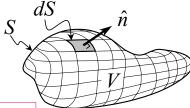
## **General Energy Transport Equation**

(microscopic energy balance)

As for the derivation of the microscopic momentum balance, the microscopic energy balance is derived on an arbitrary volume, V, enclosed by a surface, S.



Gibbs notation:

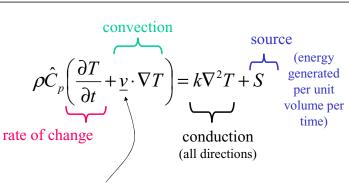
$$\rho \hat{C}_p \left( \frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T \right) = k \nabla^2 T + S$$

see handout for component notation

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## **General Energy Transport Equation**

(microscopic energy balance)



velocity must satisfy equation of motion, equation of continuity

see handout for component notation

Equation of energy for Newtonian fluids of constant density,  $\rho$ , and thermal conductivity, k, with source term (source could be viscous dissipation, electrical energy, chemical energy, etc., with units of energy/(volume time)).

#### CM310 Fall 1999 Faith Morrison

Source: R. B. Bird, W. E. Stewart, and E. N. Lightfoot, *Transport Processes*, Wiley, NY, 1960, page 319.

Gibbs notation (vector notation)

$$\left(\frac{\partial T}{\partial t} + \underline{v} \cdot \nabla T\right) = \frac{k}{\rho \hat{C}_p} \nabla^2 T + \frac{S}{\rho \hat{C}_p}$$

Note: this handout is on the web

Cartesian (xyz) coordinates:

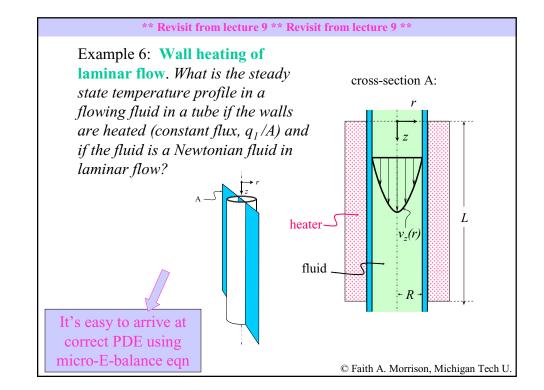
$$\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} = \frac{k}{\rho \hat{C}_p} \left( \frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + \frac{S}{\rho \hat{C}_p}$$

Cylindrical ( $r\theta z$ ) coordinates:

$$\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} = \frac{k}{\rho \hat{C}_p} \left( \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial T}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right) + \frac{S}{\rho \hat{C}_p}$$

Spherical (rθφ) coordinates:

$$\frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + \frac{v_\phi}{r \sin \theta} \frac{\partial T}{\partial \phi} = \frac{k}{\rho \hat{C}_p} \left( \frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial T}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \theta} 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\left( \sin \theta \frac{\partial T}{\partial \theta} \right) +$$





# Microscopic Energy Balance – is the correct physics for all problems!

## Tricky step:

solving for *T* field; this can be mathematically difficult

- •partial differential equation in up to three variables
- •boundaries may be complex
- •multiple materials, multiple phases present
- •may not be separable from mass and momentum balances

Strategy: solve using numerical methods

(e.g. Comsol)

\*\*\*\* Or \*\*\*\*

Develop correlations on complex systems by using

**Dimensional Analysis** 

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## There are two types of data correlations you will encounter:

- 1. Standardized correlations you can look up (e.g. for *h* to use in heat exchangers)
- 2. In-house correlations you develop yourself (e.g. for scale-up)

Type 1: Standardized Correlations for Simple Geometries (you can look them up), e.g. heat exchanger design

out

 $Q = UA\Delta T$ these are easy to sort

but what about this? For design purposes, where

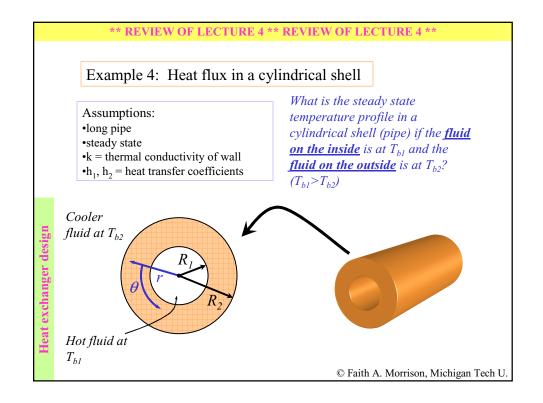
do we get U?

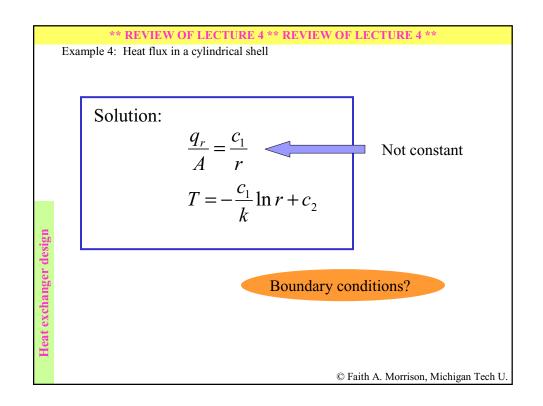
**ANSWER:** Begin at the beginning: how did we <u>define</u> (develop) *U*?

 $T_1'$  less hot  $T_2'$ hot

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eat exchanger design





#### \*\* REVIEW OF LECTURE 4 \*\* REVIEW OF LECTURE 4 \*\*

Example 4: Heat flux in a cylindrical shell

$$\frac{c_{1}}{R_{1}} = h_{1}(T_{b1} - T_{w1})$$

$$\frac{c_{1}}{R_{2}} = h_{2}(T_{w2} - T_{b2})$$

$$4 equations 4 unknown$$

$$T_{w1} = -\frac{c_{1}}{k} \ln R_{1} + c_{2}$$

$$T_{w2} = -\frac{c_{1}}{k} \ln R_{2} + c_{2}$$

4 equations

4 unknowns;  $c_1, T_{w1}, c_2, T_{w2}$ 

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#### \*\* REVIEW OF LECTURE 4 \*\* REVIEW OF LECTURE 4 \*\*

Example 4: Heat flux in a cylindrical shell

Results: Radial Heat flux in an Annulus

$$T - T_{b2} = \frac{\left(T_{b1} - T_{b2}\right) \left(\ln\left(\frac{R_2}{r}\right) + \frac{k}{h_2 R_2}\right)}{\frac{k}{h_2 R_2} + \ln\left(\frac{R_2}{R_1}\right) + \frac{k}{h_1 R_1}}$$

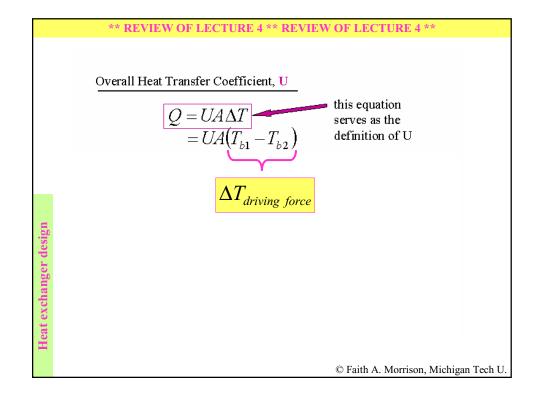
$$\frac{q_r}{A} = \left(\frac{\left(T_{b1} - T_{b2}\right)}{\frac{1}{h_2 R_2} + \frac{1}{k} \ln\left(\frac{R_2}{R_1}\right) + \frac{1}{h_1 R_1}}\right) \left(\frac{1}{r}\right)$$

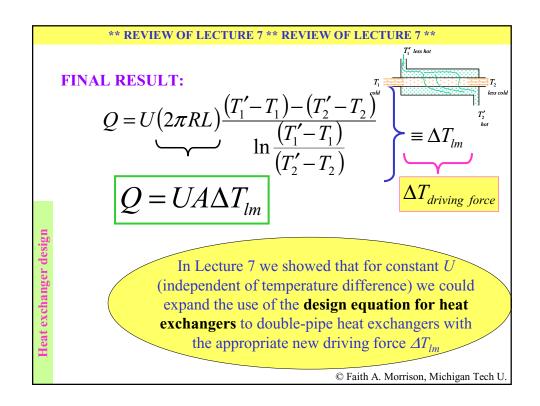
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Example 4: Heat flux in a cylindrical shell

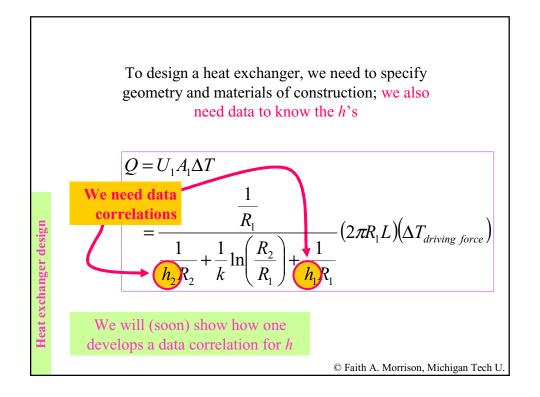
Flux:
$$\frac{q_r}{A} = \frac{\left(T_{b1} - T_{b2}\right)}{\frac{1}{h_2 R_2} + \frac{1}{k} \ln \left(\frac{R_2}{R_1}\right) + \frac{1}{h_1 R_1}} \left(\frac{1}{r}\right)$$
Total Heat flow:
$$Q = \left(\frac{q_r}{A}\right) 2\pi L \qquad \text{note that total heat flow is proportional to bulk AT and (almost) area of heat transfer}$$
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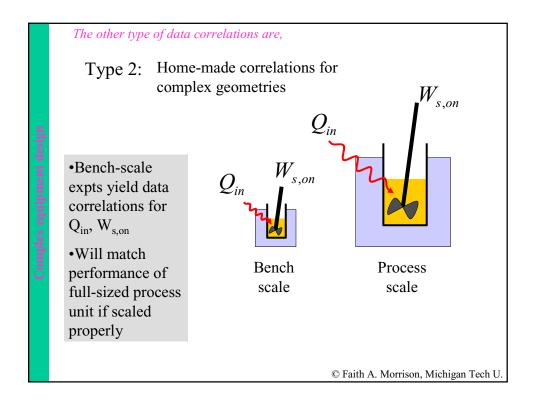


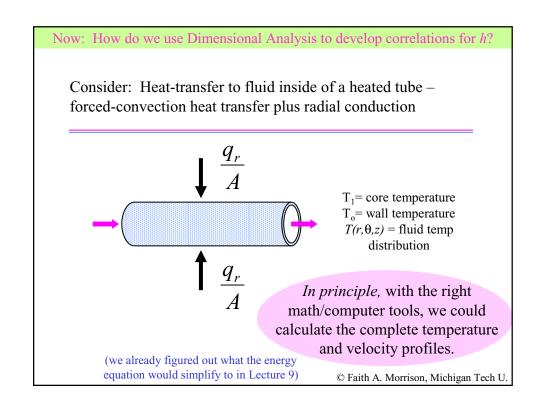


overall heat xfer coeffs in pipe 
$$Q = U_1 A_1 \Delta T$$

$$= \left(\frac{\frac{1}{R_1}}{\frac{1}{h_2 R_2} + \frac{1}{k} \ln \left(\frac{R_2}{R_1}\right) + \frac{1}{h_1 R_1}}\right) (2\pi R_1 L) \Delta T_{driving \ force}$$
The design equation for heat exchangers indicates that,
$$U \text{ depends on geometry } (R_1, R_2), \text{ materials of construction } (k), \text{ and } \dots h_1, h_2.$$
Individual "side" heat transfer coefficients
$$\text{© Faith A. Morrison, Michigan Tech U.}$$







Now: How do we use Dimensional Analysis to develop correlations for *h*?

*In principle*, with the right math/computer tools, we could calculate the complete temperature and velocity profiles.

**How:** From the microscopic mass, momentum, and energy balances, calculate velocity field, temperature field, and calculate the total amount of heat transferred. Finally, determine h from its definition.

*From the complete solution*, we could then calculate *h*:

$$h = \frac{q_r}{A(T_{bulk} - T_{wall})}$$

Integrate heat flux across the total surface area; heat flux comes from Fourier's law; Fourier's law needs T field

Therefore, the equations we need for the complete solution, and the equation we use to define h contain all the physics of h; dimensional analysis on these equations will tell us what h is a function of

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Now: How do we use Dimensional Analysis to develop correlations for *h*?

Therefore, the equations we need for the complete solution, and the equation we use to define h contain all the physics of h; dimensional analysis on these equations will tell us what h is a function of

Once we know what *h* is a function of, we can conduct experiments, measure *h* as a function of its variables, and report these data correlations in the literature for others to use.

## **Dimensional Analysis**

governing equations

continuity equation (mass conservation)
equation of motion (momentum conservation)
equation of energy (energy conservation)

strategy: render the governing equations (and boundary conditions) dimensionless to identify the important parameters that apply in every situation.

rely on experiments and data correlations

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#### Dimensional Analysis Procedure:

- 1. select appropriate differential equations and boundary conditions
- 2. select characteristic quantities with which to scale the variables, e.g.  $\underline{v}$ , x, P, T- $T_o$ 
  - characteristic quantities must be constant
  - must be representative of the system
- 3. scale all variables in the governing equations; yields dimensionless equation as a function of *dimensionless groups*The values of the dimensionless groups determine the properties of the differential equations.
- 4. design scaled-down experiments to develop *data correlations* for the system of interest
- 5. use data correlations to design and evaluate systems

#### OR

4. perform experiments on an existing system and *correlate* results using dimensionless groups

## Energy equation in Cartesian Coordinates:

$$\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} = \frac{k}{\rho \hat{C}_p} \left( \frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + \frac{S}{\rho \hat{C}_p}$$

Next step: choose scale factors and scale each variable.

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### non-dimensional variables:

$$t^* \equiv \frac{tV}{D}$$

temperature:
$$velocity.$$

$$v^* \equiv \frac{tV}{D}$$

$$v^* \equiv \frac{y}{D}$$

$$v^* \equiv \frac{v_x}{V}$$

$$v^*_x \equiv \frac{v_x}{V}$$

$$v^*_y \equiv \frac{v_y}{V}$$

$$v^*_z \equiv \frac{v_z}{V}$$

$$v^*_z \equiv \frac{v_z}{V}$$

$$v_{x}^{*} \equiv \frac{v_{x}}{V}$$

$$* \quad v_{y}$$

$$v_y^* \equiv \frac{v_y}{V}$$

$$v_z^* \equiv \frac{v_z}{V}$$

driving force:

$$P^* \equiv \frac{P}{\rho V^2}$$
$$g_z^* \equiv \frac{g_z}{\sigma}$$

temperature:

$$T^* \equiv \frac{T - T_o}{(T_1 - T_o)}$$

## Energy equation in Cartesian Coordinates:

$$\frac{\partial T}{\partial t} + v_x \frac{\partial T}{\partial x} + v_y \frac{\partial T}{\partial y} + v_z \frac{\partial T}{\partial z} = \frac{k}{\rho \hat{C}_p} \left( \frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) + \frac{S}{\rho \hat{C}_p}$$

we substitute the non-dimensional variables. e.g.

$$\frac{\partial T}{\partial t} = \frac{\partial \left[ (T_1 - T_o)T^* + T_o \right]}{\partial \left[ \frac{t^* D}{V} \right]} = \left[ \frac{(T_1 - T_o)V}{D} \right] \frac{\partial T^*}{\partial t^*}$$

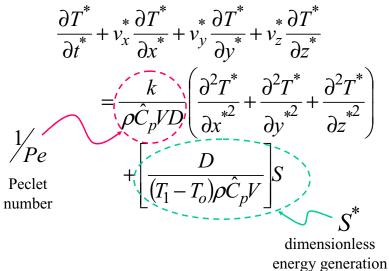
after carrying out this change of variable for each term we get the non-dimensional energy equation



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## Non-dimensional Energy Equation



Reynolds number Dimensionless Groups:

Re

$$Pe = \frac{\rho \hat{C}_p VD}{k} = \left(\frac{\hat{C}_p \mu}{k}\right) \left(\frac{\rho VD}{\mu}\right)$$
Peclet number

Prandtl number

$$Pe = Pr Re$$

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We can conclude from the energy equation that

Pe = Pr Re

dimensionless temperature T\* is a function of:
$$T^* = T^*(t^*, x^*, y^*, z^*, v_x^*, v_y^*, v_z^*, \text{Pe}, S^*)$$

We know that dimensionless velocities are a function of:

$$v_i^* = v_i^*(t^*, x^*, y^*, z^*, \text{Re, Fr})$$

$$\Rightarrow T^* = T^*(t^*, x^*, y^*, z^*, \text{Re}, \text{Pr}, \text{Fr}, S^*)$$

the equations governing temperature distributions depend on only three dimensionless groups

Now: How do we use Dimensional Analysis to develop correlations for h?

Therefore, the equations we need for the complete solution, and the equation we use to define h contain all the physics of h; dimensional analysis on these equations will tell us what h is a function of

This part is now done. From the microscopic mass, momentum, and energy balances we know that h will depend on at most Re, Pr, Fr.

Now we need to check the equation that defines *h* for more dimensionless groups