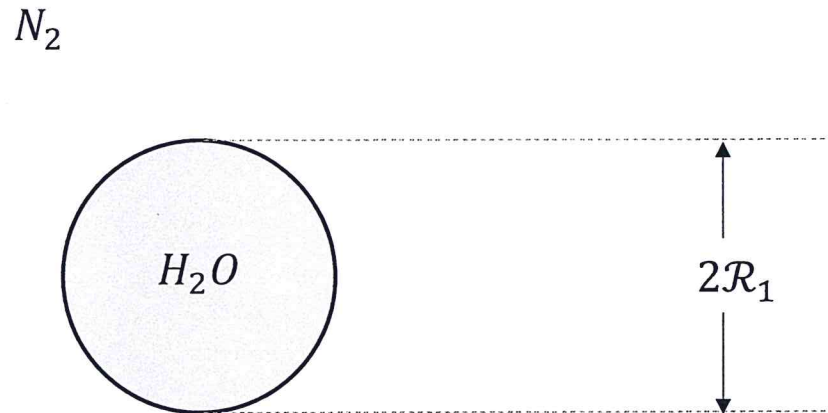


4-MAR-2019
FAM

①

QUICK START

Example: A water mist forms in an industrial printing operation. Spherical water droplets slowly and steadily evaporate into the air (mostly nitrogen). What is the water mole fraction as a function distance from the droplet?



Let's Interrogate
the problem.

REPORT:

2

The Equation of Species Mass Balance in Terms of Combined

Molar quantities in Cartesian, cylindrical, and spherical coordinates for binary mixtures of A and B.

The general case, where the combined molar flux with respect to molar velocity (\underline{N}_A), is given on page 1.

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Microscopic species mass balance, in terms of molar flux; Gibbs notation

$$\frac{\partial c_A}{\partial t} = -\nabla \cdot \underline{N}_A + R_A$$

Microscopic species mass balance, in terms of combined molar flux; Cartesian coordinates

$$\frac{\partial c_A}{\partial t} = -\left(\frac{\partial N_{Ax}}{\partial x} + \frac{\partial N_{Ay}}{\partial y} + \frac{\partial N_{Az}}{\partial z}\right) + R_A$$

Microscopic species mass balance, in terms of combined molar flux; cylindrical coordinates

$$\frac{\partial c_A}{\partial t} = -\left(\frac{1}{r} \frac{\partial(r N_{Ar})}{\partial r} + \frac{1}{r} \frac{\partial N_{A\theta}}{\partial \theta} + \frac{\partial N_{Az}}{\partial z}\right) + R_A$$

Microscopic species mass balance, in terms of combined molar flux; spherical coordinates

$$\frac{\partial c_A}{\partial t} = -\left(\frac{1}{r^2} \frac{\partial(r^2 N_{Ar})}{\partial r} + \frac{1}{r \sin \theta} \frac{\partial(N_{A\theta} \sin \theta)}{\partial \theta} + \frac{1}{r \sin \theta} \frac{\partial N_{Az}}{\partial \phi}\right) + R_A$$

no θ - ϕ symmetry

Fick's law of diffusion, Gibbs notation: $\underline{N}_A = x_A(N_A + \underline{N}_B) - cD_{AB} \nabla x_A$

$$= c_A \underline{v}^* - cD_{AB} \nabla x_A$$

shells stack θ, ϕ symmetry

Fick's law of diffusion, Cartesian coordinates:

$$\begin{pmatrix} N_{Ax} \\ N_{Ay} \\ N_{Az} \end{pmatrix}_{xyz} = \begin{pmatrix} x_A(N_{Ax} + N_{Bx}) - cD_{AB} \frac{\partial x_A}{\partial x} \\ x_A(N_{Ay} + N_{By}) - cD_{AB} \frac{\partial x_A}{\partial y} \\ x_A(N_{Az} + N_{Bz}) - cD_{AB} \frac{\partial x_A}{\partial z} \end{pmatrix}_{xyz}$$

Fick's law of diffusion, cylindrical coordinates:

$$\begin{pmatrix} N_{Ar} \\ N_{A\theta} \\ N_{Az} \end{pmatrix}_{r\theta z} = \begin{pmatrix} x_A(N_{Ar} + N_{Br}) - cD_{AB} \frac{\partial x_A}{\partial r} \\ x_A(N_{A\theta} + N_{B\theta}) - \frac{cD_{AB}}{r} \frac{\partial x_A}{\partial \theta} \\ x_A(N_{Az} + N_{Bz}) - cD_{AB} \frac{\partial x_A}{\partial z} \end{pmatrix}_{r\theta z}$$

Fick's law of diffusion, spherical coordinates:

$$\begin{pmatrix} N_{Ar} \\ N_{A\theta} \\ N_{A\phi} \end{pmatrix}_{r\theta\phi} = \begin{pmatrix} x_A(N_{Ar} + N_{Br}) - cD_{AB} \frac{\partial x_A}{\partial r} \\ x_A(N_{A\theta} + N_{B\theta}) - \frac{cD_{AB}}{r} \frac{\partial x_A}{\partial \theta} \\ x_A(N_{A\phi} + N_{B\phi}) - \frac{cD_{AB}}{r \sin \theta} \frac{\partial x_A}{\partial \phi} \end{pmatrix}_{r\theta\phi}$$

B is constant (no sink)

(3)

microscopic species A mass balance

$$0 = -\cancel{\frac{1}{r^2}} \frac{d}{dr} (r^2 N_{A,r})$$

$\equiv \text{div}$

$$\frac{d\Phi}{dr} = 0$$

$$\Phi = C_1$$

$$r^2 N_{A,r} = C_1$$

$$N_{A,r} = \frac{C_1}{r^2}$$

From
before

$$N_{Ar} = \frac{C_1}{r^2}$$

(9)

Fick's Law of Diffusion:

$$N_{Ar} = x_A N_{Ar} - c D_{AB} \frac{dx_A}{dr}$$

$$\frac{C_1}{r^2} \rightarrow$$

$$N_{Ar} (1 - x_A) = -c D_{AB} \frac{dx_A}{dr}$$

$$\frac{C_1}{r^2} dr = -c D_{AB} \frac{dx_A}{(1 - x_A)}$$

$$\int \left(\frac{-C_1}{c D_{AB}} \right) r^{-2} dr = \int \frac{-dx_A}{(1 - x_A)} = u$$

$$du = -dx_A$$

$$\left(\frac{+C_1}{CD_{AB}} \right) \frac{r^{-1}}{+1} = -\ln(1-x_A) + C_2 \quad \text{②}$$

BC: $r = R_1$ $x_A = x_A^*$ (saturated)

two possible choices

$$\left\{ \begin{array}{l} r = \infty \quad x_A = x_{A\infty} \leftarrow \text{isolated drop} \\ \text{or} \\ r = R_2 \quad x_A = x_{A2} \leftarrow \text{FILM model} \end{array} \right.$$

Solve (we use BC #2)

⋮

- ① substitute
- ② 2 eqns, 2 unknowns
- ③ algebra

$$\Rightarrow C_1 = C_{DAB} \frac{\ln\left(\frac{1-x_{A1}}{1-x_{A2}}\right)}{\frac{1}{Q_2} - \frac{1}{Q_1}}$$

$$C_2 = \begin{matrix} \rightarrow \\ \rightarrow \end{matrix}$$

$$G_2 = \ln(1 - X_{A1}) + \left(\frac{\frac{1}{Q_1}}{\frac{1}{Q_2} - \frac{1}{Q_1}} \right) \ln \left(\frac{1 - X_{A1}}{1 - X_{A2}} \right) \quad \text{⑦}$$

substitute back ...

Final result:

$$\left(\frac{1 - X_{A2}}{1 - X_{A1}} \right)^{\left(\frac{V_{R1} - V_r}{V_{R1} - V_{R2}} \right)} = \frac{(1 - X_A)}{(1 - X_{A1})}$$

dependent variable

indep variable

(compare WRF
P502)

$X_A(r) //$

FINAL MODEL



QUICK START

Example: A water mist forms in an industrial printing operation. Spherical water droplets slowly and steadily evaporate into the air (mostly nitrogen). The evaporation creates a film around the droplets through which the evaporating water diffuses. What is the water mole fraction in the film as a function of radial position? You may assume ideal gas properties for air.

