T_0$

Suddenly:

$t \geq t_0$
 $T = T(r, t) = T(0, t)$

Excel:

$t(s)$	$T(^{\circ}C)$
9.50E-02	7.46E+00
2.11E-01	7.44E+00
3.09E-01	7.44E+00
4.02E-01	7.57E+00
5.24E-01	7.46E+00
6.23E-01	7.49E+00
7.39E-01	7.53E+00
8.37E-01	7.46E+00
9.94E-01	7.59E+00
1.05E+00	7.53E+00
1.15E+00	7.58E+00
1.27E+00	7.48E+00
1.37E+00	7.57E+00
...	...

You try.

15
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer: Intermediate Biot Number

Microscopic Energy Balance

Microscopic energy balance, constant thermal conductivity; Gibbs notation

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S$$

Microscopic energy balance, constant thermal conductivity; Cartesian coordinates

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

Microscopic energy balance, constant thermal conductivity; cylindrical coordinates

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = k \left(\frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right) + S$$

Microscopic energy balance, constant thermal conductivity; spherical coordinates

$$\rho \hat{C}_p \left(\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} \right) = k \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 T}{\partial \phi^2} \right) + S$$

www.chem.mtu.edu/~fmorriso/cm310/energy2013.pdf

16
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer to a Sphere

Microscopic energy balance in the sphere:

$$\frac{\partial T}{\partial t} = \frac{k}{\rho \hat{C}_p} \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) \right) \quad \equiv \alpha$$

Boundary conditions:

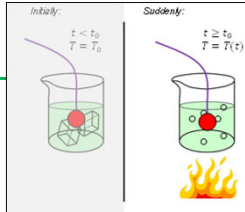
$$r = R, \quad \frac{q_r}{A} = -k \frac{\partial T}{\partial r} = h(T(r) - T_{bulk}) \quad t > 0$$

$$r = 0, \quad \frac{q_r}{A} = -k \frac{\partial T}{\partial r} = 0 \quad \forall t$$

Initial condition:

$$t = 0, \quad T = T_{initial} \quad \forall r$$

- Unsteady
- Solid ($v = 0$)
- θ, ϕ symmetry
- No current, no rxn



© Faith A. Morrison, Michigan Tech U.

17

Unsteady State Heat Transfer to a Sphere

Microscopic energy balance in the sphere:

$$\frac{\partial T}{\partial t} = \frac{k}{\rho \hat{C}_p} \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) \right) \quad \equiv \alpha$$

Boundary conditions:

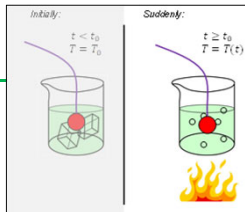
$$r = R, \quad \frac{q_r}{A} = -k \frac{\partial T}{\partial r} = h(T(r) - T_{bulk}) \quad t > 0$$

$$r = 0, \quad \frac{q_r}{A} = -k \frac{\partial T}{\partial r} = 0 \quad \forall t$$

Initial condition:

$$t = 0, \quad T = T_{initial} \quad \forall r$$

- Unsteady
- Solid ($v = 0$)
- θ, ϕ symmetry
- No current, no rxn



© Faith A. Morrison, Michigan Tech U.

18

("∀" means "for all")

Unsteady State Heat Transfer to a Sphere

Microscopic energy balance in the sphere:

$$\frac{\partial T}{\partial t} = \frac{k}{\rho \hat{C}_p} \left(\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) \right) \quad \equiv \alpha$$

- Unsteady
- Solid ($v = 0$)

Now, Solve

$r = 0, \quad \frac{\partial T}{\partial r} = -k \frac{\partial T}{\partial r} = 0 \quad (\forall t)$

Initial condition:
 $t = 0, \quad T = T_{initial} \quad (\forall r)$

("∀" means "for all")

© Faith A. Morrison, Michigan Tech U.

19

Unsteady State Heat Transfer to a Sphere

Microscopic energy balance in the sphere:

Boundary conditions:

$r = R, \quad -k \frac{\partial T}{\partial r} = h(T - T_{\infty})$

$r = 0, \quad \frac{\partial T}{\partial r} = 0$

Initial condition:
 $t = 0, \quad T = T_{initial}$

Conduction of Heat in Solids
SECOND EDITION
OXFORD AT THE CLARENDON PRESS
H. S. CARSLAW and J. C. JAEGER

Unsteady
Solid ($v = 0$)
 θ, ϕ symmetry
No current, no rxn

$t > 0$

$\forall t$

$\forall r$

© Faith A. Morrison, Michigan Tech U.

20

Unsteady State Heat Transfer to a Sphere

Solution:

$$\xi(r, t) \equiv \frac{T(r, t) - T_{bulk}}{T_{initial} - T_{bulk}}$$

$$Bi \equiv \frac{hR}{k} \quad Fo \equiv \frac{\alpha t}{R^2}$$

Bi = Biot number;
Fo = Fourier number

$$\xi = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-Fo(\lambda_n R)^2} \left(\frac{\sin r \lambda_n}{r \lambda_n} \right) \left(\frac{\sin R \lambda_n}{R \lambda_n} \right) \left(\frac{(R \lambda_n)^2 + (Bi - 1)^2}{(R \lambda_n)^2 + Bi(Bi - 1)} \right)$$

where the **eigenvalues λ_n** satisfy this equation:

$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1 = 0$$

Characteristic Equation

(Carslaw and Yeager, 1959, eqn 10, p238)
Incropera and DeWitt, 7th ed, eqn 5.51a, p303

21
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer to a Sphere

Solution:

$$\xi(r, t) \equiv \frac{T(r, t) - T_{bulk}}{T_{initial} - T_{bulk}}$$

Depends on **material** ($\alpha = k/\rho\hat{C}_p$), and heat transfer processes at **surface** (h)

$$Bi \equiv \frac{hR}{k} \quad Fo \equiv \frac{\alpha t}{R^2}$$

Bi = Biot number;
Fo = Fourier number

$$\xi = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-Fo(\lambda_n R)^2} \left(\frac{\sin r \lambda_n}{r \lambda_n} \right) \left(\frac{\sin R \lambda_n}{R \lambda_n} \right) \left(\frac{(R \lambda_n)^2 + (Bi - 1)^2}{(R \lambda_n)^2 + Bi(Bi - 1)} \right)$$

where the **eigenvalues λ_n** satisfy this equation:

$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1 = 0$$

Characteristic Equation

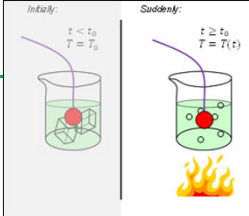
We're interested in $T(r, t)$ at the center of the sphere, $r = 0$.

(Carslaw and Yeager, 1959, eqn 10, p238)
Incropera and DeWitt, 7th ed, eqn 5.51a, p303

22
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer to a Sphere

What does *this* look like?



$$\xi(0, Fo) = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-Fo(\lambda_n R)^2} \left(\frac{\sin R\lambda_n}{R\lambda_n} \right) \left(\frac{(R\lambda_n)^2 + (Bi - 1)^2}{(R\lambda_n)^2 + Bi(Bi - 1)} \right)$$

where the eigenvalues λ_n satisfy this equation:

$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1 = 0$$

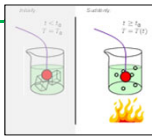
Characteristic Equation

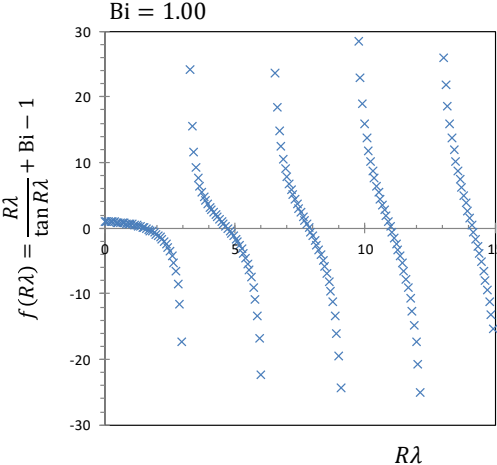
Let's plot it to find out. (Excel)

© Faith A. Morrison, Michigan Tech U. ²³

Unsteady State Heat Transfer to a Sphere

Eigenvalues are the roots of the characteristic equation

$$Bi \equiv \frac{hR}{k}$$




Characteristic Equation:

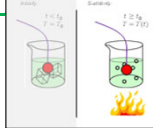
$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1$$

- The λ_n are the roots (zero crossings) of the characteristic equation
- They depend on Biot number Bi

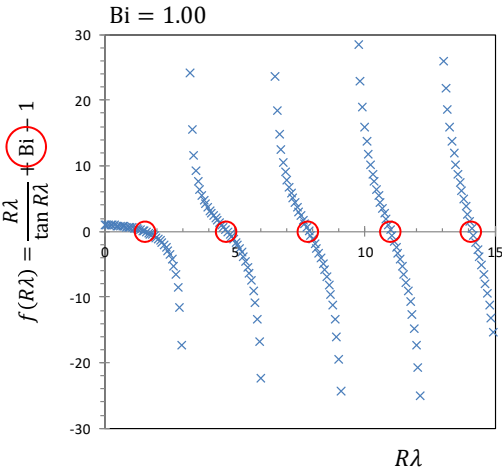
© Faith A. Morrison, Michigan Tech U. ²⁴

Unsteady State Heat Transfer to a Sphere

Eigenvalues are the roots of the characteristic equation

$$Bi \equiv \frac{hR}{k}$$


Bi = 1.00



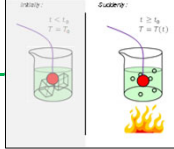
Characteristic Equation:

$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1$$

- The λ_n are the roots (zero crossings) of the characteristic equation
- They depend on Biot number Bi

© Faith A. Morrison, Michigan Tech U. 25

Let's plot it to find out: what are the variables?

$$Bi \equiv \frac{hR}{k} \quad Fo \equiv \frac{\alpha t}{R^2}$$


Solution:

$$\xi(0, Fo) = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-Fo(\lambda_n R)^2} \left(\frac{\sin R\lambda_n}{R\lambda_n} \right) \left(\frac{(R\lambda_n)^2 + (Bi - 1)^2}{(R\lambda_n)^2 + Bi(Bi - 1)} \right)$$

$\xi(0, Fo) = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-(Fo)(\lambda_n R)^2}$ (Exponential decay with Fo (scaled time) bunch of terms that vary with Bi and $\lambda_n(Bi)$)

$\lambda_n(Bi)$ varies only with Bi and n:

$$\frac{R\lambda_n}{\tan R\lambda_n} + Bi - 1 = 0$$

Characteristic Equation

If we choose a fixed Bi, then ξ only varies with Fo

© Faith A. Morrison, Michigan Tech U. 26

If we choose a fixed Bi, then ξ only varies with Fo

For a fixed Bi:

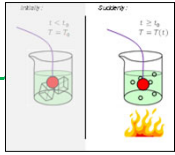
$$\xi(0, Fo) = \frac{T - T_b}{T_i - T_b} = 2Bi \sum_{n=1}^{\infty} e^{-Fo(\lambda_n R)^2} \left(\frac{\sin R\lambda_n}{R\lambda_n} \right) \left(\frac{(R\lambda_n)^2 + (Bi - 1)^2}{(R\lambda_n)^2 + Bi(Bi - 1)} \right)$$

Exponential decay with Fo (scaled time)

bunch of terms that vary with Bi and $\lambda_n(Bi)$

An infinite sum of decaying **exponentials**

- whose *argument* is Fourier number scaled by something that depends on Biot number and n
- with a prefactor that also depends on Biot number and n



© Faith A. Morrison, Michigan Tech U. 27

If we choose a fixed Bi, then ξ only varies with Fo

For a fixed Bi:

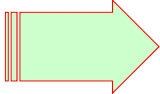
$$\xi(0, Fo) = \sum_{n=1}^{\infty} \tilde{C}_n e^{-\lambda_n^2 R^2 Fo}$$

An infinite sum of decaying **exponentials**

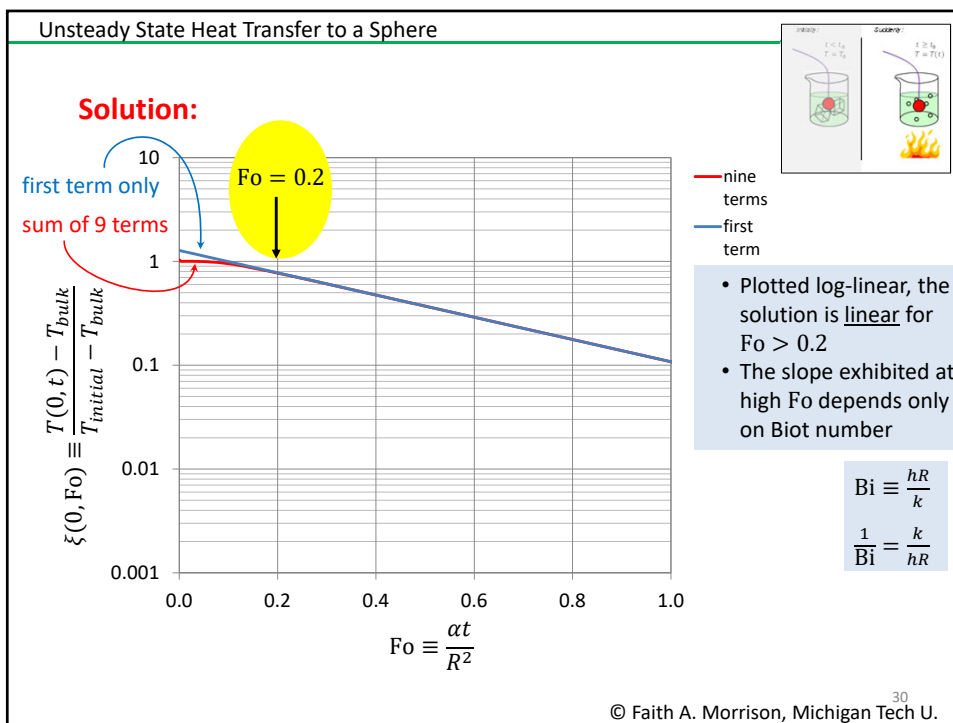
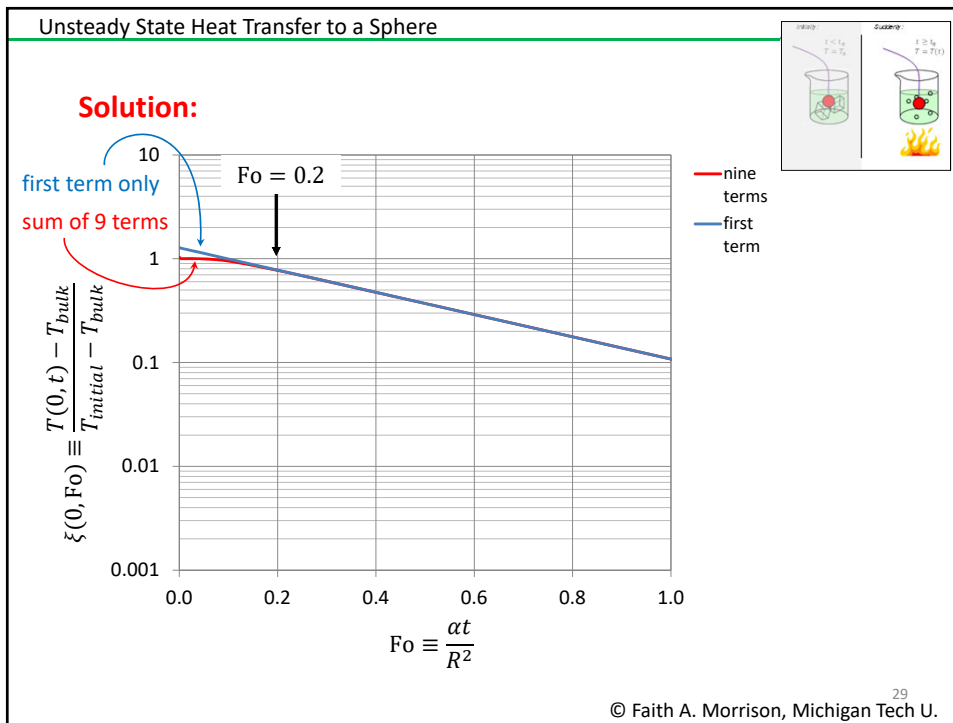
- \tilde{C}_n depends on n through λ_n
- λ_n are calculated (numerically) from the roots of this equation:

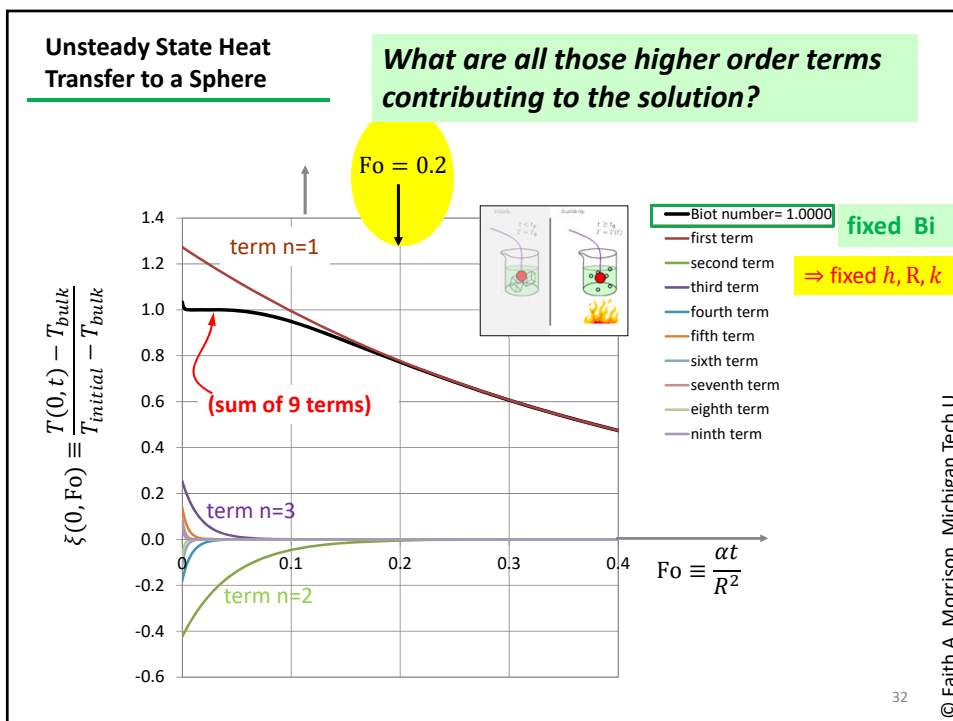
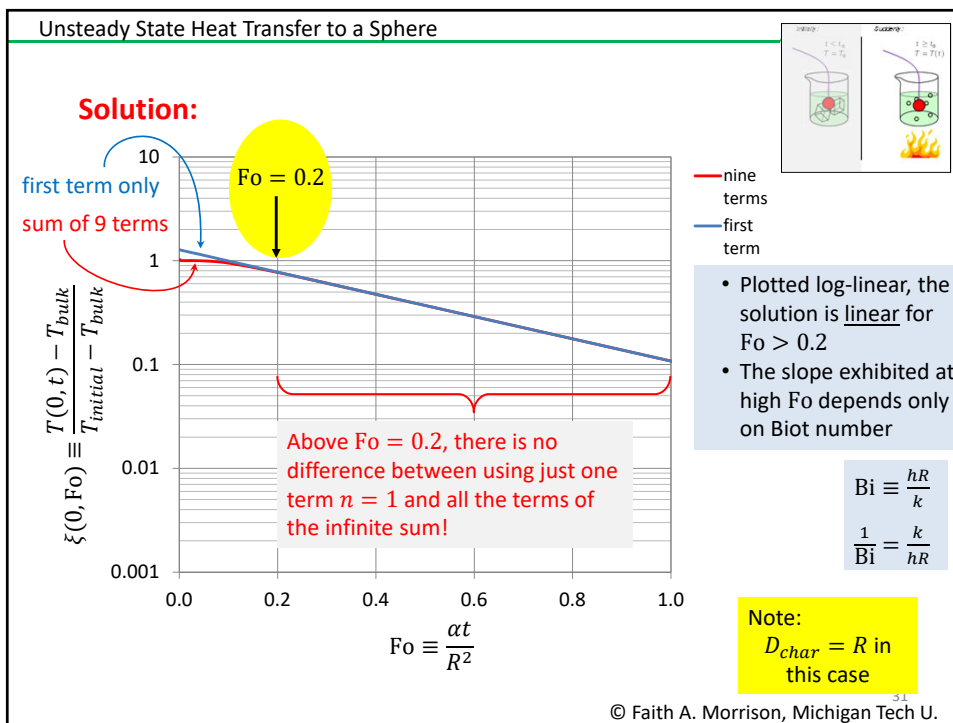
$$f(R\lambda) = \frac{R\lambda}{\tan R\lambda} + Bi - 1 = 0$$

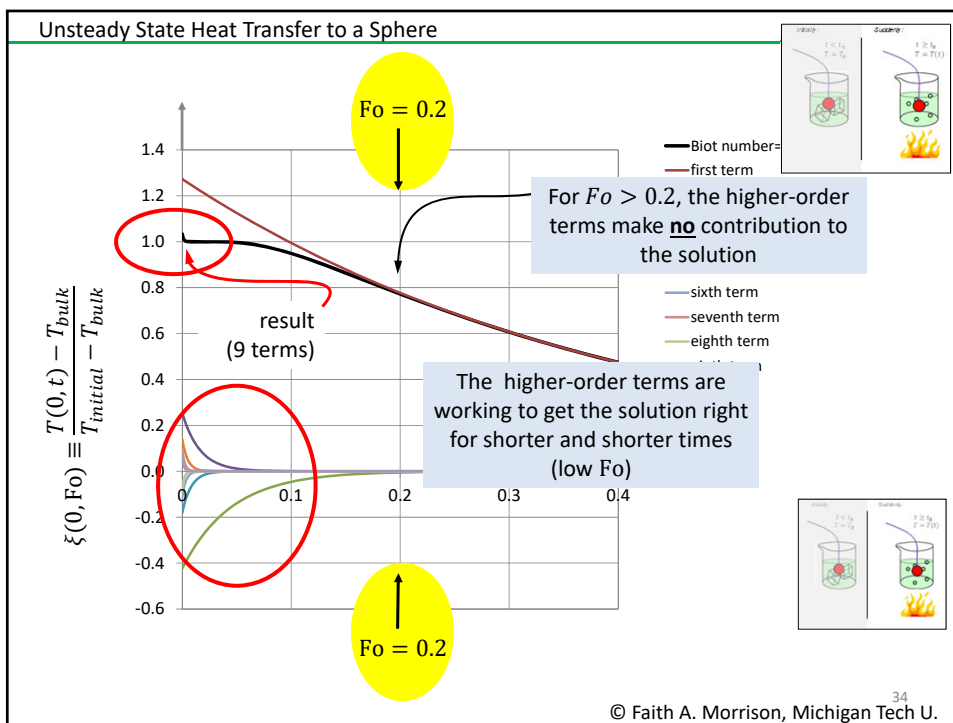
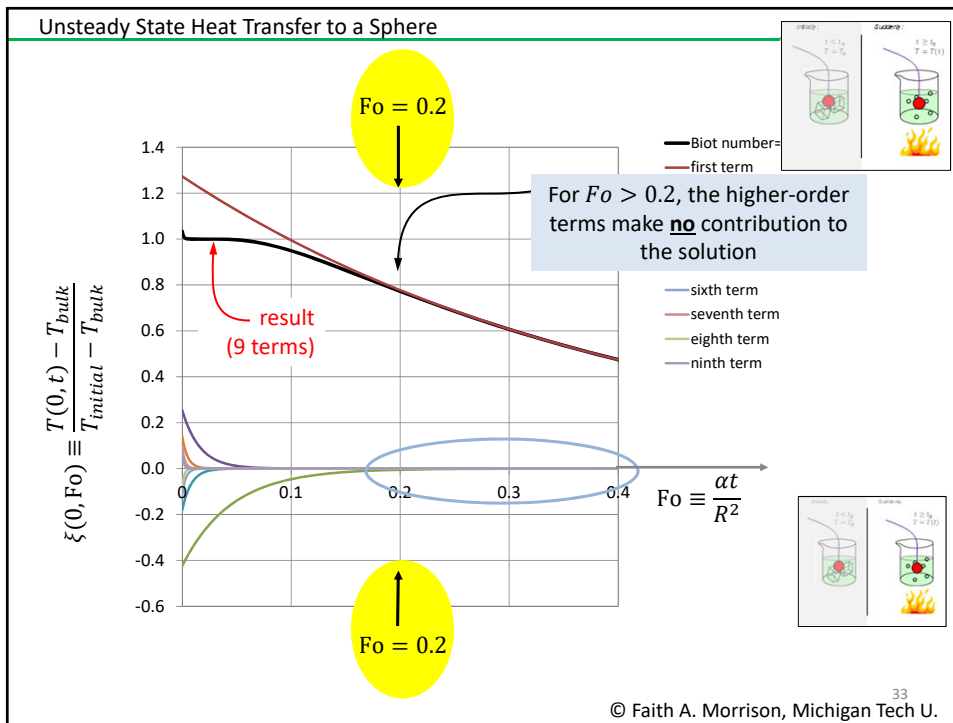
Let's plot $\xi(0, Fo)$

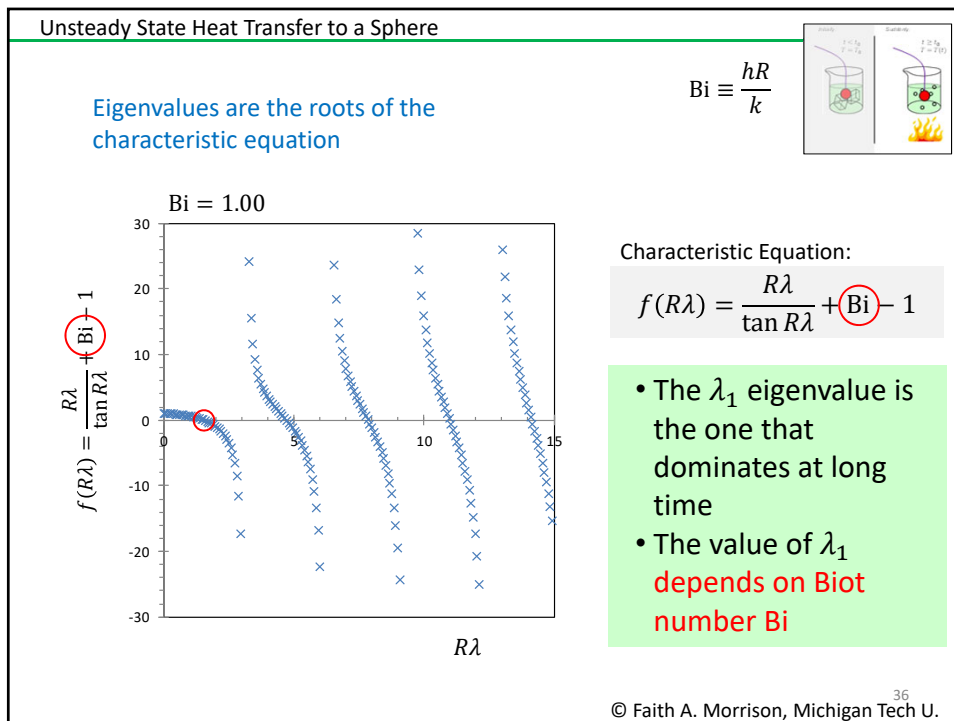
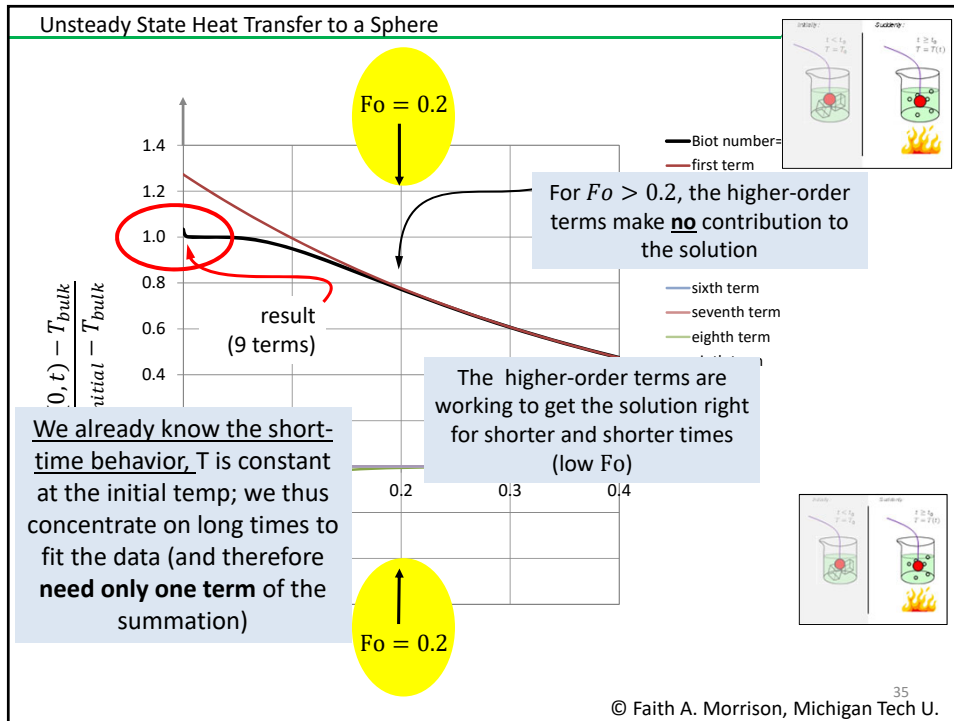


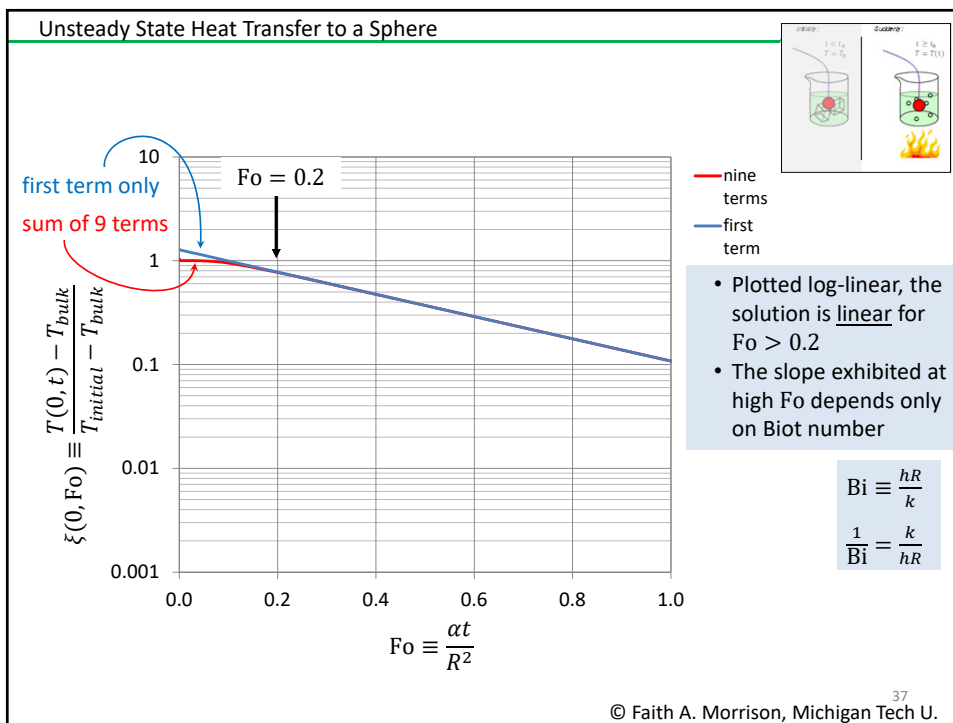
© Faith A. Morrison, Michigan Tech U. 28











Unsteady State Heat Transfer to a Sphere

No Mechanism Dominates

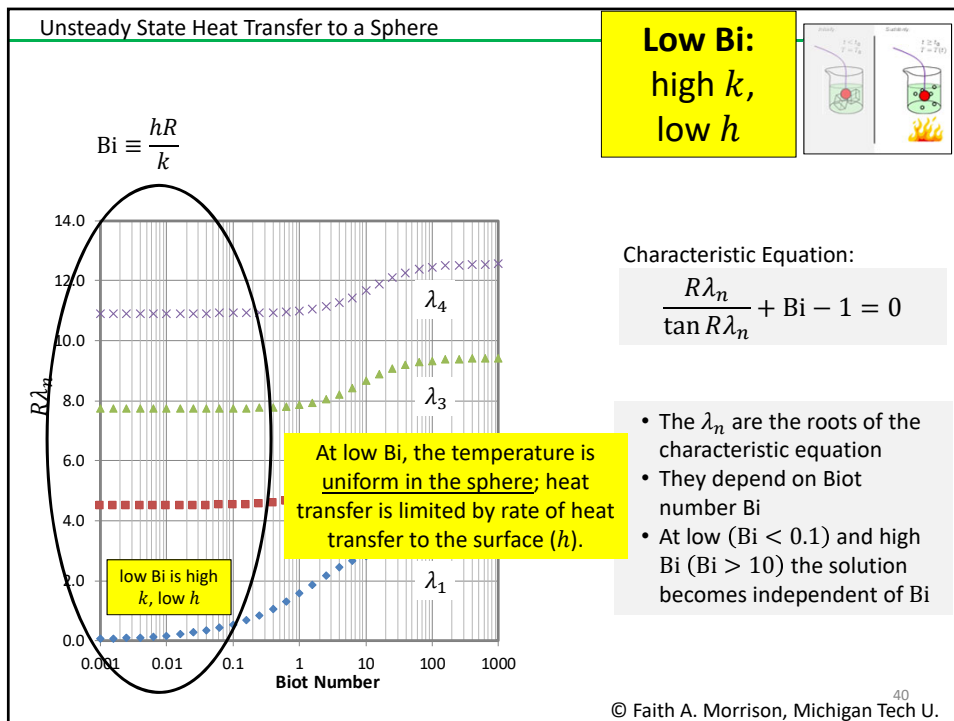
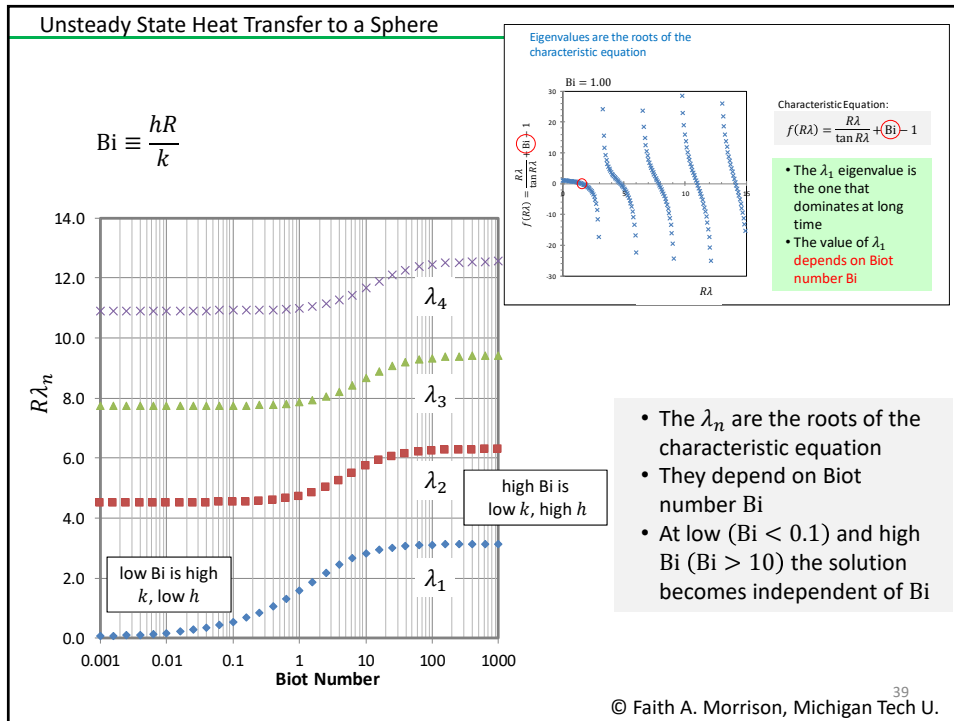
Summary

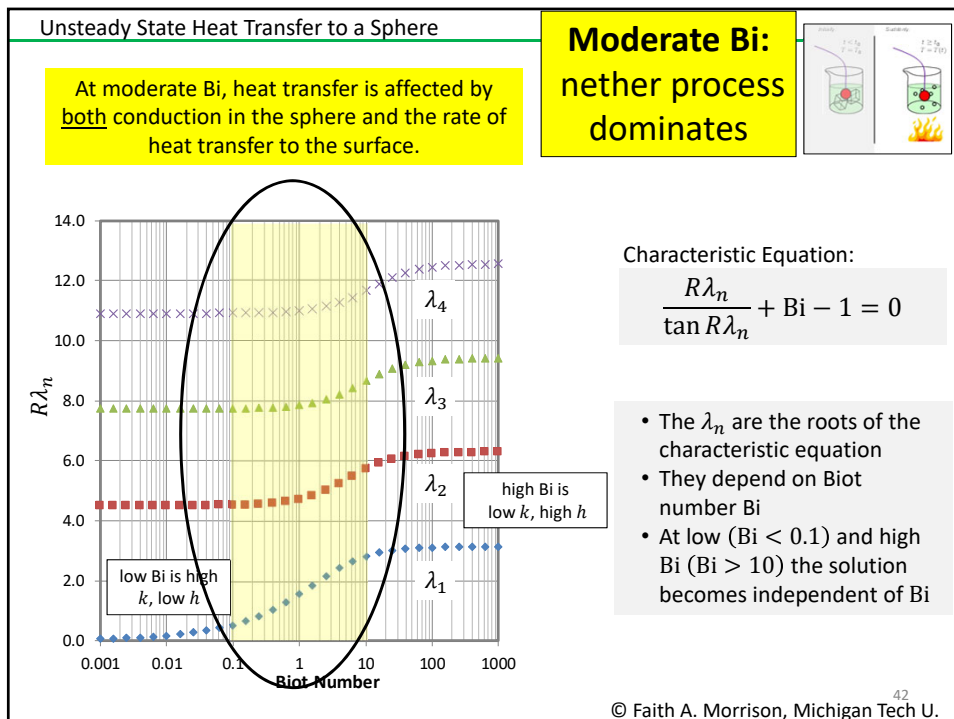
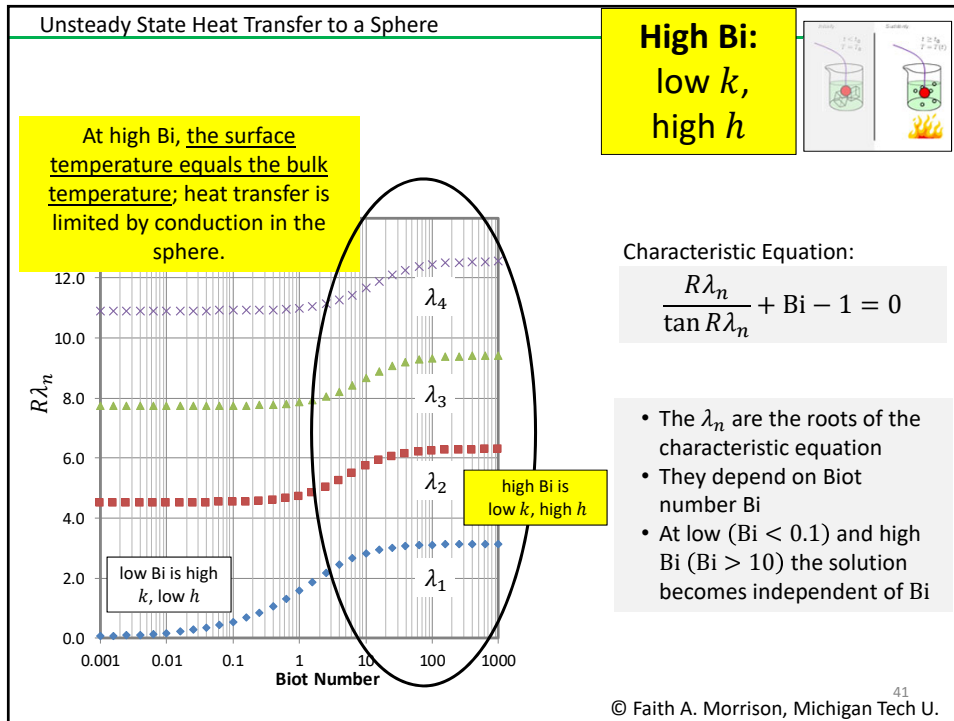
- For a fixed Bi the results are only a function of Fo .
- Solution is an infinite sum of terms.
- Each term corresponds to one eigenvalue, λ_n
- The first term $n = 1$ (λ_1) is the dominant term
- The $n > 1$ terms alternate in sign (positive and negative)
- Higher terms are “fixing” the short time behavior
- At fixed Biot number, the time-dependence is an exponential decay (for $Fo > 0.2$); this is linear on a log-linear plot versus Fo

So, actually, it turns out all we need are those slopes as a function of Biot number.

Question: How do various values of Biot number affect the heat transfer that occurs?

© Faith A. Morrison, Michigan Tech U.





What were we trying to do?

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

Where are we in the process?

- ✓ We have the model
- We need the measured center-point temperature as a function of time
- We need to compare the two to deduce h .

Unsteady State Heat Transfer: Intermediate Biot Number

Example: Measure the convective heat-transfer coefficient for heat being transferred between a fluid and a sphere.

- We need to devise an experiment
- Both internal (λ) and external (h) resistances are important
- We need to match measurable quantities with calculable quantities
- → Microscopic: Energy Balance
- → Uncertainty considerations

$T = T(r, t)$

Fluid bulk temperature = T_∞

Initially: $t < t_0$, $T = T_\infty$

Suddenly: $t \geq t_0$, $T = T(t)$

Can we meet our objective?

To determine h :

- Measure center-point temperature as a function of time
- Compare with model predictions, accounting for uncertainty in measurements
- Deduce h

The solution of the model:

$$\xi(0, Fo) = \sum_{n=1}^{\infty} \tilde{C}_n e^{-\lambda_n^2 R^2 Fo}$$

$$Fo \equiv \frac{\alpha t}{R^2}$$

$$Bi \equiv \frac{hR}{k}$$

Use to interpret data.

For a fixed Bi, $Fo > 0.2$:

$$\xi(0, Fo) \approx \tilde{C}_1 e^{-\lambda_1^2 R^2 Fo}$$

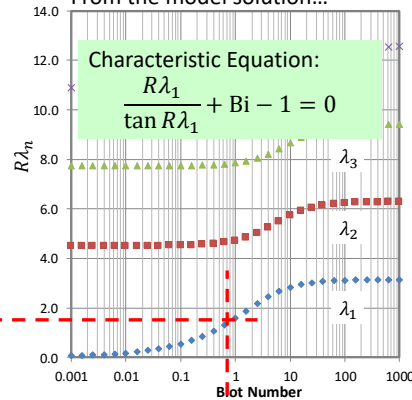
$$\ln \xi(0, Fo) = \ln(\tilde{C}_1) - \lambda_1^2 R^2 Fo$$

From experiments...

Plot: $\ln \xi = \ln\left(\frac{T - T_b}{T_i - T_b}\right)$ vs Fo

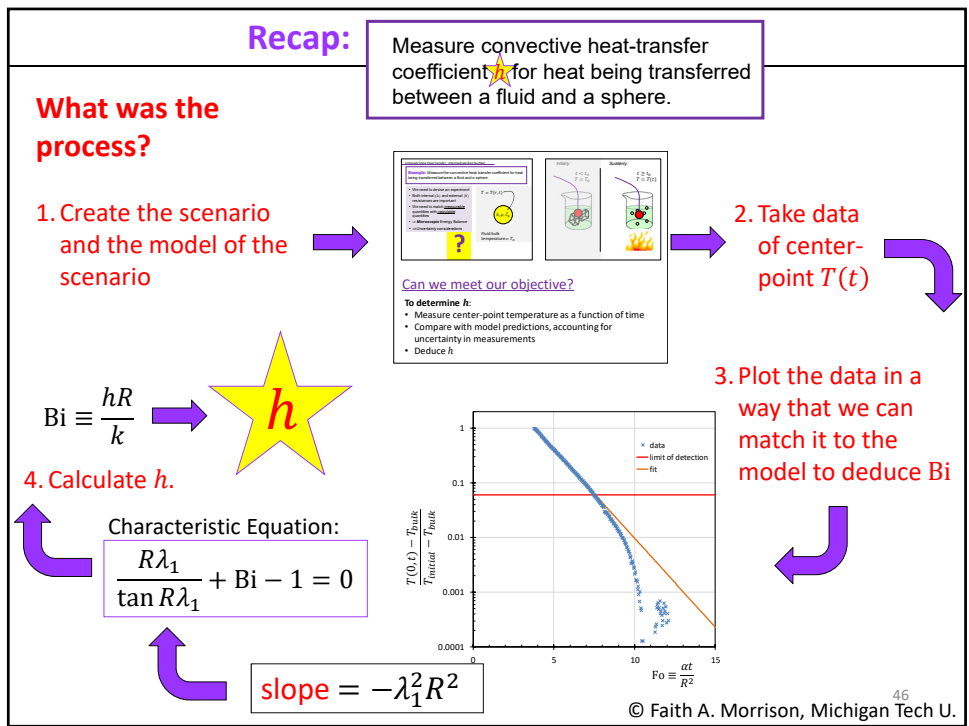
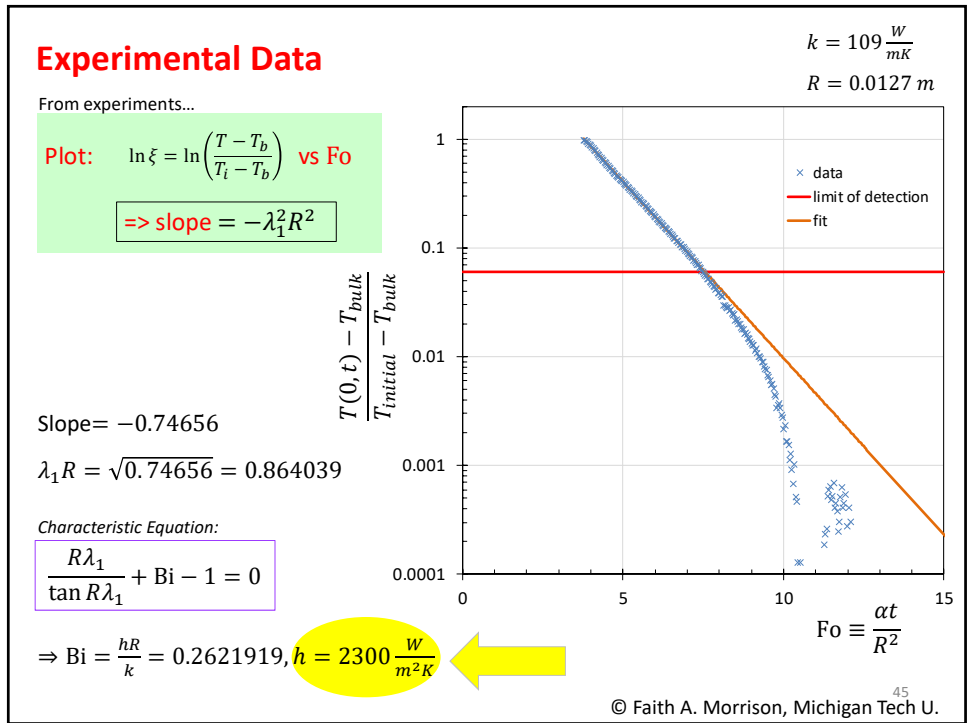
$$\Rightarrow \text{slope} = -\lambda_1^2 R^2$$

From the model solution...



Once we know Bi, we can calculate h from Bi

$$Bi \equiv \frac{hR}{k}$$



Unsteady State Heat Transfer

Summary

High Bi: dominated by internal temperature variation \Rightarrow solve with temperature boundary conditions; $Bi = hD_{char}/k$ (D_{char} varies with the problem)

Moderate Bi: The limits for "moderate" are $0.1 \leq Bi \leq 10$. When Bi is in this range, a more complete solution may be necessary; $Bi = hD_{char}/k$. (D_{char} varies with the problem)

Low Bi: no internal temperature variation \Rightarrow Lumped parameter analysis (macroscopic energy balance, unsteady); $Bi = hV/kA < 0.1$

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface.

At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h).

High Bi:
low k ,
high h

Moderate Bi:
neither process dominates

Low Bi:
high k ,
low h

47
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer

Summary

High Bi: dominated by internal temperature variation \Rightarrow solve with temperature boundary conditions; $Bi = hD_{char}/k$ (D_{char} varies with the problem)

Moderate Bi: The limits for "moderate" are $0.1 \leq Bi \leq 10$. When Bi is in this range, a more complete solution may be necessary; $Bi = hD_{char}/k$. (D_{char} varies with the problem)

Low Bi: no internal temperature variation \Rightarrow Lumped parameter analysis (macroscopic energy balance, unsteady); $Bi = hV/kA < 0.1$

Bi – Biot Number = $\frac{hD}{k}$ Quantifies the tradeoffs between the rate of internal heat flux (by conduction, k) and the rate of heat delivery to the boundary (by convection, h)

At high Bi, the surface temperature equals the bulk temperature; heat transfer is limited by conduction in the body.

At moderate Bi, heat transfer is affected by both conduction in the body and the rate of heat transfer to the surface.

At low Bi, the temperature is uniform in a finite body; heat transfer is limited by rate of heat transfer to the surface (h).

High Bi:
low k ,
high h

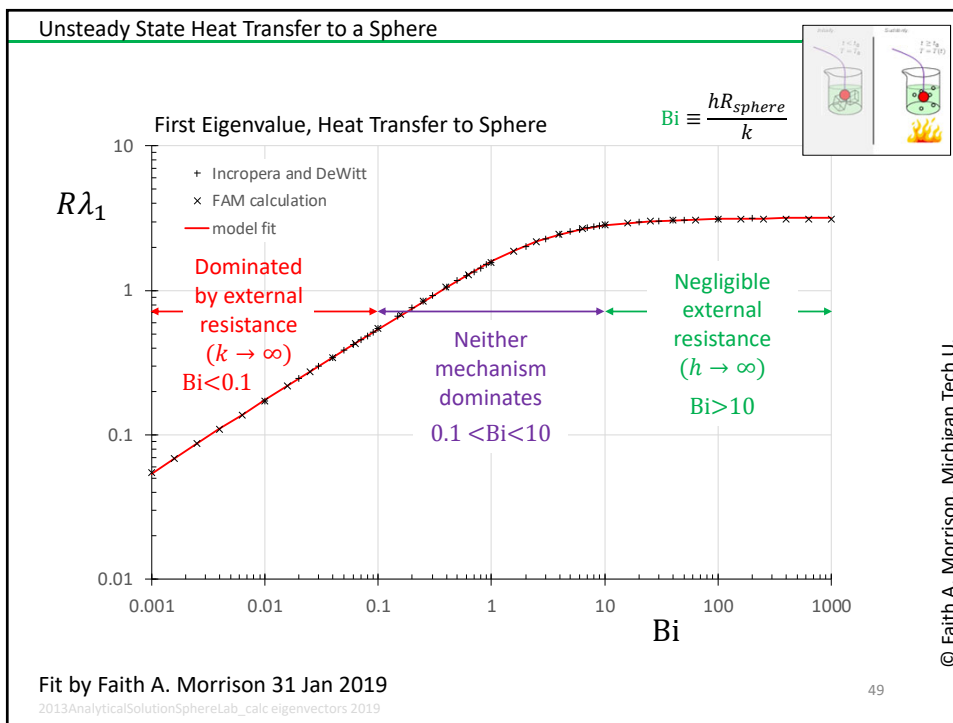
Moderate Bi:
neither process dominates

Low Bi:
high k ,
low h

D_{char} = characteristic length scale

We use $D_{LP} = V/A$ **only** for the lumped parameter analysis. We use different D_{char} in other cases.

48
© Faith A. Morrison, Michigan Tech U.



Unsteady State Heat Transfer to a Sphere

If we know $R\lambda_n$ and we're determining Bi (i.e. h), we use the characteristic equation directly.

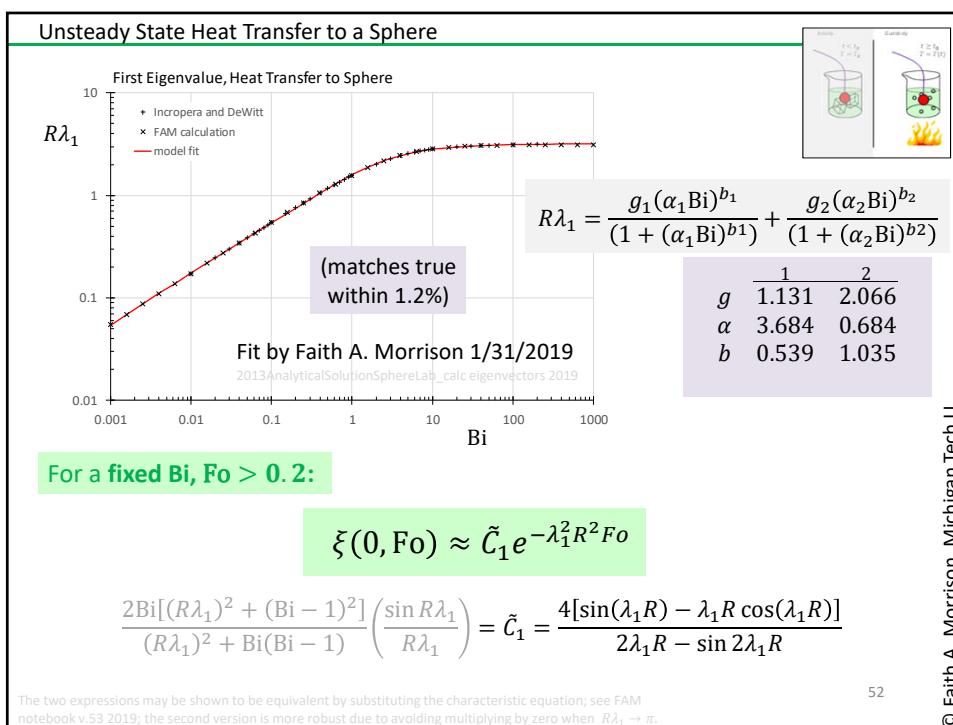
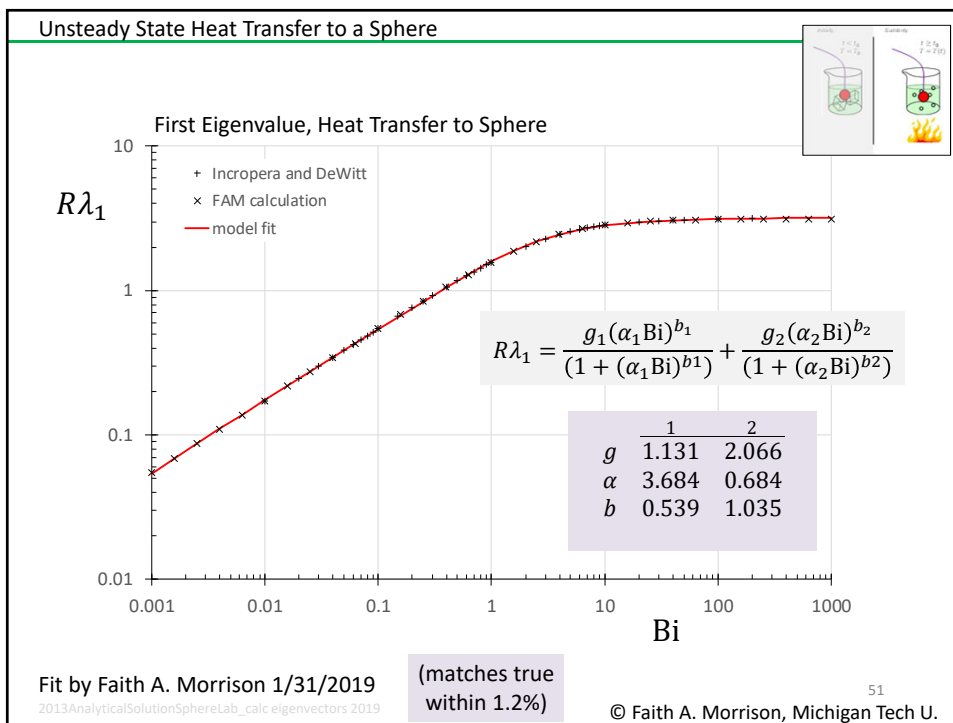
If we know h (and hence, Bi) we need to find $\lambda_1 R$ from an iterative solution of the characteristic equation.

Or use a table or correlation for the calculated roots.

Characteristic Equation:

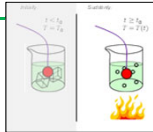
$$\frac{R\lambda_n}{\tan R\lambda_n} + Bi - 1 = 0$$

50
© Faith A. Morrison, Michigan Tech U.



Unsteady State Heat Transfer to a Sphere

Reasonable estimates of the sphere (slab, cylinder) solutions may also be obtained from the "Heisler Charts"



Heisler charts

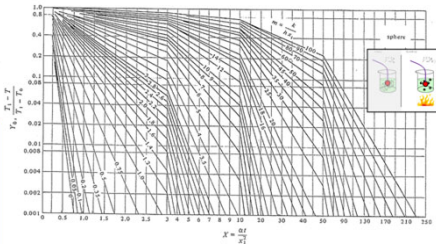
See: Geankoplis, Wikipedia, Welty, Rorrer, and Foster (appendix F)

D_{char} = characteristic length scale

For the Heisler chart for spheres, we use $D_{char} = R$. Note this is not the same as what is used in lumped parameter analysis

$$Bi = \frac{hD_{char}}{k}$$

$D_{char} = x_1 = R$ (sphere)
 R (cylinder)
 B (slab of thickness $2B$)



From Geankoplis, 4th edition, page 374

53

© Faith A. Morrison, Michigan Tech U.

Literature solutions to Unsteady State Heat Transfer to a Sphere

Heisler charts (Geankoplis; see also Wikipedia)

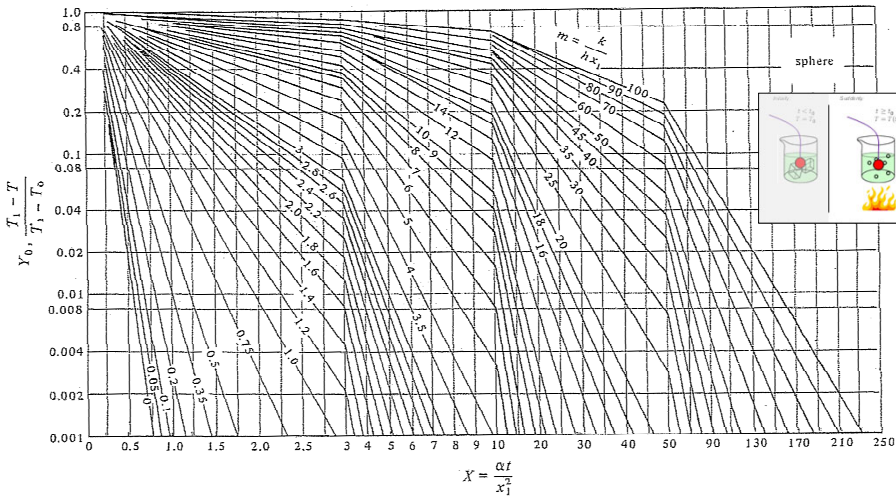
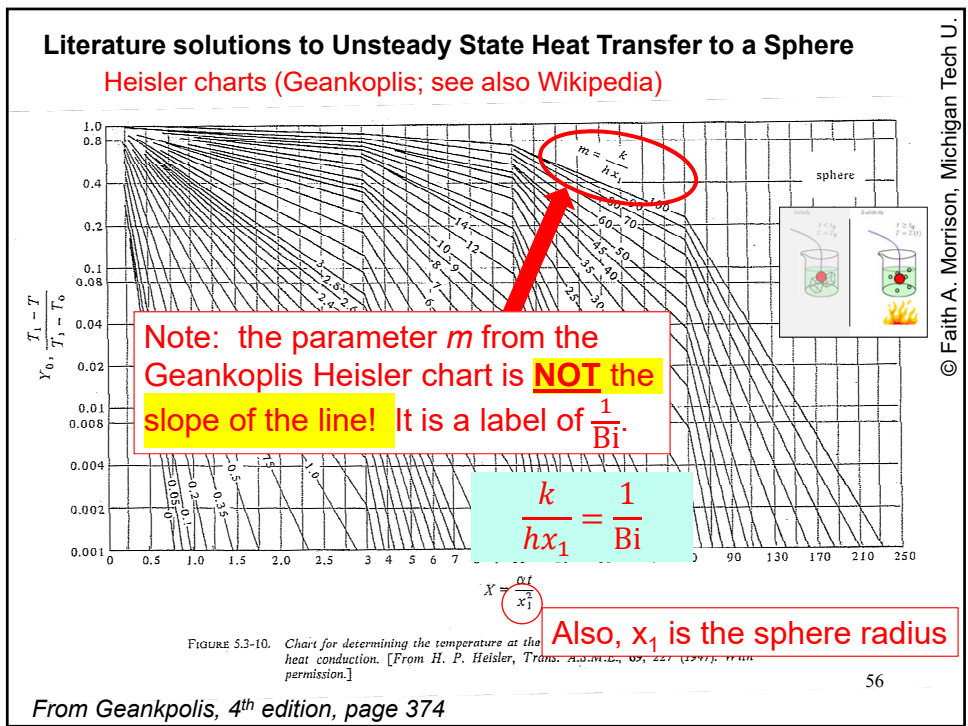
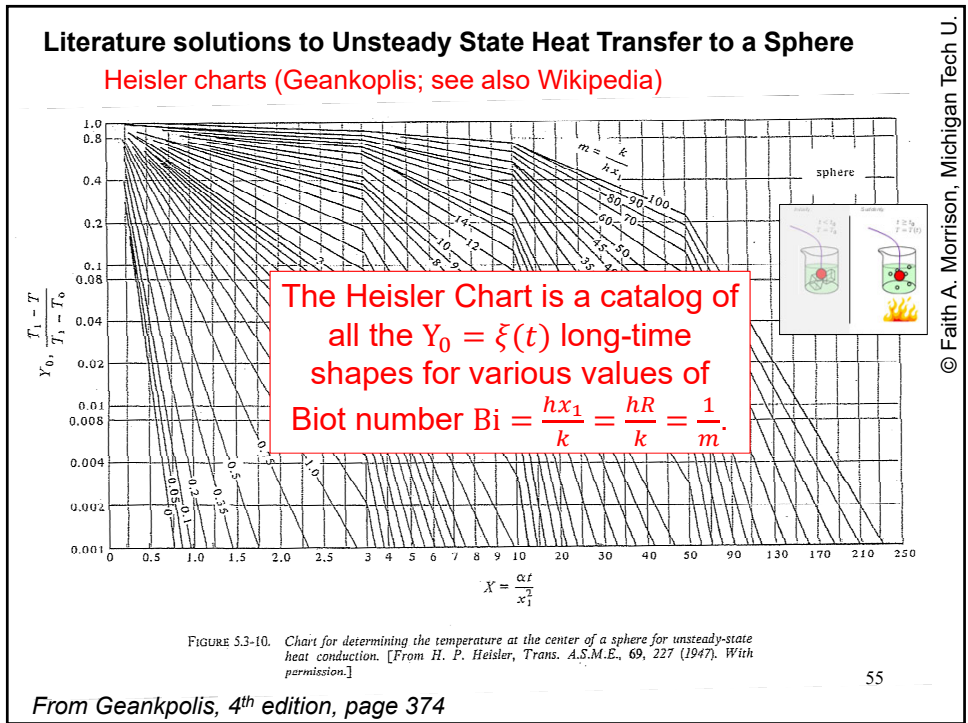


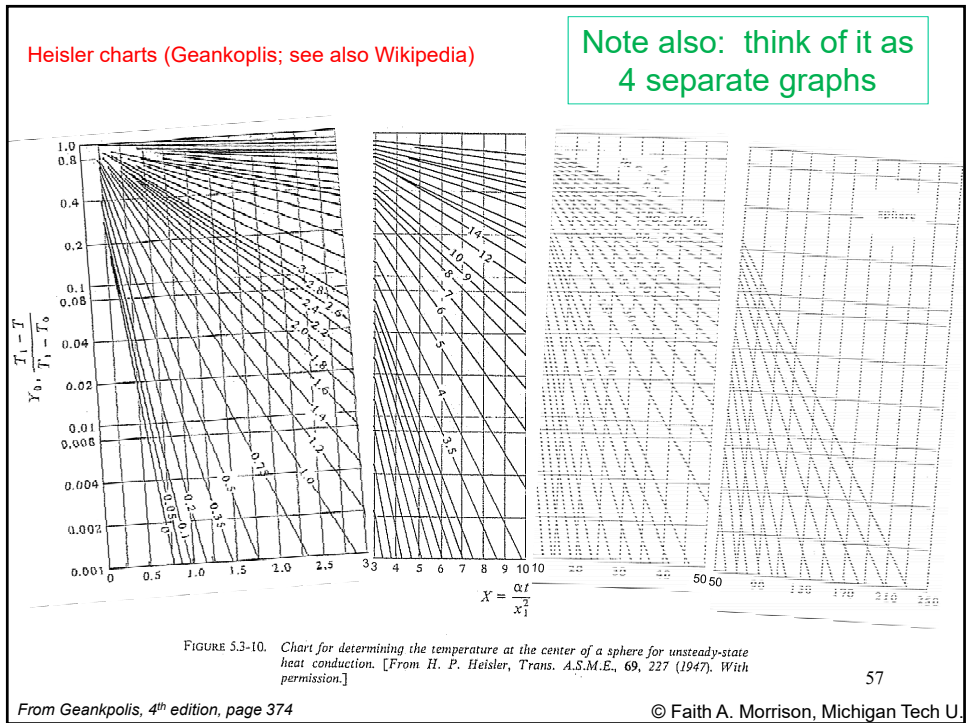
FIGURE 5.3-10. Chart for determining the temperature at the center of a sphere for unsteady-state heat conduction. [From H. P. Heisler, Trans. A.S.M.E., 69, 227 (1947). With permission.]

From Geankoplis, 4th edition, page 374

54

© Faith A. Morrison, Michigan Tech U.





Unsteady State Heat Transfer

CM3 120 Transport/Unit Operations 2

More complex Systems:
Unsteady State Heat Transfer
(Analytical Solutions)

Professor Faith A. Morrison
Department of Chemical Engineering
Michigan Technological University

Summary

- Unsteady state heat transfer is very common in the chemical process industries
- Temperature distributions depend strongly on what initiates the heat transfer (usually something at the boundary)
- **Internal resistance** (D/k) can be limiting, irrelevant, or one among many resistances
- **External resistance** ($1/h$) can be limiting irrelevant, or one among many resistances
- **Dimensional analysis**, once again, organizes the impacts of various influences (Bi, Fo)

58
© Faith A. Morrison, Michigan Tech U.

Unsteady State Heat Transfer

Summary (continued)

- In transport phenomena, we have dimensionless numbers that represent three important aspects of situations that interest us:

- The relative importance of individual terms in the equations of change
- The relative magnitudes of the diffusive transport coefficients ν, α, D_{AB}
- Scaled values of quantities of interest, e.g. wall forces, heat transfer coefficients, and mass transfer coefficients (data correlations)

CM3 120 Transport/Unit Operations 2

More complex Systems:
Unsteady State Heat Transfer
(Analytical Solutions)

Professor Faith A. Morrison
Department of Chemical Engineering
Michigan Technological University

Dimensionless Numbers

$Re - \text{Reynolds} = \frac{\rho V D}{\mu} = \frac{VD}{\nu}$ $Fr - \text{Froude} = \frac{V^2}{gD}$ $Pe - \text{Péclet}_{th} = RePr = \frac{\rho V D}{\mu} \frac{c_p D}{k} = \frac{VD}{\alpha}$ $Pe - \text{Péclet}_{tm} = ReSc = \frac{\rho V D}{\mu} \frac{D_{AB}}{D_{AB}} = \frac{VD}{D_{AB}}$ $Pr - \text{Prandtl} = \frac{c_p \mu}{k} = \frac{\nu}{\alpha}$ $Sc - \text{Schmidt} = LePr = \frac{\mu}{\rho D_{AB}} = \frac{\nu}{D_{AB}}$ $Le - \text{Lewis} = \frac{\alpha}{D_{AB}}$ $f - \text{Friction Factor} = \frac{D_{AB}}{D_{AB}} \frac{D_{AB}}{(D_{AB})^2} \frac{D_{AB}}{D_{AB}}$ $Nu - \text{Nusselt} = \frac{hD}{k}$ $Sh - \text{Sherwood} = \frac{k_m D}{D_{AB}}$	<div style="border: 1px solid #ccc; padding: 5px; margin-bottom: 5px; background-color: #ffe0e0;"> <p>These numbers from the governing equations tell us about the relative importance of the terms they precede in the microscopic balances (scenario properties).</p> </div> <div style="border: 1px solid #ccc; padding: 5px; margin-bottom: 5px; background-color: #e0ffe0;"> <p>These numbers compare the magnitudes of the diffusive transport coefficients ν, α, D_{AB} (material properties).</p> </div> <div style="border: 1px solid #ccc; padding: 5px; background-color: #e0e0ff;"> <p>These numbers are defined to help us build transport data correlations based on the fewest number of grouped (dimensionless) variables (scenario properties).</p> </div>
--	--

59

© Faith A. Morrison, Michigan Tech U.

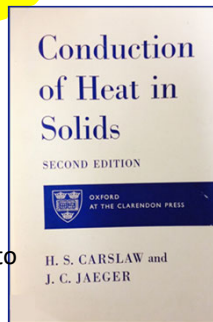
Unsteady State Heat Transfer

Summary (continued)

- If we can develop a model situation for questions of interest, the solutions of the models are often in the literature

Our responsibility in 21st century:

- Learn to develop models that will allow us to estimate or determine answers to the questions that interest us
- Learn to use published solutions (tables, charts) to answer questions that interest us



CM3 120 Transport/Unit Operations 2

More complex Systems:
Unsteady State Heat Transfer
(Analytical Solutions)

Professor Faith A. Morrison
Department of Chemical Engineering
Michigan Technological University

Dimensionless Numbers

$Re - \text{Reynolds} = \frac{\rho V D}{\mu} = \frac{VD}{\nu}$ $Fr - \text{Froude} = \frac{V^2}{gD}$ $Pe - \text{Péclet}_{th} = RePr = \frac{\rho V D}{\mu} \frac{c_p D}{k} = \frac{VD}{\alpha}$ $Pe - \text{Péclet}_{tm} = ReSc = \frac{\rho V D}{\mu} \frac{D_{AB}}{D_{AB}} = \frac{VD}{D_{AB}}$ $Pr - \text{Prandtl} = \frac{c_p \mu}{k} = \frac{\nu}{\alpha}$ $Sc - \text{Schmidt} = LePr = \frac{\mu}{\rho D_{AB}} = \frac{\nu}{D_{AB}}$ $Le - \text{Lewis} = \frac{\alpha}{D_{AB}}$ $f - \text{Friction Factor} = \frac{D_{AB}}{D_{AB}} \frac{D_{AB}}{(D_{AB})^2} \frac{D_{AB}}{D_{AB}}$ $Nu - \text{Nusselt} = \frac{hD}{k}$ $Sh - \text{Sherwood} = \frac{k_m D}{D_{AB}}$	<div style="border: 1px solid #ccc; padding: 5px; margin-bottom: 5px; background-color: #ffe0e0;"> <p>These numbers from the governing equations tell us about the relative importance of the terms they precede in the microscopic balances (scenario properties).</p> </div> <div style="border: 1px solid #ccc; padding: 5px; margin-bottom: 5px; background-color: #e0ffe0;"> <p>These numbers compare the magnitudes of the diffusive transport coefficients ν, α, D_{AB} (material properties).</p> </div> <div style="border: 1px solid #ccc; padding: 5px; background-color: #e0e0ff;"> <p>These numbers are defined to help us build transport data correlations based on the fewest number of grouped (dimensionless) variables (scenario properties).</p> </div>
--	--

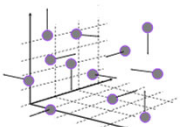
60


© Faith A. Morrison, Michigan Tech U.

NEXT: Diffusion and Mass Transfer

CM3120 Transport/Unit Operations 2

Diffusion and Mass Transfer



 **Professor Faith A. Morrison**
Department of Chemical Engineering
Michigan Technological University

www.chem.mtu.edu/~fmorriso/cm3120/cm3120.html

